

Flocculation Treatment BMPs for Construction Water Discharges

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August 2014

Research Project Final Report 2014-25

















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Flocculation Treatment BMPs for Construction Water Discharges

Final Report

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#### **Executive Summary**

This report presents the evaluation of chemical treatment practices known as flocculation for purifying construction site runoff of sediment, suspended solids and colloidal particles. Flocculation is generally used in combination with traditional sedimentation and filtration methods related to stormwater pollution prevention and dewatering fluid treatment.

In particular, this report presents the best management practices (BMPs) for evaluation and design of flocculant treatment methods and technologies and the associated sizing for flows representative of field operations. Soil samples from across Minnesota, representing the wide range of geologic and geomorphological conditions, were used to identify differences in flocculant applicability and effectiveness.

#### **Chapter 1: Introduction**

Construction site runoff carries sediment, suspended solids and colloidal particles. In Minnesota locations with fine-grained soils, traditional stormwater best management practices (silt fences, diversion ditches, temporary seeding, check dams and sediment ponds; Minnesota Stormwater Manual, 2005) are ineffective in removing many fine particles, frustrating construction personnel and leaving construction sites unable to meet stormwater regulations. Flocculation, a chemical treatment to enhance fine particle sedimentation and filtration, is a common technique in the controlled world of water treatment. Work done by the Federal Highway Administration and several states has shown great promise in the application of flocculation to construction site stormwater management, removing fine particles with predictable and flexible approaches robust enough to meet a wide range of runoff conditions.

The focus of this work is to develop treatment practices for construction site runoff that carries sediment, suspended solids and colloidal particles. In particular, this report presents the best management practices (BMPs) for evaluation and design of flocculant treatment methods and technologies and the associated sizing for flows representative of field operations.

This report represents the work done under Mn/DOT Agreement 00734, Flocculation Treatment Best Management Practices for Construction Water Discharges. This work was performed by Minnesota State University (MSU), Mankato in the Environmental Engineering laboratory, as part of the Center for Transportation Research and Implementation.

#### 1.1: Background and Theory of Flocculant Treatment

Flocculation is the chemical treatment of fine sediments in water such that particles aggregate and become larger and heavier in groups than as individual particles. Increasing the combined mass enhances the sedimentation of the particle group by increasing the downward settlement velocity.

Chemical addition to cause flocculation is well known in the controlled situation of water treatment (Crittenden, 2005). Performance is a function of both chemical characteristics (flocculent, sediment concentration, electrostatic charge, dose, pH, alkalinity and temperature) and hydraulic factors (flow, blending, mixing, sedimentation, filtration). Because so many factors apply, performance is typically evaluated using a bench scale "jar test" in which flocculent dose and other conditions are varied to assess the optimum combination. Treatments may include up to three chemical steps involving pH adjustment, coagulation for particle charge reduction and flocculation for particle aggregation. Aggregated particles may be removed from water flow by sedimentation (settlement) or by filtration. Reductions in turbidity from 300 NTUs or greater down to less than 5 NTUs are commonly achieved with proper control.

Crittenden (2005) lists flocculants in several classes including metal salts (ferric chloride, alum, ferric sulfate, polyaluminum chloride), clays (bentonite), chitosan, and polyacrylamide (PAM). PAM is supplied in three ionic states (anionic, cationic and non-ionic) across nearly one hundred different chemical compositions (NSF/ANSI Standard 60). Besides for water treatment, these products are also used in food manufacture and agriculture soil conditioning.

Davis (2010) summarizes flocculation treatment as consisting of three main steps:

- Injection and rapid (flash) mix of coagulant compounds, such that particle surface charge is changed to increase inter-particle attraction;
- Slow mix so that particles bump and aggregate into flocs but flocs do not shear; and,
- Sedimentation, in which particles leave the water column by gravity settlement.

A fourth step may also be implemented, consisting of filtration to "polish" the water and remove particles that did not settle out in the time allotted. Crittenden (2005) describes filtration for flocculant treated water as having three possible classifications, depending upon the actions taken in the filtration treatment sequence:

- Conventional filtration, consisting of rapid mix, slow mix and sedimentation prior to filtration:
- Direct filtration, consisting of rapid mix and slow mix but no sedimentation prior to filtration:
- In line filtration, consisting of rapid mix only prior to filtration.

Conventional filtration is considered appropriate for waters with turbidities up to 1000 NTUs, while direct and in line filtration approaches are considered appropriate for waters with turbidities less than 15 NTUs due to the effort involved in backwashing or cleaning filters.

Davis (2010) makes several points about key characteristics of successful flocculation treatment:

- Rapid mixture of coagulants needs very high mixing velocity gradients to cause particles to contact each other, rather than just ride along and not aggregate. High mixing power may be required. However, ferric chloride in particular can mix into water very quickly and may require a mixing time of less than 10 seconds total, hence the term "flash" mixing. Rapid mixing may be the most important factor for coagulation (chemical) efficiency.
- Slow mixing to encourage floc aggregation from particles in treated waters must be gentle, with low to very low power such that particles come into contact but flocs do not break apart or shear. Slow mixing may be the most important factor for particle size and removal. Slow mixing times are typically 20 to 30 minutes, with shorter times in this range appropriate for summertime water temperatures and longer times appropriate for wintertime temperatures. Water velocities during slow mixing should be 0.5 to 1.0 ft/s to reduce the potential for inadvertent sedimentation that could clog the slow mixing process.
- Sedimentation may be by strict gravity force action with individual particles (called Type I settlement) or be enhanced by enlargement of particles with the associated increase in particle mass and the subsequent increase in particle velocity (called Type II settlement).

Type II settlement can reduce the sedimentation time required. Typical sedimentation times are 60 minutes, after which wind currents can create sufficient water velocities to limit further particle settlement.

Note that Pizzi (2010) recommends that flocs be "pinhead size" for optimum settling. Flocs as large as quarters (3/4 in) may be too buoyant to settle, in spite of their impressive size. Pizzi (2010) also states that floc formation typically takes an average of 30 minutes, although 10 minute formation times are possible under some conditions.

During the preparation of BMPs for flocculant treatment methods and technologies for construction runoff and stormwater treatment field situations, it was assumed that structural tankage would be limited to 10,000 gallons per individual tank, the size of a transportable fractionation tank (tractor trailer size). Particular emphasis has been placed on practices for design of methods that could be implemented in ponds or channel waterways, with the water mixing and control appropriate to construction site conditions rather than "swimming pool" like conditions of drinking water treatment facilities associated with multi decadal-scale treatment durations.

#### 1.2: Application of Flocculant Treatment to Construction Waters

In recent decades, flocculation has been used in mining, construction water treatment and eutrophic lake treatment. Early leaders in flocculation included the States of Washington, Oregon and California to protect the Pacific salmon fishery (Jurries, undated; Bachand, et al., 2010). Minnesota DNR suggests using alum as a flocculent for algae-choked lakes, although the dosing is limited by concerns for aquatic toxicity. The Minnesota Stormwater Manual (2005) notes chemical treatment by flocculation, but gives few details. Discussion with vendors suggests that flocculants are seldom used in Minnesota; this omission may be a lost opportunity.

Hesitation to employ flocculation on construction sites may be caused by: the chemical complexity of flocculation (Crittenden, 2005); the concerns over aquatic toxicity from residual flocculent after treatment (McLaughlin and Zimmerman, 2009); or bad experiences associated with insufficient blending ("the creation of mud balls", Crittenden, 2005; "gelatinous masses or 'fish eyes' McLaughlin and Zimmerman, 2009). Chemical complexity is addressed by pH adjustment and jar testing, residual effects are mitigated by use of anionic compounds that adhere to natural organic sediments rather than aquatic animal tissues, while effective blending may be achieved by:

- In-pipe methods (Iwinski, 2006; McLaughlin and Zimmerman, 2009);
- Passive log or pillow methods (Iwinski, 2006);
- Static mixers (Crittenden, 2005; McLaughlin and Zimmerman, 2009); or,
- Hydraulic (baffle) mixers (Crittenden, 2005).

Previous work at Minnesota State, Mankato for the treatment of construction waters with concrete sediments (Mn/DOT contract 96273 - Concrete Slurry, Wash and Loss Water Mitigation) has shown tremendous benefit from the use of flocculants (Figure 1).

Treatment cost for construction water flocculation has been reported as between \$0.01 to \$0.03/gallon for continuous reactor (in line) treatment (McLaughlin and Zimmerman, 2009) and \$0.08/gallon for batch reactor (off line) treatment (Jurries, undated). Anecdotal reports suggest that contractors comfortable with flocculation performance may reduce construction site BMPs to eliminate operational conflicts such as haul roads crossing silt fences lines or diversion ditches and instead trust the flocculation to catch sediments.



Figure 1: Chitosan flocculent treatment (on left) of Minnesota River silt at ~200g/L. Photo taken 75 seconds after addition of flocculent.

#### **Chapter 2: Sediment Samples**

Sediment samples were created from soils collected from across Minnesota, in an attempt to represent the broad span of soil types associated with the geography of the state. This chapter describes the geographic targeting strategy, and provides details of the sample collection and characterization processes and associated results.

#### 2.1: Sample Collection

Soil samples were generally collected by MnDOT from construction projects active during 2012. Additional samples were collected by MSU researchers to represent soil types of interest. Samples were targeted to represent approximately three locations per MnDOT district, and to represent a wide range of geological, geomorphological and ecological conditions. Lastly, sample collectors were encouraged to send samples of soil types that had difficult to manage through traditional stormwater management techniques.

Samples were collected from 30 locations. In many locations, multiple samples were collected to represent different soil layers, resulting in 57 total soil samples. Figure 2 presents a map of Minnesota with sample locations named and indicated.

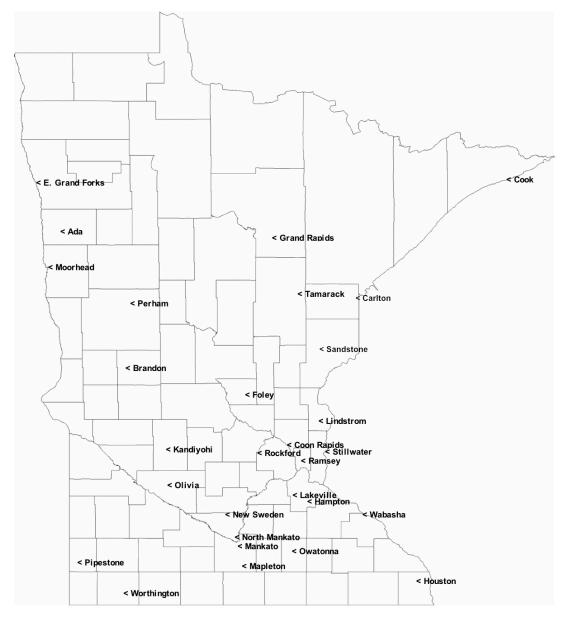
Appendix A contains a list of samples, with locations and dates of collection.

Soil samples were either brought to MSU directly or picked up by MSU staff from the MnDOT District 7 Soils Laboratory. Samples were logged and tagged then stored in a secure location in the MSU Environmental Engineering Laboratory.

#### 2.2: Characterization

Samples were evaluated for fine grain size (sizes less than 0.03 mm) using hydrometer analysis (ASTM D422), organic content (ASTM D2974), and pH (ASTM D4972). Results are provided in Appendix A, summarized on pages A-1 to A-4 and tabulated for individual soil samples on pages A-5 to A-61. Photomicrographs of the soil specimens are also provided with the individual soil sample information.

The geological and geomorphological characteristics of each sample was obtained from the mapped locations in comparison with the Quaternary geology described by Hobbs and Goebel (1982). A simplified version by Lusardi is included in Appendix B for reference; actual locations were checked against the full sized map. The ecological province for each sample was obtained from the web-based map Ecological Provinces of Minnesota (1999), also included in Appendix B. The natural vegetation for each sample was obtained from Coffin (1988); the key delineation map is also included in Appendix B.



Get Printable Maps From: Waterproof Paper.com

Figure 2. Sample locations and nomenclature.

Definitions from Hobbs and Goebel (1982) that specifically apply to the soil samples and classification of this task include:

<u>Drift:</u> All the rock materials transported by a glacier; includes till, outwash, ice-contact stratified drift, glacial lake sediment and loess.

Lobe: A major tongue of a continental glacier. Also the body of drift deposited by it.

<u>Moraine association:</u> Related bodies of till deposited during a more or less distinct phase of advance and retreat of an ice lobe. Each association contains a ground moraine and either a stagnation or an end moraine.

<u>Ground moraine:</u> A body of till deposited mainly from the bottom of a glacier as a more or less uniform blanket. Generally characterized by an undulating surface of hummocks or drumlins separated y swales. Includes some areas of low-relief ice-stagnation features.

<u>End moraine</u>: A body of drift deposited at the margin of a glacier when an approximate equilibrium in rates of ice flow and melting stabilized the position of the ice margin. Generally composed of till and ice-contact sand and gravel, but in places includes blocks of local material that was frozen to the sole of the glacier and thrust up into the moraine. Landforms range from belts of hills to a knob-and-kettle topography produced by collapse as the ice melted out.

Stagnation Moraine: A body of drift released by the melting of a glacier that has ceased flowing. Commonly, but not always, occurs near ice margins; composed of till, ice-contact, stratified drift, and small areas of glacial lake sediment. Typical landforms are knob-and-kettle topography, locally including ice-walled lake plains. Stagnation moraine is transitional to end moraine, and the distinction between them is rather arbitrary. The Vermilion and the Bemis Moraines are good examples of end moraines; the Alexandria and the Altamont Moraines are good examples of stagnation moraines. Most of the other moraines in Minnesota are transitional.

<u>Till:</u> An unsorted unstratified mixture of all sizes of rock material deposited directly by glacial ice with little or no reworking by water.

<u>Outwash</u>: Stratified drift, chiefly sand and gravel, which has been transported by glacial meltwater. Commonly pitted and collapsed by the melting of underlying ice, especially near former ice margins. Collapsed outwash is recognized by the uncollapsed remnant of the former depositional surface, as opposed to ice-contact stratified drift (included with moraines on this map).

<u>Loess:</u> Windblown silt and fine sand. Source areas include meltwater channels, outwash plains, and exposed glacial lake beds.

Descriptions from Hobbs and Goebel (1982) that specifically apply to the soil samples and classification of this task include:

<u>Deposits associated with the Des Moines lobe (Pleistocene, Late Wisconsinan):</u> gray calcareous drift (buff to brown where oxidized) shale and limestone clasts generally common, derived from Manitoba and eastern North Dakota, combined silt and clay typically exceeds 50% of till.

<u>Deposits associated with the Rainy lobe (Pleistocene, Late Wisconsinan)</u>: brown to gray noncalcareous drift; clasts predominately igneous and metamorphic rocks of the Canadian Shield.

<u>Deposits associated with the Wadena lobe (Pleistocene, Early and Late Wisconsinan):</u> gray calcareous drift (buff where oxidized); limestone clasts common, but shale rare or absent.

<u>Deposits associated with the Superior lobe (Pleistocene, Late Wisconsinan):</u> reddish-brown non-calcareous drift; clasts predominately igneous and metamorphic rocks of the Canadian Shield, but also present are distinctive clasts from the Superior basin, including red sedimentary rocks, amygdaloidal basalt, red rhyolite and agate.

Soil classifications were made using the MnDOT Triangle Textural Classification System, described in the MnDOT Draft Geotechnical Manual (MnDOT, 2008), and the Unified Soil Classification System (USCS) (ASTM D2487-00). The USCS classifications are considered presumptive because not all tests necessary for definitive classification were conducted (specifically Atterberg limits); however, sufficient information existed at the completion of the analyses of this study to select the probable classification. When doubt remained, dual USCS symbols were assigned.

#### **Chapter 3: Dosage Study**

Samples were evaluated for flocculant effectiveness by dosage using the jar test method (ASTM D2035), in which six 1 L samples (1.00 L tap water plus 25 g selected soil sample, well mixed) are simultaneously dosed with separate flocculant treatments, mixed and analyzed for improvements in clarity (Figures 3 to 5). Water analysis was done using turbidity (Oakton T-100 Waterproof Turbidity Meter) and pH (Hach HQ40d portable meter with IntelliCAL™ PHC201 pH probe) measurements at the start and end of the evaluation. Flash (rapid) mixing was done for 1 minute after flocculant dosing to thoroughly entrain the flocculant and create inter-particle collisions that are the basis for floc formation. Slow mixing was done for 20 minutes, then the samples were settled for an additional 20 minutes prior to determination of (final) treated turbidity level.

#### 3.1: Flocculants

Flocculant chemicals were obtained directly from manufacturers (Hawkins, Roseville, MN; Tramfloc, Tempe, AZ), from a contractor (Standard Contracting, Hampton, MN) or from a construction product supply company (Brock White, St. Paul, MN), as listed in Table 1. No chemical analyses were done on the flocculants to determine characteristics or composition; vendor information was accepted as sufficient. Flocculant products were kept in secure storage away from light and at standard laboratory temperature  $(20 - 22^{\circ} \text{ C})$ . Flocculant stock solutions were made up by measuring out predetermined amounts of flocculant and mixing with 250.0 mL of tap water (City of Mankato municipal water system), then stored in amber 250 mL Boston bottles with septa caps.

#### 3.2: Dosage Study

In Phase I of the dose evaluation, all 57 soils were tested with all 21 flocculant chemicals at the dose rates provided in Table 1. Dose rates were developed from manufacturer's recommendations, if available, and held constant for each flocculant chemical throughout the Phase I effort. Each soil-flocculant combination was evaluated in duplicate as a check on repeatability (2,394 total Phase I jar tests).

Results of Phase I are presented in Table 2, using symbols to represent the average of the two results determined for each soil and flocculant combination. pH results are provided in the Addendums that contain the analysis log sheets; results generally did not shift significant though a jar test, with most final pH levels around 8.

From these Phase I results, seven flocculants were deemed appropriate for further study based on turbidity reduction and representativeness of chemical class. These seven flocculants were continued into Phase II testing.

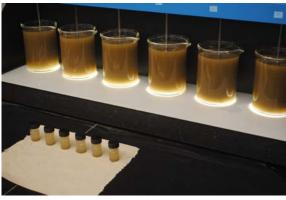
Table 1. Flocculants and doses used in this study.

Flocculant Chemical Source		Manufacturer Reported Type ¹	Phase I Dosed Concentration	Phase II Dosed Concentration Range		
Alum	Hawkins, Inc.,	Aluminum Sulfate	0.01 mL/L	0 – 0.1 mL/L		
Ferric Chloride 38%	St. Paul, MN	Ferric Chloride	0.02 mL/L	0 – 0.2 mL/L		
Biostar CH	Brock White, St. Paul, MN	Chitosen	0.002 mL/L	0 – 0.02 mL/L		
Floc Flocculating Agent (called in this analysis Standard Contracting Floc)	Standard Contracting, Inc., Hampton,	Bentonite	0.2 g/L	0 – 2.0 g/L		
Floc W Flocculating Agent (called in this analysis Standard Contracting Floc W)	MN, agent for Innovative Turf Solutions, Cincinnati, OH	Bentonite	0.2 g/L	n/a²		
AH ³ 6447		Polyamine	0.002 mL/L	0 - 0.002  mL/L		
AH 6547		Polyamine	0.002 mL/L	n/a ²		
AH 7747		Polymer blend	0.002 mL/L	n/a ²		
AH 117		Aluminium Chlorohydrate / Polyamine	0.002 mL/L	0 – 0.002 mL/L		
AH 457	Hawkins, Inc., St. Paul, MN	Aluminium Chlorohydrate / Polyamine	0.002 mL/L	n/a²		
AH 820		Cationic polymer	0.002 mL/L	0 – 0.002 mL/L		
AH 846		Cationic polymer	0.002 mL/L	n/a ²		
AH 852		Cationic polymer	0.002 mL/L	n/a ²		
AH 882		Cationic polymer	0.002 mL/L	n/a ²		
Tramfloc 111		Anionic polyacrylamide emulsion	0.002 mL/L	n/a ²		
Tramfloc 133	St. Paul, MN  Brock White, St. Paul, MN  Standard Contracting, Inc., Hampton, MN, agent for Innovative Turf Solutions, Cincinnati, OH	Nonionic polyacrylamide emulsion	0.002 mL/L	n/a ²		
Tramfloc 317		Cationic polyacrylamide emulsion	0.002 mL/L	n/a²		
Tramfloc 342		Cationic polyacrylamide emulsion	0.002 mL/L	n/a ²		
Tramfloc 550	1	Polyamine	0.002 mL/L	n/a ²		
Tramfloc 723		Polyamine	0.002 mL/L	n/a ²		
Tramfloc 865A		Polyamine	0.002 mL/L	n/a ²		

Notes: 1: Manufacture reported type from MSDS or  $\underline{www.NSF.org}$ . 2: n/a: not applicable, flocculant not tested in Phase II. 3: AH is notation for Aqua Hawk.



a) Preparation of turbid sediment-filled water.



b) Initial turbidity measurement prior to flocculant addition.



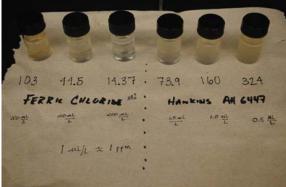
c) Simultaneous dosing of flocculants.



d) Rapid (flash) mix to thoroughly blend flocculant.



e) Slow mix for floc formation.



f) Final turbidity measurement at end of test.

Figure 3. Jar testing for flocculant response and/or dose evaluation.



Figure 4. Floc formation (third jar from left in Figure 4) approximately 1 minute after flocculant dosing (end of rapid mix).

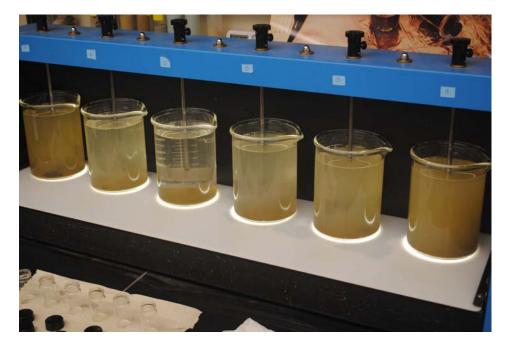


Figure 5. End of test responses for two flocculants across three dose rates for each flocculant; Worthington subsoil sediment mixture. From left to right: Ferric chloride at 0.05 mL/L (103 NTUs), 0.1 mL/L (44.5 NTUs), and 0.2 mL/L (14.37 NTUs); AH 6447 at 0.0015 mL/L (73.9 NTUs), 0.001 mL/L (160 NTUs), and 0.0005 mL/L (324 NTUs).

In Phase II of the dose evaluation, dose rates were varied by increasing or decreasing the amount of flocculant stock added to the soil and water mixture. All 57 soils were tested with the seven selected flocculant chemicals at the dose rate ranges provided in Table 1 (2,133 Phase II jar tests). Results of the Phase II evaluation are presented in Appendix C. Phase II results are graphed with turbidity as a function of flocculant dose rate. Note that the blank (no flocculant added) results are repeated in every graph of a given soil, representing a dose rate of 0 mL/L or 0 g/L.

#### 3.3: Evaluation of Results

Soils have specific turbidity characteristics, as shown by the results from the blank (no flocculent) treatments. As commonly observed in the geotechnical and sedimentation disciplines, soil sediment turbidity is a function of several soil characteristics: fines content, particle size and distribution, particle surface charge and mineral composition. Some soils may have insufficient fines to cause turbidity beyond what can be treated by gravity sedimentation alone. In this study, soil sediment turbidity ranged from a high of 1000 NTUs to a low of 35.9 NTUs, averaging 322 NTUs with a standard deviation of 265 NTUs (relative standard deviation of 82%).

Each flocculent chemical has a highly specific characterization of effectiveness across the range of soil types. Specific dose by turbidity relationships exist for each flocculent chemical and each soil, particularly for polymer compounds that can either work very well or make turbidity worse due to specific surface charge interrelationships.

When effective, dose levels can vary between 0.2 to 100 ppm. This does range is quite large, given the potential for wide cost differences between flocculent chemicals (anecdotally noted as ranging from \$2 to \$50 per gallon).

Flocculent chemicals found to be broadly effective include ferric chloride, Aqua Hawk 6447 and Tramfloc 865A. Ferric chloride was found to treat turbidity consistently to below 100 NTUs for all soils considered in this study, and typically to below 50 NTUs for most soils, at a dose of 0.2 mL/L (approximately 200 ppm, or 70 gallons of ferric chloride per acre foot of water to be treated).

Levels of pH were not adversely impacted by the dose amounts used in either Phase I or II, as values remained between 7 and 8.5 for most final pH measurements. These values were consistent with the blank treatments for the soils, and suggest sufficient alkalinity exists in waters made up from the soils considered in this study to fully buffer any pH effects from flocculent treatment at the dose levels considered.

		Flocculant Chemical									
		×				47	47	47	7	7	0
Soil	Standard Contr	Standard Contr W	Biostar CH	Alum	Ferric Chloride	Hawkins AH 6447	Hawkins AH 6547	Hawkins AH 7747	Hawkins AH 117	Hawkins AH 457	Hawkins AH 820
Ada Subsoil	~	\	\	\	\	+	+	~	~	~	~
Ada Topsoil	~	~	\	\	~	~	•	\	\	\	~
Brandon Subsoil	\	\	\	\	\	+	+	\	\	\	\
Brandon Topsoil	\	\	\	\	\	~	+	\	\	\	\
Carlton Subsoil	\	\	\	\	\	+	\	\	•	•	?
Carlton Topsoil	\	\	\	~	\	+	+	\	\	~	1
Cook Subsoil	\	\	\	\	\	+	\	\	+	~	~
Cook Topsoil	~	~	~	\	\	+	•	~	~	•	+
Coon Rapids Subsoil	~	\	\	\	\	+	+	\	~	\	~
Coon Rapids Topsoil	•	~	~	•	•	+	+	•	•	•	+
E Grand Forks Subsoil	\	\	\	\	\	~	?	\	\	~	\
E Grand Forks Topsoil	\	\	\	\	\	+	+	\	\	\	\
Foley Subsoil	\	\	\	\	\	+	+	\	~	+	+
Foley Topsoil	\	~	~	\	~	+	+	\	\	~	~
Grand Rapids Subsoil	•	•	+	+	+	+	+	+	+	+	+
Grand Rapids Topsoil	•	•	•	~	~	+	+	•	•	•	+
Hampton Subsoil	?	?	\	\	\	+	\	\	+	•	\
Hampton Topsoil	\	\	\	\	\	\	~	\	\	\	\
Houston Riverbed	\	\	\	\	\	+	+	\	\	\	~
Houston Topsoil	•	•	•	•	•	•		~	~	•	•
Kandiyohi Subsoil	~	~	~	+	+	+	+	+	+	+	+
Kandiyohi Topsoil	~	~	~	~	~	+	+	•	~	•	•
Lakeville Subsoil	\	\	~	+	+	+	+	+	+	+	+
Lakeville Topsoil	~	Ì	~	~	~	+	+	~	~	•	•
Lindstrom Subsoil	\	Ì	\	\	+	\	+	\	\	~	~
Lindstrom Topsoil	Ì	Ì	Ì	Ì	\	•	+	Ì	Ì	\	~
Mankato Topsoil	~	·	~	·	+	+	+	+	+	+	+
Mapleton Topsoil	\	١	١	~	+	+	+	•	+	+	+
Moorhead Subsoil	+	+	•	+	+	+	+	+	+	+	+
Moorhead Topsoil	1	\	\	\	\	~	+	١	1	\	١

Legend: + turbidity < 100 NTU

• turbidity 100 – 150 NTU

? conflicting results

turbidity 150 – 250 NTU
 turbidity > 250 NTU

Table 2. Phase I analys	is: flo	occul	ant e	ffect	ivene	ess so	ereen	ing r	esult	s, co	n't.
		Flocculant Chemical									
Soil	Standard Contr	Standard Contr W	Biostar CH	Alum	Ferric Chloride	Hawkins AH 6447	Hawkins AH 6547	Hawkins AH 7747	Hawkins AH 117	Hawkins AH 457	Hawkins AH 820
N Mankato Subsoil	\	\	\	\	\	\	~	+	\	\	\
N Mankato Topsoil	\	\	\	\	\	\	\	\	\	\	\
New Sweden Topsoil	\	\	\	\	\	+	+	\	~	+	+
Olivia A Subsoil	\	\	\	\	\	+	+	\	\	\	\
Olivia A Topsoil	•	•	•	?	?	•	+	•	•	+	+
Olivia B Clay	\	\	\	\	\	+	+	\	\	~	\
Owatonna Clay	~	~	•	~	•	+	+	•	•	•	+
Owatonna Topsoil	\	\	\	~	+	+	•	+	•	•	+
Perham Subsoil	\	?	\	\	\	+	+	\	\	?	?
Perham Topsoil	\	\	\	\	\	+	+	\	\	~	~
Pipestone Subsoil	\	\	~	~	~	+	+	~	•	+	+
Pipestone Topsoil	+	+	+	+	+	+	+	+	+	+	+
Ramsey Peat	+	+	+	+	+	+	+	+	+	+	+
Rockford Cornfield	+	+	+	+	+	+	+	+	+	+	+
Rockford Subsoil	\	\	\	\	\	+	+	\	\	\	\
Rockford Topsoil	/	\	\	\	\	\	\	\	\	\	\
Sandstone Subsoil	\	\	\	\	\	+	+	•	+	+	+
Sandstone Topsoil	•	•	•	~	~	+	+	•	+	+	+
Stillwater Sediment	\	\	\	~	+	+	+	+	+	+	+
Tamarack Clay	\	\	\	~	+	+	•	~	+	+	~
Tamarack Peat	+	+	+	+	~	+	+	•	~	+	+
Tamarack Topsoil	\	~	~	+	+	+	+	+	+	+	+
Wabasha Sub Box	~	~	~	~	~	+	+	~	~	\	•
Wabasha Subsoil	\	\	~	~	~	+	+	~	+	+	+
Wabasha Topsoil	\	\	\	\	\	\	•	\	\	\	\
Worthington Topsoil	\	~	1	\	1	•	+	~	~	~	~
Worthington Subsoil	\	~	\	+	+	+	+	+	+	+	+

 $\begin{tabular}{ll} $\sim$ & turbidity $150-250$ NTUs \\ $\backslash$ & turbidity $>250$ NTUs \\ \end{tabular}$ 

Table 2. Phase I analysis: flocculant effectiveness screening results, con't.										
		Flocculant Chemical								
Soil	Hawkins AH 846	Hawkins AH 852	Hawkins AH 882	Tramfloc 111	Tramfloc 133	Tramfloc 317	Tramfloc 342	Tramfloc 550	Tramfloc 723	Tramfloc 865A
Ada Subsoil	~	•	~	+	+	•	+	~	•	+
Ada Topsoil	۲	~	٧	+	+	•	•	~	•	•
Brandon Subsoil	/	\	\	~	\	\	\	\	\	~
Brandon Topsoil	/	\	\	?	?	\	\	\	\	\
Carlton Subsoil	~	\	~	~	~	\	\	\	•	+
Carlton Topsoil	•	\	~	•	~	•	+	\	~	+
Cook Subsoil	•	\	\	~	•	٧	~	\	\	~
Cook Topsoil	+	~	٧	+	+	+	+	~	•	+
Coon Rapids Subsoil	~	\	\	~	~	~	~	\	~	~
Coon Rapids Topsoil	+	~	٧	+	+	•	+	~	~	~
E Grand Forks Subsoil	\	\	\	~	~	\	?	\	\	\
E Grand Forks Topsoil	\	\	\	~	\	\	\	\	~	~
Foley Subsoil	+	\	+	~	\	~	~	\	+	+
Foley Topsoil	~	\	~	+	~	~	~	\	~	~
Grand Rapids Subsoil	+	+	+	+	+	+	+	+	+	+
Grand Rapids Topsoil	+	~	•	+	+	•	+	•	+	+
Hampton Subsoil	\	\	\	?	\	\	\	\	+	+
Hampton Topsoil	\	\	\	\	•	\	\	\	\	\
Houston Riverbed	~	•	•	+	+	+	+	~	+	+
Houston Topsoil	+	•	٧	+	+	+	+	•	•	+
Kandiyohi Subsoil	+	+	+	+	+	+	+	+	+	+
Kandiyohi Topsoil	+	~	٧	+	+	+	+	\	•	+
Lakeville Subsoil	+	~	٧	+	+	+	+	\	•	+
Lakeville Topsoil	+	+	+	+	+	+	+	+	+	+
Lindstrom Subsoil	+	\	\	~	+	•	+	~	•	+
Lindstrom Topsoil	~	~	\	•	•	~	•	\	\	~
Mankato Topsoil	+	+	+	~	+	+	•	•	+	+
Mapleton Topsoil	+	+	+	+	+	+	+	~	+	+
Moorhead Subsoil	+	+	+	+	+	+	+	~	+	+
Moorhead Topsoil	\	\	\	\	\	\	\	\	\	~

 $\begin{tabular}{ll} $\sim$ & turbidity $150-250$ NTUs \\ $\backslash$ & turbidity $>$ 250$ NTUs \\ \end{tabular}$ 

Table 2. Phase I analysis: flocculant effectiveness screening results, con't.										
		Flocculant Chemical								
Soil	Hawkins AH 846	Hawkins AH 852	Hawkins AH 882	Tramfloc 111	Tramfloc 133	Tramfloc 317	Tramfloc 342	Tramfloc 550	Tramfloc 723	Tramfloc 865A
N Mankato Subsoil	\	\	\	\	\	\	\	\	\	\
N Mankato Topsoil	\	\	\	\	\	\	\	\	\	~
New Sweden Topsoil	+	•	•	•	•	•	+	~	+	+
Olivia A Subsoil	•	\	\	~	\	\	\	\	•	+
Olivia A Topsoil	+	•	•	+	+	+	+	•	+	+
Olivia B Clay	•	\	\	•	\	~	~	\	+	+
Owatonna Clay	+	+	+	+	+	+	+	+	+	+
Owatonna Topsoil	+	~	•	+	+	~	+	~	~	~
Perham Subsoil	+	+	+	+	+	+	+	•	+	+
Perham Topsoil	~	~	~	•	•	•	+	•	+	+
Pipestone Subsoil	+	+	+	+	+	+	+	•	+	+
Pipestone Topsoil	+	+	+	+	+	+	+	+	+	+
Ramsey Peat	+	+	+	+	+	+	+	+	+	+
Rockford Cornfield	+	+	+	+	+	+	+	~	+	+
Rockford Subsoil	\	\	\	\	\	\	\	\	\	~
Rockford Topsoil	\	\	\	\	\	\	\	\	\	\
Sandstone Subsoil	\	\	\	\	\	\	\	\	+	+
Sandstone Topsoil	+	•	+	+	+	+	+	+	+	+
Stillwater Sediment	+	•	+	+	+	+	+	\	+	+
Tamarack Clay	+	\	~	\	\	\	\	\	+	+
Tamarack Peat	+	+	+	+	+	+	+	•	•	+
Tamarack Topsoil	+	+	+	+	+	+	+	•	+	+
Wabasha Sub Box	~	•	•	+	+	+	+	\	+	+
Wabasha Subsoil	+	+	+	+	+	+	+	~	+	+
Wabasha Topsoil	/	/	?	/	/	/	~	~	?	~
Worthington Topsoil	~	~	\	+	+	+	+	~	•	•
Worthington Subsoil	+	+	+	+	+	+	+	•	+	+

 $\begin{tabular}{ll} $\sim$ & turbidity $150-250$ NTUs \\ & turbidity $>250$ NTUs \\ \end{tabular}$ 

#### Chapter 4: Design Feature and Scale Study

Flocculation effectiveness is dependent upon the surficial chemical attraction of particles, the dispersion of the flocculant and the contact of the particles. These factors are normally expressed as: (a) what flocculant to add, (b) how to mix the flocculant, and (c) how to mix the particles in the presence of the flocculant. Chapter 3 addressed the question of what flocculant to add. This chapter addresses methods both to mix the flocculant and to mix the particles, using methods and techniques suited for construction fieldwork rather than laboratory determinations or even municipal water treatment.

The two mixing efforts are very different in both appearance and imparted energy. Flocculant mixing is done using "flash" or "rapid" mixing, in which the chemical is quickly and fully dispersed within the fluid. Rapid mixture of flocculants needs very high mixing velocity gradients to cause particles to contact each other, rather than just ride along and not aggregate. High mixing power may be required. However, ferric chloride in particular can mix into water very quickly and may require a mixing time of less than 10 seconds total, hence the term "flash" mixing. Rapid mixing may be the most important factor for coagulation (chemical) efficiency (Davis, 2010).

Particle mixing is done using "slow" mixing, in which the particle filled water is gently stirred such that flocculant-rich particles may bump into each other and stick (creating flocs); the energy level is low enough such that the particle flocs aggregate but do not break apart. Slow mixing to encourage floc aggregation from particles in treated waters must be gentle, with low to very low power such that particles come into contact but flocs do not break apart or shear. Slow mixing may be the most important factor for particle size and removal. Slow mixing times are typically 20 to 30 minutes, with shorter times in this range appropriate for summertime water temperatures and longer times appropriate for wintertime temperatures. Water velocities during slow mixing should be 0.5 to 1.0 ft/s to reduce the potential for inadvertent sedimentation that could clog the slow mixing process (Davis, 2010).

#### 4.1: Methods

#### 4.1.1: Turbid Water Stock Preparation

Turbid water stock was prepared by adding selected soil to water at a rate of 25 g/L and mixing (Figure 6). Water was mixed continually for all mixing experiments unless an experiment incorporated settlement during a sedimentation period. Mixing of stock was done using air injection, with particular emphasis on air circulation near the bottom of the stock container.



Figure 6. Turbid water stock prepared in a 30 gallon polyethylene container from soil sediment and tap water mixed by injection of air into bottom of fluid column. Staff gage units are feet of depth.

#### 4.1.2: Turbidity Measurement

Turbidity was measured using 20-ml specimens placed into optical-quality glass vials analyzed in an Oakton Instruments T-100 turbidity meter with a range of 0 to 1000 NTUs (Figure 7).





Figure 7. Turbidity specimens and turbidity meter.

#### 4.2: Evaluation of Flocculant Treatment Methods and Technologies

Four separate evaluations were done of flocculant treatment methods and technologies:

- 1) Mixing technology evaluation, in which different mixing methods and technologies were tested and compared;
- 2) Open pond treatment model evaluation, in which potential pond treatments were evaluated;
- 3) Gravity settlement evaluation, in which depths up to 6 feet were evaluated for occurrence of Type II settlement and sedimentation effectiveness related to extra depth; and,
- 4) Flocculant aided filtration evaluation, in which filtration effectiveness was evaluated for different conditions of flocculation, slow mix and sedimentation.

Each of these evaluations are described below.

#### *4.2.1: Mixing Technology Evaluation*

Mixing technologies were evaluated in a drainage channel/drainage ditch model made from 4-inch wide plastic rain gutter, 10 feet long and placed at an approximate 3% slope (Figure 8). Turbid water was pumped from a 25 gallon stock at a rate of 500 mL/min (+/- 20 mL/min), calibrated by measuring the volume in a graduated cylinder filled per unit of time. Mixing equipment or material was placed in the gutter beginning near the uphill point. The gutter provided laminar (not turbulent) flow representing a condition of no mixing energy imparted to the flow. A solution of ferric choride was typically injected at a rate of 25 mL/min to the mixing equipment, although some mixing technologies were soaked and allowed to passively exude with ferric chloride.

Discharge at the low end of the gutter was to a 20 gallon glass aquarium that acted as a sedimentation tank with a hydraulic retention time of approximately 80 minutes (Figure 9). The aquarium was tilted lengthwise and positioned to discharge over the low edge to a waste container. An inlet zone of the aquarium sedimentation tank was created using a 1-gallon plastic box approximately 4 inches deep by 6 inches long, perforated on the upstream side to limit turbulence within the sedimentation tank and to distribute flow across the whole sedimentation tank width.

Samples were taken at 10-minute intervals from the stock tank, the gutter channel discharge (representing the end of the mixing zone) and the discharge of the aquarium sedimentation tank (representing the point of complete treatment). Samples were measured for turbidity.

Results of the mixing experiment are presented in Appendix D.

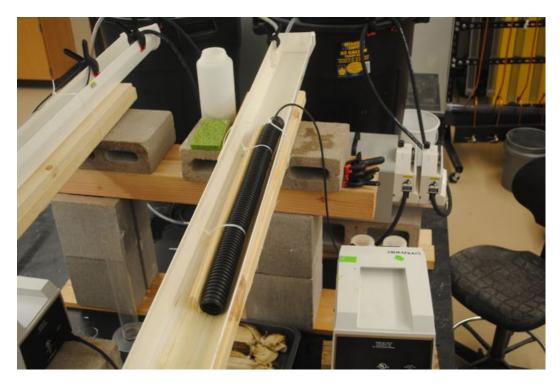


Figure 8. Twin channels for evaluation of mixing technologies, with 2-foot length of corrugated tubing shown in channel on right. Note flocculant stock injection line consisting of small-diameter black tubing at top of corrugated tubing.



Figure 9. Twin 20-gallon sedimentation tanks, with orange snow fence submerged in inlet zone to reduce short-circuiting. Note two sampling locations: the end of channel/prior to inlet and the overflow weir of sedimentation tank.

Mixing technologies evaluated included:

- Laminar flow;
- 1 1/4th inch diameter corrugated pipe, 2 and 8 feet long;
- ½ inch gravel, 2 and 6 feet long;
- Baffled flow, consisting of a gutter section 3.5 feet long with baffles set approximately 3 inches apart alternating side to side with approximately 80% flow blockage;
- Floc pillow, consisting of a sponge material cut to a 40 mm x 190 mm x 25 mm size and injected with 6.0 mL undiluted ferric chloride (supplied at 35% purity);
- Floc log, consisting of five sponge sections cut to the same dimensions as with the floc pillow and soaked to saturation by squeezing and releasing in a pan of 100 mL ferric chloride diluted 10x with tap water;
- Drip line mixing, consisting of an approximately 4 foot length of commercial "soaker hose" placed above the flow to distribute the ferric chloride solution;
- Sump recirculation, consisting of a 1-gallon polyethylene pail with a 8 gpm centrifugal pump recirculating flow, with discharge via the pail overflow;
- Turbulent air injection, consisting of a 3/8 inch pressurized air line fed into the flow transiting a 1 L HDPE bottle;
- Double bladed mixer, consisting of twin, overlapping turbine mixer blades circulating in the flow transiting a 1 gallon polyethylene pail; and,
- Single bladed mixer, consisting of a single turbine mixer blade circulating in the flow transiting a 1 L HDPE bottle.

Technologies are shown in photographs accompanying the applicable results provided in Appendix D. All ferric chloride dose rates were 0.2 mL/L, except for the floc pillow and floc log technologies which were considered passive dosing and therefore with indeterminate doses.

Results of the mixing technology evaluation are provided in Table 3, using a relative performance scale based on discharge turbidity levels.

#### Table 3. Relative performance of mixing technologies.

High performance (discharge turbidity < 50 NTUs)

- 6 feet of gravel (91% removal)
- 8 feet of corrugated pipe (85% removal)
- Turbulent air injection (95% removal)
- Single bladed mixer (95% removal)

#### Medium performance (discharge turbidity 51 - 150 NTUs)

- 2 feet of corrugated pipe (78% removal)
- Baffled flow (64% removal)
- Floc log (82% removal)
- Drip line mixing (73% removal)
- Sump recirculation (59% removal)
- Sedimentation only/no flocculation (removal varies by soil)

#### Low performance (discharge turbidity 151 - 250 NTUs)

- Laminar flow (57% removal)
- Floc pillow (66% removal)
- Double bladed mixer (76% removal)

#### Very low performance (discharge turbidity > 250 NTUs)

• 2 feet of gravel (36% removal)

#### 4.2.2: Open Pond Treatment Model Evaluation

Larger scale treatment tests were performed with the goal to model open pond treatment and evaluate effectiveness. Two pond surrogates were used: a 6-inch deep, 27-gallon kiddie pool and a 4-foot deep, 80 gallon recycling bin-tank (Figure 10). Turbid water was created by mixing Mankato topsoil into the water at a rate of 25 g/L. Ferric chloride was directly applied to the surface of the pond surrogates. Mixing was done for 2 minutes using either a vertically oriented, drill-mounted



a) Preparation of turbid sediment-filled water.



b) Flocculant addition.



c) Mixing by drill-mounted verticallyoriented turbine to 6-inch deep water in tank made from kiddie pool.



d) Mixing by air injection to 6-inch deep water in tank made from kiddie pool.



e) Mixing by air injection to 36-inch deep water in tank made from recycling bin.



f) Sample collection for turbidity measurement.

Figure 10. Open pond modeling and treatment evaluation.

turbine blade mixer or pressurized air injection. Upon completion of the mixing, samples were taken from the top 3 inches of the pond surrogate water and turbidity measurements made. Turbidity measurements continued for at least 15 minutes and until the measurements were stable and relatively consistent. Results are provided in Appendix E.

The ferric chloride dose of 0.2 mL/L (equivalent to 66 gallons/acre foot) achieved a treatment from 505 and 351 NTUs down to 20 and 45 NTUs in the 4-foot deep tank with 2 minutes of pressurized air injection mixing and the 6-inch deep tank with 2 minutes of turbine blade mixing, respectively. An earlier experiment with a dose of 0.1 mL/L did not achieve much reduction at all (results not provided as the test was not extended beyond 5 minutes).

A repeated test with the 4-foot tank with 2 minutes of pressurized air injection mixing resulted in a reduction from 727 NTUs to 57 NTUs, similar results to the first test. However, using 2 minutes of pressurized air injection mixing with the 6-inch deep tank did not achieve similar results as the turbine blade mixed trial, as a reduction from 530 NTUs to only 280 NTUs was achieved in 26 minutes of sedimentation time. It appears that the pressurized air injection mixing needed more vertical length (depth) in the water column of an open pond to effectively mix the ferric chloride or cause particle aggregation. This result somewhat contrasts with the result for pressurized air injection in the mixing technologies evaluation, which achieved high performance with turbidity consistently below 50 NTUs. However, the pressurized air mixing in the previous evaluation was done in a confined volume (a 1 L bottle), not in an open water column.

#### 4.2.3: Gravity Settlement Evaluation

Sedimentation column treatment tests were performed to model deeper sedimentation and evaluate effectiveness, particularly whether Type II settlement was occurring with the ferric chloride dose used in this study. Columns were constructed of 4 inch diameter PVC piping and were 8-feet tall with sampling ports at 0, 2, 4 and 6 feet depths (Figure 11). Sampling ports consist of a 3/8 inch ball valve connected to a saddle tap, with a  $\frac{1}{2}$  inch hole drilled through the pipe wall.

Turbid water was created by mixing Mankato topsoil into 8 gallons of water at a rate of 25 g/L. Ferric chloride was directly applied to the surface of the turbid water at a rate of 0.2 mL/L or 6 mL/8 gallons. Mixing was done for 2 minutes using a vertically oriented, drill-mounted turbine blade mixer. Upon completion of the mixing, water was poured into the sedimentation column. Sampling and turbidity measurements began immediately after and continued for 1 hour at 10 minute increments. Nine tests were performed. Results are provided in Appendix F as both turbidity and removal percentage.



Figure 11. Sedimentation columns. Only bottom third of each 8-foot column shown, with 6-foot depth valved sampling port (attached to blue saddle tap) and two sludge clean outs. Open top and 2-foot and 4-foot depth valved sampling ports not shown.

Results generally showed an achievement of 80% turbidity reduction by 30 minutes, and an average of 94% removal at 60 minutes. Results for times less than 30 minutes suggest substantial variability and poor removal. Type II settlement was observed, as the results are not strictly linear and do show acceleration of removal with increasing time and depth. These results agree with the recommendation of Davis (2010) that a 60 minute sedimentation period should be used for best removals and sufficient factor of safety.

#### 4.2.4: Flocculant-Aided Filtration Evaluation

The additional effectiveness in turbidity removal provided by filtration was evaluated for flocculant treated waters, using both granular (sand) and membrane (geotextile) filtration. Flocculant-aided filtration was evaluated by comparing flocculant treated waters after filtration with non-flocculated (blank) waters after filtration and determining the reduction provided by the flocculant addition. Three different dose levels of ferric chloride were used: 1.25, 2.5 and 5.0 mL/L. The dose range was a response to commentary in Crittenden (2005) that flocs might catch on the filter media then continue to aggregate as additional particles approach the filter, even with incomplete sedimentation after flocculation.

Three filtration processes were evaluated:

- Conventional filtration, consisting of rapid mix, slow mix and sedimentation prior to filtration;
- Direct filtration, consisting of rapid mix and slow mix but no sedimentation prior to filtration:
- In line filtration, consisting of rapid mix only prior to filtration.

As discussed previously, conventional filtration is considered appropriate for waters with turbidities up to 1000 NTUs, while direct and in line filtration approaches are considered appropriate for waters with turbidities less than 15 NTUs due to the effort involved in backwashing or cleaning filters.

As shown in Figure 12, six granular filters were constructed:

- #4 Sieve (passing) Granular Activated Carbon (#4 GAC);
- Sand passing #4 sieve retained on a #10 sieve (4-10 sand);
- Sand passing #10 sieve retained on a #20 sieve (10-20 sand);
- Garnet sand passing a #36 sieve retained on a #54 sieve (54 garnet);
- #4 GAC on top of 10-20 sand, in equal thicknesses; and,
- #4 GAC on top of 54 garnet, in equal thicknesses.

Each granular filter was 2 inches in diameter. Filter media was a total of 6 inches in depth for each filter, whether a mono or a dual media filter. The filter apparatus was a manufactured apparatus designed to pair with the 6-place jar tester used in this study, both manufactured by Phipps and Bird (Richmond, VA).

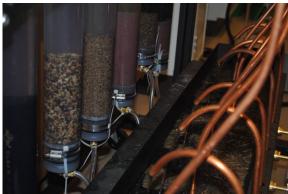
Turbid water was prepared in 2 L batches using Mankato topsoil at 25 g/L. After the sediment was thoroughly mixed, the dose of ferric chloride was added and mixed for 1 minute in the rapid mix phased. Slow mixing for 20 minutes and sedimentation for 20 minutes were employed, depending upon the filtration process being tested. Water was then released through the filter column to pass downward across the filter media. During the middle of the filter run, triplicate samples were taken from the filtrate and analyzed for turbidity.

Membrane filtration evaluation had the same procedure, except that the membrane filters were placed in filter cones for support beneath the stock containers and above beakers that served to collect the filtrate (Figure 13). Three membranes were evaluated in this study:

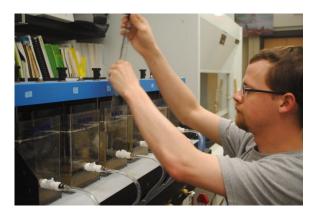
- Geotex 401 non-woven needle punched geotextile (5 oz/sy, Propex, Inc.);
- Geotex 104 F woven geotextile (7 oz/sy, Propex, Inc., Chattanooga, TN); and
- Paper filter, consisting of a Melita #4 coffee filter.



a) Preparation of turbid sediment-filled water for six different filter columns.



b) Filter columns with 4-inches of granular media in a range of particle sizes.



c) Sampling of flocculant-treated water after being allowed to settle.



d) Sampling of flocculant-treated filtered water (turbidity ~10 NTU).

Figure 12. Flocculation-aided sand filtration evaluation.

Results of the granular filter evaluations and the membrane filter evaluations are summarized in Tables 4 to 9 and Tables 10 and 12, respectively. Full results are provided in Appendix G. The following observations may be made about the results:

- Ferric chloride improves filtration removal of turbidity;
- Granular filter materials are all good, no clear benefit to one filter material than other in regards to turbidity removal, though #36 garnet and #54 garnet both are slower hydraulically and therefore may not be suitable for construction site and stormwater runoff applications;
- Membrane filter materials were all good for conventional filtration, though the nonwoven geotextile did not provide results as high as the woven geotextile and paper filters;
- Dose levels higher than 1.25 mL/L appear to provide insignificant improvement for turbidity removal with conventional filtration;

- Direct and in line filtration are not as effective as conventional filtration, and generally require larger doses of ferric chloride, for the conditions investigated;
- Velocity through the filters ranged from 0.3 to 1 L/min for a rate of flow of about 4 to 11 gpm/ft², which matches the guidelines presented in Crittenden (2005). Note that these rates assume the granular filter can be backwashed and does not clog; and,
- Filter run time, defined as the time of operation until filter clogging, was not determined in this experiment.



Figure 13. Membrane filtration apparatus with six parallel treatment lines, each consisting of mixing tank, valved delivery tubing, filter cone and filtrate collection beaker.

Table 4. Effectiveness provided by flocculant addition when filtering through 6 inches of #4 sieve retained granulated activated carbon (#4 GAC) as filter media.

	Turbidity of Blank	Average F	Filtrate Turbidity (NTUs) ⁴		Reduction Due to Flocculant Addition		
Filtration Method	(NTUs)	1.5 mL/L FeCl ₃	2.5 mL/L FeCl ₃	5 mL/L FeCl ₃	1.25 mL/L FeCl ₃	2.5 mL/L FeCl ₃	5 mL/L FeCl ₃
Conventional Filtration ¹	99.7	11.9	33.1	19.7	88.1%	66.8%	80.2%
Direct Filtration ²	132.1	100.0	69.4	91.6	24.3%	47.5%	30.7%
In Line Filtration ³	155.1	205.6	138.7	120.3	-32.6%	10.6%	22.4%

- 1) Conventional filtration = 1 minute rapid mix, 20 minutes slow mix, 20 minutes sedimentation then filter.
- 2) Direct filtration = 1 minute rapid mix, 20 minutes slow mix then filter.
- 3) In line filtration = 1 minute rapid mix then filter.
- 4) Three replicates analyzed for average filtrate turbidity.

Table 5. Effectiveness provided by flocculant addition when filtering through 6 inches of #4 sieve passing #10 sieve retained sand (4-10 sand) as filter media.

	Turbidity of Blank	Average F	iltrate Turbidit	rate Turbidity (NTUs) ⁴		Reduction Due to Flocculant Addition		
Filtration Method	(NTUs)	1.5 mL/L FeCl ₃	2.5 mL/L FeCl ₃	5 mL/L FeCl ₃	1.25 mL/L FeCl ₃	2.5 mL/L FeCl ₃	5 mL/L FeCl ₃	
Conventional Filtration ¹	54.4	11.2	9.1	13.5	79.4%	83.3%	75.2%	
Direct Filtration ²	98.7	61.5	31	9.2	37.7%	68.6%	90.7%	
In Line Filtration ³	129.8	148.4	167.1	172.8	-14.3%	-28.7%	-33.1%	

- 1) Conventional filtration = 1 minute rapid mix, 20 minutes slow mix, 20 minutes sedimentation then filter.
- 2) Direct filtration = 1 minute rapid mix, 20 minutes slow mix then filter.
- 3) In line filtration = 1 minute rapid mix then filter.
- 4) Three replicates analyzed for average filtrate turbidity.

Table 6. Effectiveness provided by flocculant addition when filtering through 6 inches of #10 sieve passing #20 sieve retained sand (10-20 sand) as filter media.

	Turbidity of Blank	Average F	iltrate Turbidit	y (NTUs) ⁴	Reduction Due to Flocculant Addition		
Filtration Method	(NTUs)	1.5 mL/L FeCl ₃	2.5 mL/L FeCl ₃	5 mL/L FeCl ₃	1.25 mL/L FeCl ₃	2.5 mL/L FeCl ₃	5 mL/L FeCl ₃
Conventional Filtration ¹	64.2	9.7	24.6	13.5	84.9%	61.7%	79.0%
Direct Filtration ²	110.0	65.1	53.9	16.1	40.8%	51.0%	85.4%
In Line Filtration ³	177.0	301.5	138.8	137.7	-70.3%	21.6%	22.2%

- 1) Conventional filtration = 1 minute rapid mix, 20 minutes slow mix, 20 minutes sedimentation then filter.
- 2) Direct filtration = 1 minute rapid mix, 20 minutes slow mix then filter.
- 3) In line filtration = 1 minute rapid mix then filter.
- 4) Three replicates analyzed for average filtrate turbidity.

Table 7. Effectiveness provided by flocculant addition when filtering through 6 inches of #54 sieve retained garnet sand (#54 garnet) as filter media.

	Turbidity of Blank	Average F	Average Filtrate Turbidity (NTUs) ⁴			Reduction Due to Flocculant Addition		
Filtration Method	(NTUs)	1.5 mL/L FeCl ₃	2.5 mL/L FeCl ₃	5 mL/L FeCl ₃	1.25 mL/L FeCl ₃	2.5 mL/L FeCl ₃	5 mL/L FeCl ₃	
Conventional Filtration ¹	52.6	9.2	31.5	10.7	82.5%	40.1%	79.7%	
Direct Filtration ²	87.7	69.6	62.9	11.3	20.6%	28.3%	87.1%	
In Line Filtration ³	200.7	219.7	308.0	200.4	-9.5%	-53.5%	0.1%	

- 1) Conventional filtration = 1 minute rapid mix, 20 minutes slow mix, 20 minutes sedimentation then filter.
- 2) Direct filtration = 1 minute rapid mix, 20 minutes slow mix then filter.
- 3) In line filtration = 1 minute rapid mix then filter.
- 4) Three replicates analyzed for average filtrate turbidity.

Table 8. Effectiveness provided by flocculant addition when filtering through 3 inches of #4 sieve granulated activated carbon and 3 inches of #10 sieve passing #20 sieve retained sand (#4 GAC + 10-20 sand) as filter media.

	Turbidity of Blank	Average F	iltrate Turbidit	y (NTUs) ⁴	Reduction Due to Flocculant Addition		
Filtration Method	(NTUs)	1.5 mL/L FeCl ₃	2.5 mL/L FeCl ₃	5 mL/L FeCl ₃	1.25 mL/L FeCl ₃	2.5 mL/L FeCl ₃	5 mL/L FeCl ₃
Conventional Filtration ¹	103.2	16.3	25.3	11.7	84.2%	75.5%	88.7%
Direct Filtration ²	99.7	112.3	58.5	30.7	-12.6%	41.3%	69.2%
In Line Filtration ³	222.5	263.7	361.5	189.4	-18.5%	-62.5%	14.9%

- 1) Conventional filtration = 1 minute rapid mix, 20 minutes slow mix, 20 minutes sedimentation then filter.
- 2) Direct filtration = 1 minute rapid mix, 20 minutes slow mix then filter.
- 3) In line filtration = 1 minute rapid mix then filter.
- 4) Three replicates analyzed for average filtrate turbidity.

Table 9. Effectiveness provided by flocculant addition when filtering through 3 inches of #4 sieve granulated activated carbon and 3 inches of #54 sieve garnet sand (#4 GAC + 54 garnet) as filter media.

	Turbidity of Blank	Average F	Average Filtrate Turbidity (NTUs) ⁴			Reduction Due to Flocculant Addition		
Filtration Method	(NTUs)	1.5 mL/L FeCl ₃	2.5 mL/L FeCl ₃	5 mL/L FeCl ₃	1.25 mL/L FeCl ₃	2.5 mL/L FeCl ₃	5 mL/L FeCl ₃	
Conventional Filtration ¹	64.3	10.4	35.4	10.4	83.8%	44.9%	83.8%	
Direct Filtration ²	138.7	75.7	54.3	11.7	45.4%	60.9%	91.6%	
In Line Filtration ³	196.0	166.3	139.7	166.1	15.2%	28.7%	15.3%	

- 1) Conventional filtration = 1 minute rapid mix, 20 minutes slow mix, 20 minutes sedimentation then filter.
- 2) Direct filtration = 1 minute rapid mix, 20 minutes slow mix then filter.
- 3) In line filtration = 1 minute rapid mix then filter.
- 4) Three replicates analyzed for average filtrate turbidity.

Table 10. Effectiveness provided by flocculant addition when filtering through nonwoven needle punched geotextile filter.

	Turbidity of Blank as	Average F	Average Filtrate Turbidity (NTUs) ⁴			Reduction Due to Flocculant Addition		
Filtration Method	Filtered (NTUs)	1.5 mL/L FeCl ₃	2.5 mL/L FeCl ₃	5 mL/L FeCl ₃	1.25 mL/L FeCl ₃	2.5 mL/L FeCl ₃	5 mL/L FeCl ₃	
Conventional Filtration ¹	100	24.0	61.3	29.3	76.1%	38.7%	70.0%	
Direct Filtration ²	178	117.1	88.8	113.2	34.1%	50.0%	36.3%	
In Line Filtration ³	261	314.7	223.2	225.5	-20.6%	14.4%	13.5%	

- 1) Conventional filtration = 1 minute rapid mix, 20 minutes slow mix, 20 minutes sedimentation then filter.
- 2) Direct filtration = 1 minute rapid mix, 20 minutes slow mix then filter.
- 3) In line filtration = 1 minute rapid mix then filter.
- 4) Six replicates analyzed for average filtrate turbidity.

Table 11. Effectiveness provided by flocculant addition when filtering through woven geotextile filter.

	Turbidity of Blank as	Average F	Average Filtrate Turbidity (NTUs) ⁴			Reduction Due to Flocculant Addition		
Filtration Method	Filtered (NTUs)	1.5 mL/L FeCl ₃	2.5 mL/L FeCl ₃	5 mL/L FeCl ₃	1.25 mL/L FeCl ₃	2.5 mL/L FeCl ₃	5 mL/L FeCl ₃	
Conventional Filtration ¹	82	9.3	19.3	7.6	88.7%	76.6%	90.8%	
Direct Filtration ²	121	101.5	62.6	10.3	15.8%	48.0%	91.4%	
In Line Filtration ³	228	185.7	137.9	154.2	18.5%	39.4%	32.3%	

- 1) Conventional filtration = 1 minute rapid mix, 20 minutes slow mix, 20 minutes sedimentation then filter.
- 2) Direct filtration = 1 minute rapid mix, 20 minutes slow mix then filter.
- 3) In line filtration = 1 minute rapid mix then filter.
- 4) Six replicates analyzed for average filtrate turbidity.

Table 12. Effectiveness provided by flocculant addition when filtering through paper filter.								
	Turbidity of Blank as	Average F	iltrate Turbidit	y (NTUs) ⁴	Reduction 1	Reduction Due to Flocculant Addition		
Filtration Method	Filtered (NTUs)	1.5 mL/L FeCl ₃	2.5 mL/L FeCl ₃	5 mL/L FeCl ₃	1.25 mL/L FeCl ₃	2.5 mL/L FeCl ₃	5 mL/L FeCl ₃	
Conventional Filtration ¹	49	4.1	12.9	4.5	91.7%	74.0%	90.8%	
Direct Filtration ²	40	45.3	31.5	4.1	-13.1%	21.4%	89.7%	
In Line Filtration ³	48	22.6	12.7	13.8	52.5%	73.3%	71.0%	

- 5) Conventional filtration = 1 minute rapid mix, 20 minutes slow mix, 20 minutes sedimentation then filter.
- 6) Direct filtration = 1 minute rapid mix, 20 minutes slow mix then filter.
- 7) In line filtration = 1 minute rapid mix then filter.
- 8) Six replicates analyzed for average filtrate turbidity.

## 4.3: Application of Results to Field Operations

The main results of the lab studies presented here, augmented by the literature recommendations, may be summarized as follows:

- Ferric chloride at a dose rate of 0.2 mL/L, equal to 66 gallons of ferric chloride per acre foot of water, is an effective flocculant under many mixing conditions for Minnesota sediments.
- Rapid mixing of ferric chloride can be done in as little as 6 seconds (Table 13), but requires intentional effort for best results, not simply pouring on a pond surface.

Table 13. Sizing of the flash mix vessel to have 6 seconds of hydraulic retention time.					
Flow rate (gpm)	Size of flash mix vessel (gal)				
40	4				
100	10				
200	20				
400	40				
1000	100				

• Mixing to create flocs (slow mix) can be done well with flow through a corrugated pipe, through a small mixing zone with injected air or a turbine blade mixer, or perhaps

through a rough surface such as between rip rap stones. In general, a longer slow mix flow path is better, keeping a minimum flow velocity of 1 foot/second.

• Sedimentation following flocculation and slow mix should be 60 minutes in duration before discharge (Table 14).

Table 14. Sizing of the sedimentation basin to provide 60 minutes of hydraulic retention time.						
Flow rate (gpm)  Size of sedimentation basin (gal)  Surface area of sedimentation basin if 4 feet deep (sf)						
40	2400	80				
100	6000	200				
200	12,000	400				
400	24,000	800				
1000	60,000	2000				

- Pre-flocculation (untreated) sedimentation should be done for minimization of sediment and the associated ferric flocculant dose rate.
- Deeper sedimentation basins will likely improve turbidity removal; and,
- Filtration can be effective to polish and improve flocculant-treated waters after sedimentation (Table 15).

Table 15. Sizing of a granular filter to provide 2 gpm/sf flow, assuming filter can be cleaned of sediment.						
Flow rate (gpm)	Filter area (sf)	Potential length of 3-foot high berm to achieve filter area (ft)				
40	20	4				
100	50	10				
200	100	20				
400	200	40				
1000	500	100				

As a last concern, use of ferric chloride for flocculant treatment of turbid water has the side effect of adding chloride (Cl⁻) to surface water upon discharge of the treated water. The question is what is the potential concentration of the Cl⁻ and how does this concentration compare to ecological standards.

Checking the molar concentrations for the flocculant dose of 0.2 mL/L of ferric chloride, it is estimated that the resulting chloride (only) concentration would be 133 mg Cl⁻ per liter of treated water. This level is well below the Total Maximum Daily Load (TMDL) concentration of 230 mg Cl⁻/L typically set for chloride-impaired waters.

The 230 mg CI⁻/L is also the maximum level allowable for the chronic 96-hour/4-day ecological exposure. The maximum level allowable for the acute level is frequently set as 860 mg CI⁻/L. Acute exposure limits may be more appropriate in discussions of releasing water in a batch as a one-time event such after pond treatment. Therefore, the 133 mg Cl-/L concentration associated with flocculation at a ferric chloride dose of 0.2 mL/L would be well below the acute level, even in undiluted (only treated) waters.

## **Chapter 5: Best Management Practices**

This chapter presents methods of design and implementation for best management practices (BMPs) for the reduction, control and capture of erosion products related to sediments and contact waters potentially released during concrete construction or demolition. This assessment assumes full compliance with and adherence to the guidance of the Minnesota Stormwater Manual (2005) and requirements of MPCA General Permit MN R 100001, Authorization to Discharge Stormwater Associated with Construction Activity Under the National Pollution Discharge Elimination System. This assessment of BMPs for concrete sediments and contact waters primarily addresses what changes and/or adjustments may be required to adapt existing soil sediment BMPs.

## **5.1: Best Management Practices Overview**

Sediments have significant potential to cause habitat loss, change waterway hydraulics, asphyxiate aquatic and benthic creatures, degrade navigation and plug drainage pipes and culverts. Construction sites are of particular concern due to the typical amount of disturbed ground, the stockpiles of earthen or particulate materials, the disturbance caused by construction equipment and operations, and the exposure to precipitation, sun and wind. Before flocculant treatment is considered, general BMPs for sediment control should be implemented to achieve broad reductions in sediment occurrence.

Preventing sediments from leaving a construction site requires a strategy built upon multiple lines of sediment control, if cost- and labor-efficiency is important. Such an approach provides flexibility for adjustment around both changing site operations and shifting seasonal weather, and can be strengthened through proactive maintenance. From the guidance provided in the Minnesota Stormwater Manual (2005), the following general classification of BMPs are suggested for construction sites:

- Diversion to limit run-on water;
- Reduction of erosional forces by surface water velocity reduction;
- Reduction of sediment development through sediment collection or anchoring;
- Sedimentation of mobilized sediments;
- Filtration of sediment-carrying flows;
- Collection of captured or contained sediments;
- Treatment of pH (hydronium and hydroxide);
- General housekeeping, including collection of trash and prevention of hazardous waste releases;
- Maintenance of erosion and sediment control devices/installations;
- Regular inspections; and,
- Recordkeeping.

Beyond guidance, erosion and sediment control are required by Minnesota regulation implanted through the requirements of MPCA General Permit MN R 100001, Authorization to Discharge Stormwater Associated with Construction Activity Under the National Pollution Discharge Elimination System.

## **5.2:** Best Management Practices for Flocculation Treatment

The treatment of construction site waters by flocculation must be approached not as a typical earthwork construction process, likely focused on geometry and materials, but as a treatment process focused on effectiveness, flexibility and resiliency. Geometry and materials must follow, not lead. Guiding objectives may be summarized as in Table 16.

Overall, flocculant treatment of construction site waters occurs with nine phases:

- 1. Characterization of Design Objectives;
- 2. Process Design;
- 3. Treatment Works and Site Design;
- 4. Installation/Construction;
- 5. Operation;
- 6. Evaluation;
- 7. Adjustment;
- 8. Maintenance; and,
- 9. Deconstruction and Salvage.

These nine phases are shown in Figure 14. For best management, each of these nine phases should be considered separate and distinct, with individual sequence direction and review. Skipping a phase evaluation could result in a lack of problem recognition, and strongly contribute to a systematization of problems and ultimate treatment failure, particularly to be true due to the interaction of hydraulic, chemical and physical factors that provides the foundation of flocculant treatment.

Table 16. Guiding objectiv	es of flocculation treatment of	construction site waters.
Guiding Objective	Effect(s) If Not Observed	Factors Influencing Success
Select appropriate treatment chemical and dose	Ineffective sedimentation	<ul> <li>Particle mineralogy</li> <li>Particle size</li> <li>Particle surface charge</li> <li>pH</li> <li>Treatment chemical reactivity</li> <li>Jar test evaluation effectiveness</li> </ul>
Manage hydraulics	System flood; potential incomplete or over dosing; potential ineffective settlement	<ul><li>Water source quantity and rate</li><li>Headworks design</li><li>Treatment unit design</li></ul>
Foster optimal conditions of mixing	Poor floc formation; potential overdosing	<ul><li>Rapid mix unit volume</li><li>Rapid mix method</li><li>Rapid mix energy</li></ul>
Preserve contrast of very high energy for mixing and very low energy for sedimentation	Poor floc formation; ineffective sedimentation; potential overdosing	<ul> <li>Rapid and slow mix unit volumes</li> <li>Rapid and slow mix methods</li> <li>Rapid and slow mix energies</li> </ul>
Consider operational details	System difficulty and inflexibility; inability to adjust to potential changes	<ul> <li>Receiving water requirements</li> <li>Access</li> <li>Power delivery</li> <li>Measurement and control strategy and implementation</li> <li>Sediment sludge removal</li> <li>Contingencies for unexpected conditions</li> </ul>

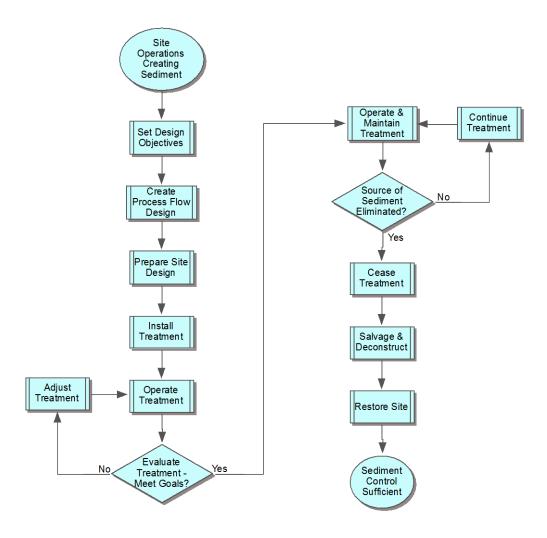


Figure 14. Phases, relative order and evaluation cycles of flocculant treatment of construction site waters.

Process steps for the characterization of design objectives are shown in Figure 15. Flocculant selection, optimization and effectiveness as compared to traditional sediment control approaches are all detailed with consideration of flocculant chemistry and dosing for site-specific sediments.

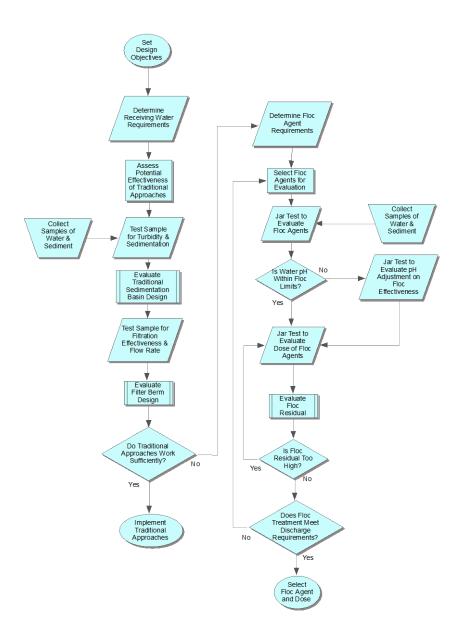


Figure 15. Process steps for the characterization of design objectives.

Treatment process flow design is detailed in Figure 16. Individual treatment processes including hydraulics, mixing, sludge trapping, sludge removal, and monitoring are all described by design steps. Evaluation cycles relate to whether design objectives are met.

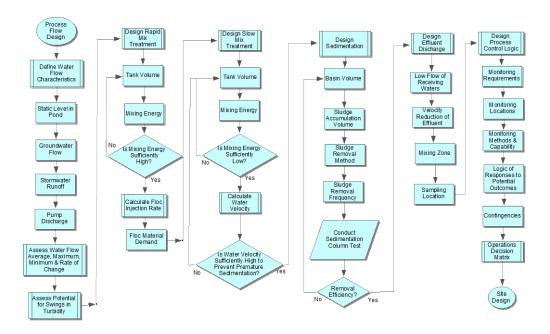


Figure 16. Treatment process flow design.

Treatment works and site design steps are detailed in Figure 17. These operations are the more traditional layout geometry and utility design processes that follow treatment process design.

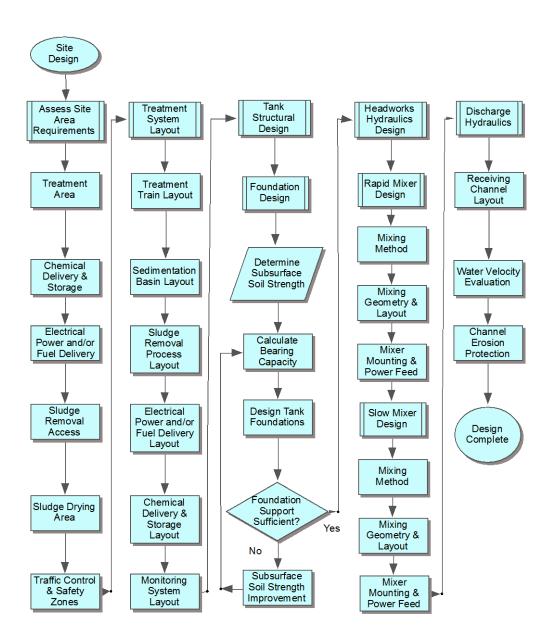


Figure 17. Treatment works and site design steps.

## **Chapter 6: Conclusions**

The chemical treatment practice known as flocculation can be effective for purifying construction site runoff of sediment, suspended solids and colloidal particles. Flocculation is generally used in combination with traditional sedimentation and filtration methods related to stormwater pollution prevention and dewatering fluid treatment.

Soil samples from across Minnesota, representing the wide range of geologic and geomorphological conditions, were used to identify differences in flocculant applicability and effectiveness. Soils were observed to have specific turbidity characteristics, as shown by the results from the blank (no flocculent) treatments. As commonly observed in the geotechnical and sedimentation disciplines, soil sediment turbidity is a function of several soil characteristics: fines content, particle size and distribution, particle surface charge and mineral composition. Some soils were found to have insufficient fines to cause turbidity beyond what can be treated by gravity sedimentation alone. In this study, soil sediment turbidity ranged from a high of 1000 NTUs to a low of 35.9 NTUs, averaging 322 NTUs with a standard deviation of 265 NTUs (relative standard deviation of 82%).

Flocculent chemicals were found to have highly specific effectiveness across the range of soil types. Specific dose by turbidity relationships exist for each flocculent chemical and each soil, particularly for polymer compounds that can either work very well or make turbidity worse due to specific surface charge interrelationships. When effective, dose levels can vary between 0.2 to 100 ppm. This does range is quite large, given the potential for wide cost differences between flocculent chemicals (anecdotally noted as ranging from \$2 to \$50 per gallon).

Flocculent chemicals found to be broadly effective in this study include: ferric chloride, Aqua Hawk 6447 and Tramfloc 865A. Ferric chloride was found to treat turbidity consistently to below 100 NTUs for all soils considered in this study, and typically to below 50 NTUs for most soils, at a dose of 0.2 mL/L (approximately 200 ppm, or 70 gallons of ferric chloride per acre foot of water to be treated).

Levels of pH were not adversely impacted by the dose amounts used in either Phase I or II, as values remained between 7 and 8.5 for most final pH measurements. These values were consistent with the blank treatments for the soils, and suggest sufficient alkalinity exists in waters made up from the soils considered in this study to fully buffer any pH effects from flocculent treatment at the dose levels considered.

Pre-flocculation (untreated) sedimentation should be done for minimization of sediment and the associated ferric flocculant dose rate.

Rapid mixing of ferric chloride can be done in as little as 6 seconds but requires intentional effort for best results, not simply pouring on a pond surface. Mixing to create flocs (slow mix) can be done well with flow through a corrugated pipe, through a small mixing zone with injected air or a turbine blade mixer, or perhaps through a rough surface such as between rip rap stones. In general, a longer slow mix flow path is better, keeping a minimum flow velocity of 1 foot/second.

Sedimentation following flocculation and slow mix should be 60 minutes in duration before discharge. Deeper sedimentation basins will likely improve turbidity removal. Filtration can be effective to polish and improve flocculant-treated waters after sedimentation, and may involve sedimentation bank infiltration or in-channel check dam features to work as filters.

Best management practices (BMPs) were also developed for evaluation and design of flocculant treatment methods and technologies and the associated sizing for flows representative of field operations.

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# Appendix A – Soil Sample Characterization

Soil Characterization Data Sheet MnDOT Floc Enviromental Engineering Lab-TN 382 11/14/2012 Revised 03/25/2013

Site & Strata			Hydro	Hydrometer Reading	eading			Нф	Orga	Organic Burn	0,	Sample Date	Sample Location
Ada Subsoil	22	19	17	16	16	14	13	8.3	38.1986 3	37.6312 1.5	1.5%	7/25/12	TH 9: RP 192 50'E
Ada Topsoil	31	56	22	21	18	14.5	13	7.8	42.2853 39.7849		2.9%	7/25/12	TH 9: RP 192 50'E
<b>Brandon Subsoil</b>	32.5	30.5	29	27	26.5	24	20	7.4	43.5556 4	41.6424 4.4	4.4%	7/21/12	l 94 SP#8824-100
Brandon Topsoil	98	33.5	30.5	53	27.5	25	20.5	7.2	35.192 3.	32.0479 8.9	8.9%	7/21/12	l 94 SP#8824-100
Carlton Subsoil	51.0	45.0	37.5	32.5	27.5	19.5	15.0	8.3	50.2041 4	49.5052 1.4	1.4%	10/10/12	TH 210 / Jay Cooke State Park
Carlton Topsoil	21.0	18.0	15.5	14.5	13.0	12.5	10.5	8.8	57.9638 5	57.3842 1.0	1.0%	10/10/12	TH 210 / Jay Cooke State Park
Cook Subsoil	15	13	12	11.5	10.5	10	8.5	7.6	40.0828 33	39.3778 1.8	1.8%	not stated	not stated
Cook Topsoil	11	10	6	8.5	8	8	7	9.9	39.2792 3	38.607 1.7	1.7%	not stated	not stated
Coon Rapids Subsoil	11.5	11	10	9.2	9.2	6	8	7.9	45.115 4	44.8176 0.7	0.7%	7/27/12	CR 14 & Flintwood Ave.
Coon Rapids Topsoil	10.5	10	9.2	6	6	8.5	8	7.3	42.367 4	41.0656 3.1	3.1%	7/27/12	CR 14 & Flintwood Ave.
East Grand Forks Subsoil	47	39	28.5	56	24.5	21.5	10.5	8.1	38.1747 30	36.9997 3.1	3.1%	7/25/12	TH 220: RP 23 W Ditch
East Grand Forks Topsoil	41.5	33.5	27	25	23.5	21.5	18.5	7.9	38.2869 3	34.8095 9.1	9.1%	7/25/12	TH 220: RP 23 W Ditch
Foley Subsoil	21	18	16.5	15.5	15	12	10.5	7.1	48.6811 48	48.1813 1.0	1.0%	7/9/12	TH 23 SP#0503-75
Foley Topsoil	22	20.5	17.5	16.5	15.5	12.5	11.5	7.3	46.8383 4.	45.6372 2.6	7.6%	7/9/12	TH 23 SP#0503-75
<b>Grand Rapids Subsoil</b>	11	10.5	9.2	8.5	8	8	7.5	8.2	47.117	46.64 1.0	1.0%	7/5/12	TH 169 SP#3115-51
<b>Grand Rapids Topsoil</b>	15	12.5	11	11	9.5	8	7.5	7.7	43.2758 4	43.0439 0.5	0.5%	7/5/12	TH 169 SP#3115-51
Hampton Subsoil	32	27	22	20	18	16.5	15	7.9	41.6513 40	40.7795 2.1	2.1%	9/21/12	TH 52 NE Pond
Hampton Topsoil	32.5	27.5	23.5	21.5	20	18	15	7.2	38.8719 3.	35.4016 8.9	8.9%	9/21/12	TH 52 NE Pond
Houston Riverbed	22	17	15	14	13	12	10	7.6	45.1316 4	44.3344 1.8	1.8%	9/27/12	CSAH 22 & Hop Hollow Rd
Houston Topsoil	22.5	18.5	15	13.5	12.5	11.5	9.5	7.4	44.6971 4.	42.1139 5.8	2.8%	9/27/12	CSAH 22 & Hop Hollow Rd
Kandiyohi Subsoil	28.5	25.5	20.5	19.5	18	15.5	13	8.1	38.1984 3	37.7686 1.1	1.1%	8/1/12	CSAH 5 Sta 70+50 CL
Kandiyohi Topsoil	38	34.5	29	27	25.5	22.5	18.5	7.7	46.6952 43	42.6838 8.6	8.6%	8/1/12	CSAH 5 Sta 74+00 Lt
Lakeville Subsoil	31.5	25.5	21.5	20	18.5	16	14	7.6	44.1547 4	43.2905 2.0	2.0%	6/6/12	Holyoke Rd & 194th St W
Lakeville Topsoil	41	31	25.5	23	22	18.5	16	8			4.6%	6/6/12	Holyoke Rd & 194th St W
Lindstrom Subsoil	18	16.5	15	14.5	13.5	12.5	11	8	45.586 4	44.9655 1.4	1.4%	7/13/12	TH 8 N Lindstrom Lake
Lindstrom Topsoil	22	19	16	15	13.5	11.5	10	7.9	43.7376 4		3.5%	7/13/12	TH 8 N Lindstrom Lake
Mankato Topsoil	18	14	12	11.5	11	10.5	9.5	7.79	41.0658 40		1.8%	11/20/12	TH 90 & Blue Earth R
Mapleton Topsoil	37	34.5	31	29	27.5	23	18	7.5	41.9586 40	40.2246 4.1	4.1%	11/4/12	Klein Pond N of TH 30
Moorhead Subsoil	48	40	34	31	28.5	26.5	22.5	8.3	43.4978 4.	42.1036 3.2	3.2%	7/21/12	TH 75 & 50th Ave SP#1406-69
Moorhead Topsoil	39	33.5	29	27	25.5	23.5	19.5	7.8	43.5906 40	40.8702 6.2	6.2%	7/21/12	TH 75 & 50th Ave SP#1406-69
New Sweden Topsoil	18	16.5	15	14.5	14	13	11.5	7.97	44.5727 4	43.8493 1.6	1.6%	11/4/12	TH 22 at CD1
North Mankato Subsoil	31.5	29.5	27	25.5	23	21	17	7.6	39.8362	39.1343 1.8	1.8%	8/1/12	TH 14 Sta 518+50 Ravine
North Mankato Topsoil	35	32	28	26.5	24.5	22	18.5	7.1	45.1482 4.	42.4729 5.9	2.9%	8/1/12	TH 14 Sta 518+50 Ravine
Olivia A Subsoil	24	21.5	17.5	16.5	14.5	13	11.5	8.2	46.0579 4.	45.7711 0.6	%9.0	8/21/12	CR 62 at 1/4 mi W of US 71

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Site & Strata								Нф				Sample Date	Sample Location
			Hydro	Hydrometer Reading	eading				J	Organic Burn	u		
Olivia A Topsoil	36.5	31	25.5	22.5	20	16.5	13	7.4	44.4135	40.368	9.1%	8/21/12	CR 62 at 1/4 mi W of US 71
Olivia B Clay	32	28.5	25.5	23.5	22	20	16	7.8	44.5888	43.7853	1.8%	8/21/12	CSAH 4 btwn CSAH 1 & US 71
Owatonna Clay	30	28	25.5	24	22.5	20.5	17	7.3	42.9901	41.8209	2.7%	8/21/12	TH 14 Sta 1569+48 SP#7401-34
Owatonna Topsoil	30	27.5	24	22.5	21	19	16	7.2	42.9447	40.8306	4.9%	8/21/12	TH 14 Sta 1569+48 SP#7401-34
Perham Subsoil	6	8.5	8.5	8.5	8	8	7	7	44.8753	44.5782	0.7%	7/21/12	TH 10 & CR 34 SP#5607-42
Perham Topsoil	12	11.5	10.5	10	9.5	6	8	7.5	44.8498	43.736	2.5%	7/21/12	TH 10 & CR 34 SP#5607-42
Pipestone Subsoil	23	21	18.5	17	16	15	13	8.1	40.6169	39.8541	1.9%	not stated	CSAH 10 at 1.5 mi E of TH 75
Pipestone Topsoil	22	22.5	19	17	16	14	12	7.9	39.1515	36.4854	%8.9	not stated	CSAH 10 at 1.5 mi E of TH 75
Ramsey Peat	20.5	17.5	15	13.5	12	10.5	6	9.9	45.8799	36.8351	19.7%	not stated	35 E & Maryland Ave
Rockford Cornfield	24.5	23	20	18.5	17.5	16.5	14	7.5	34.8561	33.306	4.4%	not stated	TH 55
Rockford Subsoil	90	27.5	25	23.5	22	19.5	16.5	6.9	33.5297	32.7203	2.4%	not stated	TH 55
Rockford Topsoil	32.5	28.5	24.5	22	21	17	13.5	7.4	36.8538	35.5014	3.7%	not stated	TH 55
Sandstone Subsoil	43.0	41.0	38.0	36.5	34.0	31.0	24.5	8.0	53.2787	51.9657	2.5%	10/10/12	North of TH 18, West of CH 61
Sandstone Topsoil	21.0	17.5	15.5	14.0	13.0	11.5	10.5	7.9	25.5695	53.2740	4.1%	10/10/12	North of CR 32, East of CH 22
Stillwater Sediment	44.5	34	53	24	19.5	14	10	7.5	35.5434	32.1189	%9.6	7/27/12	Bridge Pier River Boring
Tamarack Clay	20'2	49	47	44	40.5	33	23.5	7.1	38.2459	37.0922	3.0%	9/17/12	CSAH 32 Sta 137+36 40'Lt
Tamarack Peat	67	28	26.5	25	18	13	14	6.9	7.7385	0.7205	%2'06	9/17/12	CSAH 32 Sta 152+85 30'Lt
Tamarack Topsoil	22	19	17	15.5	14.5	11.5	6	6.9	37.4814	35.828	4.4%	9/17/12	CSAH 32 Sta 120+70 34'Lt
Wabasha Subsoil (Box)	27.5	20.5	16	15	13.5	12.5	11	7.8	40.738	39.5298	3.0%	9/21/12	CR 4 @ Handshaw Coulee (?)
Wabasha Subsoil (Rock)	17	16	15.5	15.5	15	14.5	14	8	48.1498	47.2421	1.9%	9/21/12	CR 4 @ Handshaw Coulee (?)
Wabasha Topsoil	34	28	22	20	18.5	16.5	14	7.5	43.8146	41.9462	4.3%	9/21/12	CR 4 @ Handshaw Coulee (?)
Worthington Subsoil	28	25.5	22.5	21	20	18	16.5	7.6	46.2904	45.2711	2.2%	7/16/12	TH 60 SP#5305-58
Worthington Topsoil	56	23	20	18.5	17.5	15	13	7.4	45.0201	45.0201 42.4371	5.7%	7/16/12	TH 60 SP#5305-58

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xtural Presumptive Unified Classification System Designation	SM	ım SM	lay Loam SC-SM	.oam SC-SM	ML	SM	SM	SM	SM	Loam	Loam CL-ML-OL	n OL	SM	Loam SM	SM	SM	Loam	ım SM	SM	ım SM	SM	n OL	' Loam SM	Loam ML-OL	SM	Loam SM	SM	lay Loam SC-SM	Loam CL-ML-OL	n OL	SM	SC-SM	.oam SC-SM	CM
MnDOT Triangular Textural Classification	Sandy Loam	Organic Sandy Loam	Slightly Organic Sandy Clay Loam	Organic Sandy Clay Loam	Loam	Sandy Loam	Sandy Loam	Sandy Loam	Sandy Loam	Slightly Organic Sandy Loam	Slightly Organic Clay Loam	Organic Clay Loam	Sandy Loam	Slightly Organic Sandy Loam	Sandy Loam	Sandy Loam	Slightly Organic Sandy Loam	Organic Sandy Loam	Sandy Loam	Organic Sandy Loam	Sandy Loam	Organic Clay Loam	Slightly Organic Sandy Loam	Slightly Organic Clay Loam	Sandy Loam	Slightly Organic Sandy Loam	Sandy Loam	Slightly Organic Sandy Clay Loam	Slightly Organic Clay Loam	Organic Clay Loam	Sandy Loam	Sandy Clay Loam	Organic Sandy Clay Loam	meo I ybue?
Natural Vegetation	Upland Prairie	Upland Prairie	Oak Woodland	Oak Woodland	<b>Great Lakes Pine Forest</b>	<b>Great Lakes Pine Forest</b>	<b>Boreal Hardwood-Conifer</b>	<b>Boreal Hardwood-Conifer</b>	Oak Woodland	Oak Woodland	Floodplain Forest	Floodplain Forest	Maple Basswood Forest	Maple Basswood Forest	Boreal Hardwood-Conifer	Boreal Hardwood-Conifer	Upland Prairie	<b>Upland Prairie</b>	Oak Woodland	Oak Woodland	Upland Prairie	<b>Upland Prairie</b>	Oak Woodland	Oak Woodland	Maple Basswood Forest	Maple Basswood Forest	Maple Basswood Forest	Prairie Wetland	Floodplain Forest	Floodplain Forest	<b>Upland Prairie</b>	Prairie Wetland	Prairie Wetland	eirierd brieful I
Ecological Province	Prairie Parkland	Prairie Parkland	Prairie Parkland	Prairie Parkland	Laurentian Mixed Forest	Laurentian Mixed Forest	Laurentian Mixed Forest	Laurentian Mixed Forest	<b>Eastern Broadleaf Forest</b>	Eastern Broadleaf Forest	Prairie Parkland	Prairie Parkland	Laurentian Mixed Forest	Laurentian Mixed Forest	Laurentian Mixed Forest	Laurentian Mixed Forest	Eastern Broadleaf Forest	Eastern Broadleaf Forest	Eastern Broadleaf Forest	Eastern Broadleaf Forest	Prairie Parkland	Prairie Parkland	Eastern Broadleaf Forest	Eastern Broadleaf Forest	Laurentian Mixed Forest	Laurentian Mixed Forest	Prairie Parkland	Prairie Parkland	Prairie Parkland	Prairie Parkland	Prairie Parkland	Prairie Parkland	Prairie Parkland	Prairie Parkland
Geomorphology	Des Moines Lobe	Des Moines Lobe	Wadena Lobe	Wadena Lobe	Superior Lobe	Superior Lobe	Superior Lobe	Superior Lobe	Des Moines Lobe	Des Moines Lobe	Des Moines Lobe	Des Moines Lobe	Superior Lobe	Superior Lobe	Des Moines Lobe	Des Moines Lobe	Illinoisian Glaciation	Illinoisian Glaciation	Weathered Residual	Weathered Residual	Wadena Lobe	Wadena Lobe	Kansan Glaciation	Kansan Glaciation	Des Moines Lobe	Des Moines Lobe	Des Moines Lobe	Des Moines Lobe	Des Moines Lobe	Des Moines Lobe	Des Moines Lobe	Des Moines Lobe	Des Moines Lobe	Des Moines Lohe
Strata	Glacial Lake Sediment	Glacial Lake Sediment	Stagnation Moraine	Stagnation Moraine	Glacial Lake Sediment	<b>Glacial Lake Sediment</b>	Ground Moraine	<b>Ground Moraine</b>	Outwash	Outwash	Glacial Lake Sediment	Glacial Lake Sediment	<b>Ground Moraine</b>	<b>Ground Moraine</b>	Outwash	Outwash	Red Drift	Red Drift	Colluvuum	Colluvuum	Stagnation Moraine	Stagnation Moraine	Gray Drift	Gray Drift	<b>Ground Moraine</b>	<b>Ground Moraine</b>	Glacial Lake/Alluvium	<b>Glacial Lake Sediment</b>	Glacial Lake Sediment	<b>Glacial Lake Sediment</b>	Stagnation Moraine	<b>Ground Moraine</b>	<b>Ground Moraine</b>	Ground Moraine
Site & Strata	Ada Subsoil	Ada Topsoil	Brandon Subsoil	Brandon Topsoil	Carlton Subsoil	Carlton Topsoil	Cook Subsoil	Cook Topsoil	Coon Rapids Subsoil	Coon Rapids Topsoil	<b>East Grand Forks Subsoil</b>	East Grand Forks Topsoil	Foley Subsoil	Foley Topsoil	<b>Grand Rapids Subsoil</b>	<b>Grand Rapids Topsoil</b>	Hampton Subsoil	Hampton Topsoil	Houston Riverbed	Houston Topsoil	Kandiyohi Subsoil	Kandiyohi Topsoil	Lakeville Subsoil	Lakeville Topsoil	Lindstrom Subsoil	Lindstrom Topsoil	Mankato Topsoil	Mapleton Topsoil	Moorhead Subsoil	Moorhead Topsoil	New Sweden Topsoil	North Mankato Subsoil	North Mankato Topsoil	Olivia A Subsoil

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Site & Strata	Strata	Geomorphology	Ecological Province	Natural Vegetation	MnDOT Triangular Textural Classification	Presumptive Unified Classification System Designation
Olivia A Topsoil	<b>Ground Moraine</b>	Des Moines Lobe	Prairie Parkland	Upland Prairie	Organic Sandy Loam	SM
Olivia B Clay	<b>Ground Moraine</b>	Des Moines Lobe	Prairie Parkland	Prairie Wetland	Sandy Clay Loam	SC-SM
Owatonna Clay	<b>Ground Moraine</b>	Des Moines Lobe	Eastern Broadleaf Forest	Oak Woodland	Slightly Organic Sandy Clay Loam	SC-SM
Owatonna Topsoil	<b>Ground Moraine</b>	Des Moines Lobe	Eastern Broadleaf Forest	Oak Woodland	Slightly Organic Sandy Clay Loam	SC-SM
Perham Subsoil	Outwash	Des Moines Lobe	Eastern Broadleaf Forest	Oak Woodland	Sandy Loam	SM
Perham Topsoil	Outwash	Des Moines Lobe	Eastern Broadleaf Forest	Oak Woodland	Slightly Organic Sandy Loam	SM
Pipestone Subsoil	Gray Drift	Kansan Glaciation	Prairie Parkland	Upland Prairie	Sandy Loam	SM
Pipestone Topsoil	Gray Drift	Kansan Glaciation	Prairie Parkland	Upland Prairie	Organic Sandy Loam	SM
Ramsey Peat	End Moraine	Des Moines Lobe	Eastern Broadleaf Forest	Oak Woodland	Highly Organic Sandy Loam	SM
Rockford Cornfield	End Moraine	Des Moines Lobe	Eastern Broadleaf Forest	Maple Basswood Forest	Slightly Organic Sandy Loam	SM
Rockford Subsoil	End Moraine	Des Moines Lobe	Eastern Broadleaf Forest	Maple Basswood Forest	Slightly Organic Sandy Loam	SM
Rockford Topsoil	End Moraine	Des Moines Lobe	Eastern Broadleaf Forest	Maple Basswood Forest	Slightly Organic Sandy Loam	SM
Sandstone Subsoil	<b>Ground Moraine</b>	Superior Lobe	Laurentian Mixed Forest	Laurentian Mixed Forest Northern Hardwood Forest	Slightly Organic Clay Loam	CL-ML-OL
Sandstone Topsoil	<b>Ground Moraine</b>	Superior Lobe	Laurentian Mixed Forest	Laurentian Mixed Forest Northern Hardwood Forest	Slightly Organic Sandy Loam	SM
Stillwater Sediment	End Moraine	Superior Lobe	Eastern Broadleaf Forest	Oak Woodland	Organic Loam	OF
Tamarack Clay	End Moraine	Des Moines Lobe	Laurentian Mixed Forest	Peatland	Slightly Organic Clay	CI-OI
Tamarack Peat	Peat	Des Moines Lobe	Laurentian Mixed Forest	Peatland	Peat	PT
Tamarack Topsoil	End Moraine	Des Moines Lobe	Laurentian Mixed Forest	Peatland	Slightly Organic Sandy Loam	SM
Wabasha Subsoil (Box)	Colluvuum	Weathered Residual	Weathered Residual Eastern Broadleaf Forest	Oak Woodland	Slightly Organic Sandy Loam	SM
Wabasha Subsoil (Rock)	Colluvuum	Weathered Residual	Eastern Broadleaf Forest	Oak Woodland	Sandy Loam	SM
Wabasha Topsoil	Colluvuum	Weathered Residual	Eastern Broadleaf Forest	Oak Woodland	Slightly Organic Sandy Loam	SM
Worthington Subsoil	<b>Ground Moraine</b>	Des Moines Lobe	Prairie Parkland	Upland Prairie	Slightly Organic Sandy Loam	SM
Worthington Topsoil	<b>Ground Moraine</b>	Des Moines Lobe	Prairie Parkland	Upland Prairie	Organic Sandy Loam	SM

8.3

Material	Ada Subsoil
Sample Date	July 25, 2012
Sample Loc	TH 9: RP 192 50'E

Strata: Glacial Lake Sediment Geomorphology: Des Moines Lobe Ecological Province: Prairie Parkland Natural Vegetation: Upland Prairie

pH (ASTM D4972) =

Hydrometer (ASTM D	422)	100.0 g
Estimated Gs =	2.65	
Gs Corr, a =	0.99	ASTM D-422 Table 1
Lab Temp =	21	deg C
K factor =	0.01328	ASTM D-422 Table 3
Effective L (cm) =	(10.5 cm - 8.2	2 cm*R / 50 g/L)

Organic Content (ASTM D2974)

Dry Total Mass (g) = 38.1986

Mineral Mass (g) = 37.6312

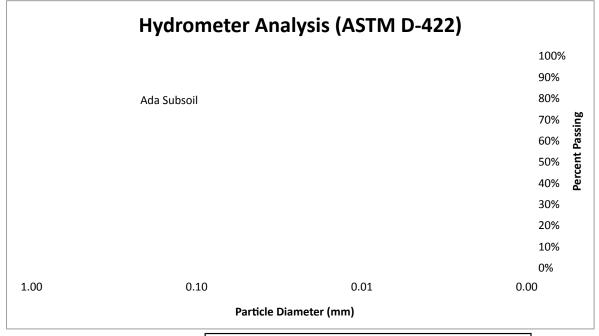
Organic Content = 1.5%

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	22	12.69	0.03344737	21.8%
5	19	13.18	0.02156025	18.8%
15	17	13.51	0.01260176	16.8%
30	16	13.67	0.00896473	15.8%
60	16	13.67	0.00633902	15.8%
250	14	14.00	0.00314251	13.9%
1440	13	14.16	0.00131702	12.9%





Soil Classification: Sandy Loam SM

7.8

Material Ada Topsoil
Sample Date
July 25, 2012
Sample Loc TH 9: RP 192 50'E

Strata: Glacial Lake Sediment Geomorphology: Des Moines Lobe Ecological Province: Prairie Parkland Natural Vegetation: Upland Prairie

pH (ASTM D4972) =

Hydrometer (ASTM D	422)	100.0 g
Estimated Gs =	2.65	
Gs Corr, a =	0.99	ASTM D-422 Table 1
Lab Temp =	21	deg C
K factor =	0.01328	ASTM D-422 Table 3
Effective L (cm) =	(10.5 cm - 8.2	2 cm*R / 50 g/L)

Organic Content (ASTM D2974)

Dry Total Mass (g) = 42.2853

Mineral Mass (g) = 39.7849

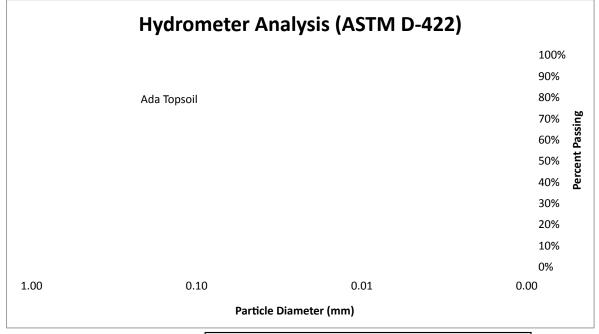
Organic Content = 5.9%

 $+ 0.5*(14.0 \text{ cm} - 67.0 \text{ cm}^3/27.8 \text{ cm}^2)$ 

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	31	11.21	0.03144159	30.7%
5	26	12.03	0.02059981	25.7%
15	22	12.69	0.01221325	21.8%
30	21	12.85	0.00869171	20.8%
60	18	13.34	0.00626251	17.8%
250	14.5	13.92	0.00313329	14.4%
1440	13	14.16	0.00131702	12.9%





Soil Classification: Organic Sandy Loam SM

Material	Brandon Subsoil
Sample Date	July 21, 2012
Sample Loc	I 94 SP#8824-100

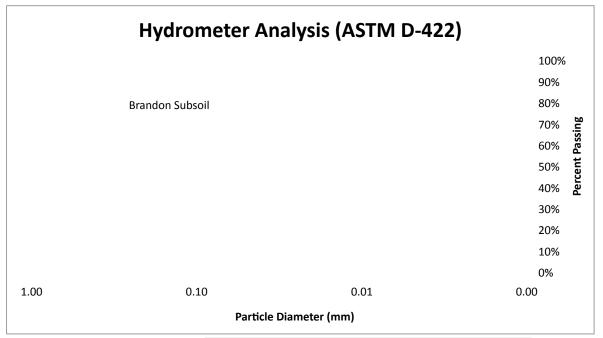
Strata: Stagnation Moraine Geomorphology: Wadena Lobe Ecological Province: Prairie Parkland Natural Vegetation: Oak Woodland

Hydrometer (ASTM D	422)	100.0	g	pH (ASTM D4972) =	7.4
Estimated Gs =	2.65				
Gs Corr, a =	0.99	ASTM D-422	Table 1	Organic Content (ASTM D	)2974)
Lab Temp =	21	deg C		Dry Total Mass (g) =	43.5556
K factor =	0.01328	ASTM D-422	Table 3	Mineral Mass (g) =	41.6424
Effective L (cm) =	(10.5 cm - 8.	2 cm*R / 50 g,	/L)	Organic Content =	4.4%
	+ 0.5*(14.0	cm - 67.0 cm ³ ,	$/27.8 \text{ cm}^2)$	_	

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	32.5	10.96	0.03109472	32.2%
5	30.5	11.29	0.019958	30.2%
15	29	11.54	0.01164758	28.7%
30	27	11.87	0.00835232	26.7%
60	26.5	11.95	0.00592635	26.2%
250	24	12.36	0.0029527	23.8%
1440	20	13.01	0.00126252	19.8%





Soil Classification: Slightly Organic Sandy Clay Loam SC-SM

Material	Brandon Topsoil
Sample Date	July 21, 2012
Sample Loc	I 94 SP#8824-100

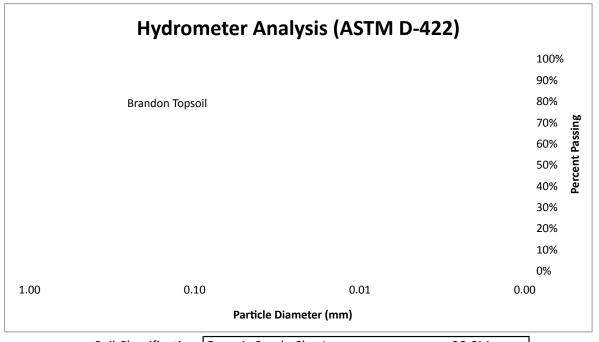
Strata: Stagnation Moraine Geomorphology: Wadena Lobe Ecological Province: Prairie Parkland Natural Vegetation: Oak Woodland

Hydrometer (ASTM D422)		100.0	g	pH (ASTM D4972) =[	7.2
Estimated Gs =	2.65				
Gs Corr, a =	0.99	ASTM D-422	Table 1	Organic Content (ASTM [	2974)
Lab Temp =	21	deg C		Dry Total Mass (g) =	35.192
K factor =	0.01328	ASTM D-422	Table 3	Mineral Mass (g) =	32.0479
Effective L (cm) =	2 cm*R / 50 g	/L)	Organic Content =	8.9%	
$+ 0.5*(14.0 \text{ cm} - 67.0 \text{ cm}^3/27.8 \text{ cm}^2)$					

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	36	10.39	0.0302699	35.6%
5	33.5	10.80	0.01951841	33.2%
15	30.5	11.29	0.01152276	30.2%
30	29	11.54	0.00823609	28.7%
60	27.5	11.78	0.00588554	27.2%
250	25	12.19	0.00293304	24.8%
1440	20.5	12.93	0.00125854	20.3%





Soil Classification: Organic Sandy Clay Loam SC-SM

8.3

Carlton Subsoil Material Sample Date October 10, 2012 Sample Loc TH 210 / Jay Cooke State Park

Strata: Glacial Lake Sediment

Geomorphology: Superior Lobe

Ecological Province: Laurentian Mixed Forest Natural Vegetation: Great Lakes Pine Forest

pH (ASTM D4972) =

Hydrometer (ASTM D422) 100.0 g Estimated Gs = 2.65 ASTM D-422 Table 1 Gs Corr. a = 0.99 Lab Temp = 21 deg C K factor = 0.01328 ASTM D-422 Table 3 Effective L (cm) = (10.5 cm - 8.2 cm*R / 50 g/L)

Organic Content (ASTM D2974) Dry Total Mass (g) = 50.2041

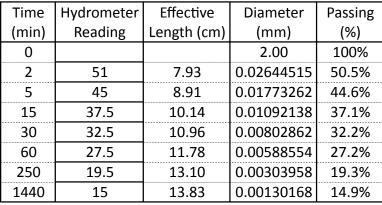
> Mineral Mass (g) = 49.5052

Organic Content =

1.4%

 $+ 0.5*(14.0 \text{ cm} - 67.0 \text{ cm}^3/27.8 \text{ cm}^2)$ 

Photomicroscopy (field of view 9 x 12 mm)





## **Hydrometer Analysis (ASTM D-422)** 100% 90% Carlton Subsoil 80% 70% Percent Passing 60% 50% 40% 30% 20% 10% 0% 1.00 0.10 0.01 0.00 Particle Diameter (mm)

Soil Classification: Loam ML

Material	Carlton Topsoil	
Sample Date	October 10, 2012	
Sample Loc	TH 210 / Jay Cooke State P	ark

Strata: Glacial Lake Sediment

Geomorphology: Superior Lobe

Ecological Province: Laurentian Mixed Forest Natural Vegetation: Great Lakes Pine Forest

Hydrometer (ASTM D	100.0	g	
Estimated Gs =	2.65		-
Gs Corr, a =	0.99	ASTM D-422	Table 1
Lab Temp =	21	deg C	
K factor =	0.01328	ASTM D-422	Table 3
Effective L (cm) = $(10.5 \text{ cm} - 8.2 \text{ cm*R} / 50 \text{ g/L})$			

pH (ASTM D4972) = 8.8

Organic Content (ASTM D2974)

Dry Total Mass (g) =	8.8
Mineral Mass (g) =	8.8
Organic Content =	0.0%

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	21	12.85	0.03366285	20.8%
5	18	13.34	0.02169398	17.8%
15	15.5	13.75	0.012716	15.3%
30	14.5	13.92	0.00904502	14.4%
60	13	14.16	0.00645208	12.9%
250	12.5	14.24	0.00317	12.4%
1440	10.5	14.57	0.00133595	10.4%



**Hydrometer Analysis (ASTM D-422)** 100% 90% 80% **Carlton Topsoil Percent Passing** 70% 60% 50% 40% 30% 20% 10% 0% 0.01 1.00 0.10 0.00 Particle Diameter (mm)

Soil Classification: Sandy Loam SM

Material	Cook Subsoil	
Sample Date	not stated	
Sample Loc	not stated	

Strata: Ground Moraine Geomorphology: Superior Lobe

Ecological Province: Laurentian Mixed Forest Natural Vegetation: Boreal Hardwood-Conifer

Hydrometer (ASTM D	100.0	g	
Estimated Gs =	2.65		-
Gs Corr, a =	0.99	ASTM D-422	Table 1
Lab Temp =	21	deg C	
K factor =	0.01328	ASTM D-422	Table 3
Effective L (cm) = $(10.5 \text{ cm} - 8.2 \text{ cm*R} / 50 \text{ g/L})$			

pH (ASTM D4972) = 7.6

Organic Content (ASTM D2974)

Dry Total Mass (g) = 40.0828 Mineral Mass (g) = 39.3778 Organic Content = 1.8%

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	15	13.83	0.03492787	14.9%
5	13	14.16	0.02235065	12.9%
15	12	14.33	0.01297865	11.9%
30	11.5	14.41	0.00920352	11.4%
60	10.5	14.57	0.0065448	10.4%
250	10	14.65	0.00321529	9.9%
1440	8.5	14.90	0.0013509	8.4%



**Hydrometer Analysis (ASTM D-422)** 100% 90% 80% Cook Subsoil 70% 60% 50% 40% 30% 20% 10% 0% 1.00 0.10 0.01 0.00 Particle Diameter (mm)

Soil Classification: Loam ML

6.6

Material	Cook Topsoil	
Sample Date	not stated	
Sample Loc	not stated	

Strata: Ground Moraine Geomorphology: Superior Lobe

Ecological Province: Laurentian Mixed Forest Natural Vegetation: Boreal Hardwood-Conifer

pH (ASTM D4972) =

Hydrometer (ASTM D422)		100.0	g
Estimated Gs =	2.65		
Gs Corr, a =	0.99	ASTM D-422	Table 1
Lab Temp =	21	deg C	
K factor =	0.01328	ASTM D-422	Table 3
Effective L (cm) = $(10.5 \text{ cm} - 8.2 \text{ cm*R} / 50 \text{ g/L})$			

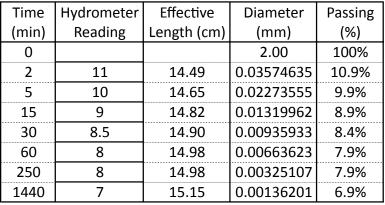
Organic Content (ASTM D2974)

Dry Total Mass (g) = 39.2792 Mineral Mass (g) = 38.607

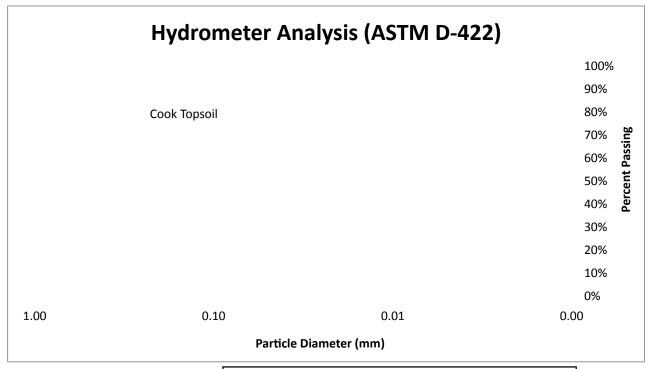
Organic Content = 1.7%

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

Photomicroscopy (field of view 9 x 12 mm)







Soil Classification: Sandy Loam SM

Coon Rapids Subsoil Material Sample Date July 27, 2012 Sample Loc | CR 14 & Flintwood Ave.

Strata: Outwash Geomorphology: Des Moines Lobe **Ecological Province: Eastern Broadleaf Forest** 

Natural Vegetation: Oak Woodland

Hydrometer (ASTM D422) 100.0 Estimated Gs = 2.65 0.99 Gs Corr, a = ASTM D-422 Table 1 Lab Temp = 21 deg C K factor = 0.01328 ASTM D-422 Table 3

Effective L (cm) = (10.5 cm - 8.2 cm*R / 50 g/L)

 $+ 0.5*(14.0 \text{ cm} - 67.0 \text{ cm}^3/27.8 \text{ cm}^2)$ 

pH (ASTM D4972) =	7.9
p (	

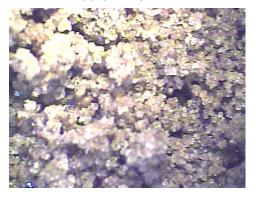
Organic Content (ASTM D2974)

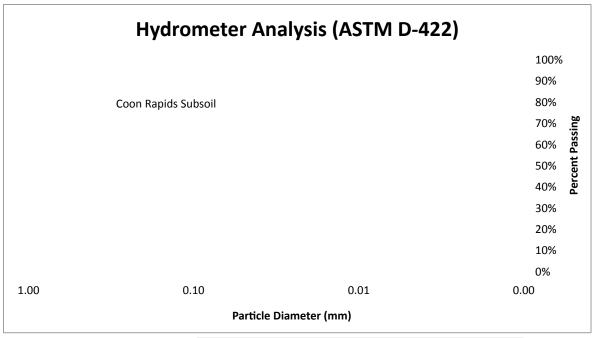
Dry Total Mass (g) = Mineral Mass (g) = 44.8176 0.7%

Organic Content =

Photomicroscopy (field of view 9 x 12 mm)

Time	Lydromotor	Effective	Diameter	Daccing
	Hydrometer		Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	11.5	14.41	0.03564507	11.4%
5	11	14.49	0.02260798	10.9%
15	10	14.65	0.01312638	9.9%
30	9.5	14.74	0.00930768	9.4%
60	9.5	14.74	0.00658152	9.4%
250	9	14.82	0.00323323	8.9%
1440	8	14.98	0.00135461	7.9%





Soil Classification: Sandy Loam SM Material Coon Rapids Topsoil
Sample Date July 27, 2012
Sample Loc CR 14 & Flintwood Ave.

Strata: Outwash Geomorphology: Des Moines Lobe Ecological Province: Eastern Broadleaf Forest

Natural Vegetation: Oak Woodland

Hydrometer (ASTM D422) 100.0 g

Estimated Gs = 2.65

Gs Corr, a = 0.99 ASTM D-422 Table 1

Lab Temp = 21 deg C

K factor = 0.01328 ASTM D-422 Table 3

Effective L (cm) =  $\overline{(10.5 \text{ cm} - 8.2 \text{ cm*R} / 50 \text{ g/L})}$ +  $0.5*(14.0 \text{ cm} - 67.0 \text{ cm}^3/27.8 \text{ cm}^2)$  pH (ASTM D4972) = 7.3

Organic Content (ASTM D2974)

Dry Total Mass (g) = 42.3

Ory Total Mass (g) = 42.367

Mineral Mass (g) = 41.0656

Organic Content = 3.1%

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	10.5	14.57	0.03584735	10.4%
5	10	14.65	0.02273555	9.9%
15	9.5	14.74	0.01316305	9.4%
30	9	14.82	0.00933354	8.9%
60	9	14.82	0.00659981	8.9%
250	8.5	14.90	0.00324217	8.4%
1440	8	14.98	0.00135461	7.9%



**Hydrometer Analysis (ASTM D-422)** 100% 90% 80% Coon Rapids Topsoil 70% 60% 50% 40% 30% 20% 10% 0% 1.00 0.10 0.01 0.00 Particle Diameter (mm)

Soil Classification: Slightly Organic Sandy Loam SM

Material East Grand Forks Subsoil
Sample Date July 25, 2012
Sample Loc TH 220: RP 23 W Ditch

Strata: Glacial Lake Sediment Geomorphology: Des Moines Lobe Ecological Province: Prairie Parkland Natural Vegetation: Floodplain Forest

Hydrometer (ASTM D422)		100.0 g	
Estimated Gs =	2.65	-	
Gs Corr, a =	0.99	ASTM D-422 Table 1	
Lab Temp =	21	deg C	
K factor =	0.01328	ASTM D-422 Table 3	
Effective L (cm) = $\frac{10.5 \text{ cm} - 8.2 \text{ cm*R}}{10.5 \text{ cm} - 8.2 \text{ cm*R}}$			

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

pH (ASTM D4972) =	8.1

Organic Content (ASTM D2974)

Dry Total Mass (g) = 38.1747

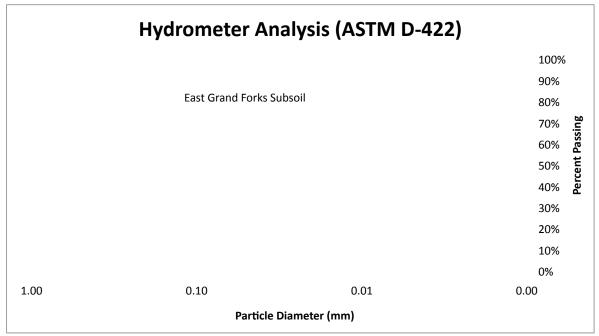
Mineral Mass (g) = 36.9997

Organic Content = 3.1%

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	47	8.59	0.02751712	46.5%
5	39	9.90	0.01868564	38.6%
15	28.5	11.62	0.0116889	28.2%
30	26	12.03	0.00840984	25.7%
60	24.5	12.28	0.00600714	24.3%
250	21.5	12.77	0.00300128	21.3%
1440	10.5	14.57	0.00133595	10.4%





Soil Classification: Slightly Organic Clay Loam CL-ML-OL

East Grand Forks Topsoil Material Sample Date July 25, 2012 Sample Loc TH 220: RP 23 W Ditch

Strata: Glacial Lake Sediment Geomorphology: Des Moines Lobe Ecological Province: Prairie Parkland Natural Vegetation: Floodplain Forest

Hydrometer (ASTM D422)		100.0 g
Estimated Gs =	2.65	-
Gs Corr, a =	0.99	ASTM D-422 Table 1
Lab Temp =	21	deg C
K factor =	0.01328	ASTM D-422 Table 3
Effective L (cm) =	(10.5 cm - 8.3	2 cm*R / 50 g/L)

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

Organic Content (ASTM D2974)

Dry Total Mass (g) = 38.2869 Mineral Mass (g) = 34.8095 9.1%

Organic Content =

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	41.5	9.49	0.02892627	41.1%
5	33.5	10.80	0.01951841	33.2%
15	27	11.87	0.01181197	26.7%
30	25	12.19	0.00846696	24.8%
60	23.5	12.44	0.00604713	23.3%
250	21.5	12.77	0.00300128	21.3%
1440	18.5	13.26	0.0012744	18.3%



#### **Hydrometer Analysis (ASTM D-422)** 100% 90% 80% East Grand Forks Topsoil 70% 60% 50% 40% 30% 20% 10% 0% 1.00 0.10 0.01 0.00 Particle Diameter (mm)

Soil Classification: Organic Clay Loam OL

Material Foley Subsoil
Sample Date July 9, 2012
Sample Loc TH 23 SP#0503-75

Strata: Ground Moraine Geomorphology: Superior Lobe

Ecological Province: Laurentian Mixed Forest Natural Vegetation: Maple Basswood Forest

pH (ASTM D4972) =

Hydrometer (ASTM D	422)	100.0 g
Estimated Gs =	2.65	-
Gs Corr, a =	0.99	ASTM D-422 Table 1
Lab Temp =	21	deg C
K factor =	0.01328	ASTM D-422 Table 3
Fffective L (cm) =	(10.5 cm - 8.3	2 cm*R / 50 g/L)

Organic Content (ASTM D2974)

Dry Total Mass (g) = 48.6811

Mineral Mass (g) = 48.1813

Organic Content = 1.0%

- (10.3 cm - 8.2 cm K / 30 g/L)

 $+ 0.5*(14.0 \text{ cm} - 67.0 \text{ cm}^3/27.8 \text{ cm}^2)$ 

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	21	12.85	0.03366285	20.8%
5	18	13.34	0.02169398	17.8%
15	16.5	13.59	0.01263996	16.3%
30	15.5	13.75	0.00899157	15.3%
60	15	13.83	0.00637693	14.9%
250	12	14.33	0.00317911	11.9%
1440	10.5	14.57	0.00133595	10.4%



#### **Hydrometer Analysis (ASTM D-422)** 100% 90% 80% Foley Subsoil 70% Percent Passing 60% 50% 40% 30% 20% 10% 0% 1.00 0.10 0.01 0.00 Particle Diameter (mm)

Soil Classification: Sandy Loam SM

Material Foley Topsoil
Sample Date
Sample Loc TH 23 SP#0503-75

Strata: Ground Moraine Geomorphology: Superior Lobe

Ecological Province: Laurentian Mixed Forest Natural Vegetation: Maple Basswood Forest

pH (ASTM D4972) =

Hydrometer (ASTM D	422)	100.0 g
Estimated Gs =	2.65	_
Gs Corr, a =	0.99	ASTM D-422 Table 1
Lab Temp =	21	deg C
K factor =	0.01328	ASTM D-422 Table 3
Effective L (cm) =	(10.5 cm - 8.2	2 cm*R / 50 g/L)

Organic Content (ASTM D2974)

Dry Total Mass (g) = 46.8383

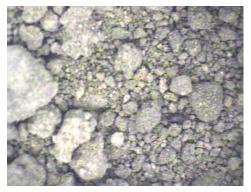
Mineral Mass (g) = 45.6372

Organic Content = 2.6%

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	22	12.69	0.03344737	21.8%
5	20.5	12.93	0.02135808	20.3%
15	17.5	13.42	0.01256345	17.3%
30	16.5	13.59	0.0089378	16.3%
60	15.5	13.75	0.006358	15.3%
250	12.5	14.24	0.00317	12.4%
1440	11.5	14.41	0.00132841	11.4%



**Hydrometer Analysis (ASTM D-422)** 100% 90% 80% **Foley Topsoil** 70% Percent Passing 60% 50% 40% 30% 20% 10% 0% 1.00 0.10 0.01 0.00 Particle Diameter (mm)

Material Grand Rapids Subsoil
Sample Date July 5, 2012
Sample Loc TH 169 SP#3115-51

Strata: Outwash Geomorphology: Des Moines Lobe Ecological Province: Laurentian Mixed Forest Natural Vegetation: Boreal Hardwood-Conifer

pH (ASTM D4972) =

Hydrometer (ASTM D422)		100.0	g
Estimated Gs =	2.65		
Gs Corr, a =	0.99	ASTM D-422	Table 1
Lab Temp =	21	deg C	
K factor =	0.01328	ASTM D-422	Table 3
Effective L (cm) = $\frac{10.5 \text{ cm} - 8.2 \text{ cm*R} / 50 \text{ g/L}}{10.5 \text{ cm} - 8.2 \text{ cm*R}}$			

Organic Content (ASTM D2974)

Dry Total Mass (g) = 47.117

Mineral Mass (g) = 46.64

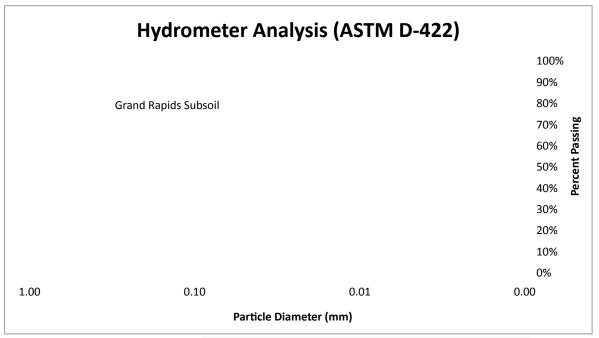
Organic Content = 1.0%

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	11	14.49	0.03574635	10.9%
5	10.5	14.57	0.02267185	10.4%
15	9.5	14.74	0.01316305	9.4%
30	8.5	14.90	0.00935933	8.4%
60	8	14.98	0.00663623	7.9%
250	8	14.98	0.00325107	7.9%
1440	7.5	15.06	0.00135832	7.4%





Soil Classification: Sandy Loam SM

Material	Grand Rapids Topsoil
Sample Date	July 5, 2012
Sample Loc	TH 169 SP#3115-51

Strata: Outwash Geomorphology: Des Moines Lobe Ecological Province: Laurentian Mixed Forest Natural Vegetation: Boreal Hardwood-Conifer

pH (ASTM D4972) =

Hydrometer (ASTM D422)		100.0	g
Estimated Gs =	2.65		
Gs Corr, a =	0.99	ASTM D-422	Table 1
Lab Temp =	21	deg C	
K factor =	0.01328	ASTM D-422	Table 3
Effective L (cm) = $(10.5 \text{ cm} - 8.2 \text{ cm*R} / 50 \text{ g/L})$			
	cm - 67.0 cm ³	$/27.8 \text{ cm}^2$ )	

Organic Content (ASTM D2974)

Dry Total Mass (g) = 43.2758

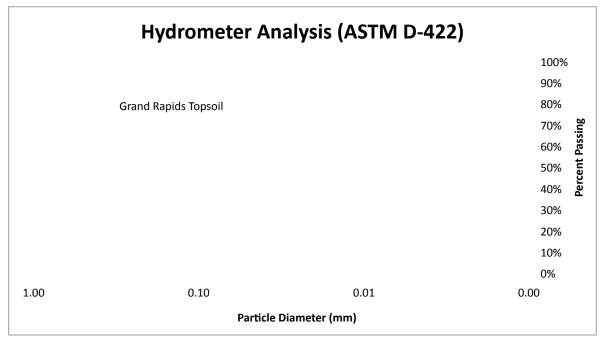
Mineral Mass (g) = 43.0439

Organic Content = 0.5%

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	15	13.83	0.03492787	14.9%
5	12.5	14.24	0.02241526	12.4%
15	11	14.49	0.01305272	10.9%
30	11	14.49	0.00922967	10.9%
60	9.5	14.74	0.00658152	9.4%
250	8	14.98	0.00325107	7.9%
1440	7.5	15.06	0.00135832	7.4%





Soil Classification: Sandy Loam SM

Material	Hampton Subsoil
Sample Date	September 21, 2012
Sample Loc	TH 52 NE Pond

Strata: Red Drift

Geomorphology: Illinoisian Glaciation Ecological Province: Eastern Broadleaf Forest

Natural Vegetation: Upland Prairie

Hydrometer (ASTM D422)		100.0 g
Estimated Gs =	2.65	_
Gs Corr, a =	0.99	ASTM D-422 Table 1
Lab Temp =	21	deg C
K factor =	0.01328	ASTM D-422 Table 3
Fffective L (cm) =	(10.5 cm - 8.3	2 cm*R / 50 g/L)

pH (ASTM D4972) = 7.9

Organic Content (ASTM D2974)

Dry Total Mass (g) = 41.6513

Mineral Mass (g) = 40.7795 Organic Content = 2.1%

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	32	11.05	0.03121077	31.7%
5	27	11.87	0.02045893	26.7%
15	22	12.69	0.01221325	21.8%
30	20	13.01	0.008747	19.8%
60	18	13.34	0.00626251	17.8%
250	16.5	13.59	0.00309614	16.3%
1440	15	13.83	0.00130168	14.9%



**Hydrometer Analysis (ASTM D-422)** 100% 90% 80% Hampton Subsoil 70% 60% 50% 40% 30% 20% 10% 0% 1.00 0.10 0.01 0.00 Particle Diameter (mm)

	Hampton Topsoil
Sample Date	September 21, 2012
Sample Loc	TH 52 NE Pond

Strata: Red Drift

Geomorphology: Illinoisian Glaciation Ecological Province: Eastern Broadleaf Forest

Natural Vegetation: Upland Prairie

Hydrometer (ASTM D	422)	100.0 g
Estimated Gs =	2.65	_
Gs Corr, a =	0.99	ASTM D-422 Table 1
Lab Temp =	21	deg C
K factor =	0.01328	ASTM D-422 Table 3
Effective L (cm) =	(10.5 cm - 8.2	2 cm*R / 50 g/L)

pH (ASTM D4972) = 7.2

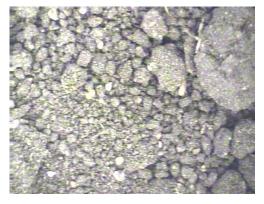
Organic Content (ASTM D2974)

Dry Total Mass (g) = 38.8719 Mineral Mass (g) = 35.4016 Organic Content = 8.9%

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	32.5	10.96	0.03109472	32.2%
5	27.5	11.78	0.02038812	27.2%
15	23.5	12.44	0.01209426	23.3%
30	21.5	12.77	0.00866394	21.3%
60	20	13.01	0.00618506	19.8%
250	18	13.34	0.00306799	17.8%
1440	15	13.83	0.00130168	14.9%



## **Hydrometer Analysis (ASTM D-422)** 100% 90% 80% **Hampton Topsoil** 70% 60% 50% 40% 30% 20% 10% 0% 1.00 0.10 0.01 0.00 Particle Diameter (mm)

Material	Houston Riverbed
Sample Date	September 27, 2012
Sample Loc	CSAH 22 & Hop Hollow Rd

Strata: Colluvuum

Geomorphology: Weathered Residual **Ecological Province: Eastern Broadleaf Forest** 

Natural Vegetation: Oak Woodland

Hydrometer (ASTM D	422)	100.0 g
Estimated Gs =	2.65	
Gs Corr, a =	0.99	ASTM D-422 Table 1
Lab Temp =	21	deg C
K factor =	0.01328	ASTM D-422 Table 3
Fffective L (cm) =	(10.5 cm - 8.3	2 cm*R / 50 g/L)

Organic Content (ASTM D2974)

pH (ASTM D4972) =

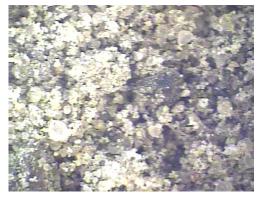
Dry Total Mass (g) = 45.1316 Mineral Mass (g) = 44.3344

Organic Content = 1.8%

 $+ 0.5*(14.0 \text{ cm} - 67.0 \text{ cm}^3/27.8 \text{ cm}^2)$ 

Photomicroscopy (field of view 9 x 12 mm)

	1	l		
Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	22	12.69	0.03344737	21.8%
5	17	13.51	0.02182689	16.8%
15	15	13.83	0.01275385	14.9%
30	14	14.00	0.00907163	13.9%
60	13	14.16	0.00645208	12.9%
250	12	14.33	0.00317911	11.9%
1440	10	14.65	0.00133971	9.9%



**Hydrometer Analysis (ASTM D-422)** 100% 90% 80% **Houston Riverbed** 70% 60% 50% 40% 30% 20% 10% 0% 1.00 0.10 0.01 0.00 Particle Diameter (mm)

Soil Classification: Sandy Loam SM

Material	Houston Topsoil
Sample Date	September 27, 2012
	CSAH 22 & Hop Hollow Rd

Strata: Colluvuum

Geomorphology: Weathered Residual Ecological Province: Eastern Broadleaf Forest

Natural Vegetation: Oak Woodland

Hydrometer (ASTM D422)		100.0 g	
Estimated Gs =	2.65	_	
Gs Corr, a =	0.99	ASTM D-422 Table 1	
Lab Temp =	21	deg C	
K factor =	0.01328	ASTM D-422 Table 3	
Fffective L (cm) =	(10.5 cm - 8.3	2 cm*R / 50 g/L)	

pH (ASTM D4972) = 7.4

Organic Content (ASTM D2974)

Dry Total Mass (g) = 44.6971 Mineral Mass (g) = 42.1139 Organic Content = 5.8%

 $+0.5*(14.0 \text{ cm} - 67.0 \text{ cm}^3/27.8 \text{ cm}^2)$ 

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	22.5	12.60	0.0333391	22.3%
5	18.5	13.26	0.02162722	18.3%
15	15	13.83	0.01275385	14.9%
30	13.5	14.08	0.00909816	13.4%
60	12.5	14.24	0.00647073	12.4%
250	11.5	14.41	0.00318819	11.4%
1440	9.5	14.74	0.00134345	9.4%



**Hydrometer Analysis (ASTM D-422)** 100% 90% 80% **Houston Topsoil** 70% 60% 50% 40% 30% 20% 10% 0% 1.00 0.10 0.01 0.00 Particle Diameter (mm)

Material	Kandiyohi Subsoil
Sample Date	August 1, 2012
Sample Loc	CSAH 5 Sta 70+50 CL

Strata: Stagnation Moraine Geomorphology: Wadena Lobe Ecological Province: Prairie Parkland Natural Vegetation: Upland Prairie

pH (ASTM D4972) =

Hydrometer (ASTM D422)		100.0 g
Estimated Gs =	2.65	
Gs Corr, a =	0.99	ASTM D-422 Table 1
Lab Temp =	21	deg C
K factor =	0.01328	ASTM D-422 Table 3
Effective L (cm) = $\frac{10.5 \text{ cm} - 8.2 \text{ cm}^*\text{R} / 50 \text{ g/L}}{10.5 \text{ cm} - 8.2 \text{ cm}^*\text{R} / 50 \text{ g/L}}$		

Organic Content (ASTM D2974)

Dry Total Mass (g) = 38.1984

Mineral Mass (g) = 37.7686

Organic Content = 1.1%

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	28.5	11.62	0.03201136	28.2%
5	25.5	12.11	0.0206699	25.2%
15	20.5	12.93	0.01233109	20.3%
30	19.5	13.10	0.00877451	19.3%
60	18	13.34	0.00626251	17.8%
250	15.5	13.75	0.00311477	15.3%
1440	13	14.16	0.00131702	12.9%



## **Hydrometer Analysis (ASTM D-422)** 100% 90% 80% Kandiyohi Subsoil 70% 60% 50% 40% 30% 20% 10% 0% 1.00 0.10 0.01 0.00 Particle Diameter (mm)

Soil Classification: Sandy Loam OL

Material	Kandiyohi Topsoil
Sample Date	August 1, 2012
Sample Loc	CSAH 5 Sta 74+00 Lt

Strata: Stagnation Moraine Geomorphology: Wadena Lobe Ecological Province: Prairie Parkland Natural Vegetation: Upland Prairie

pH (ASTM D4972) =

Hydrometer (ASTM D422)		100.0 g
Estimated Gs =	2.65	
Gs Corr, a =	0.99	ASTM D-422 Table 1
Lab Temp =	21	deg C
K factor =	0.01328	ASTM D-422 Table 3
Effective L (cm) = $(10.5 \text{ cm} - 8.2 \text{ cm*R} / 50 \text{ g/L})$		

Organic Content (ASTM D2974)

Dry Total Mass (g) = 46.6952

Mineral Mass (g) = 42.6838

Organic Content = 8.6%

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	38	10.06	0.02978832	37.6%
5	34.5	10.64	0.01936966	34.2%
15	29	11.54	0.01164758	28.7%
30	27	11.87	0.00835232	26.7%
60	25.5	12.11	0.00596688	25.2%
250	22.5	12.60	0.00298194	22.3%
1440	18.5	13.26	0.0012744	18.3%



100%
90%
80%
<b>2</b> . 70%
70% 50% 50% 50%
50%
40%
30%
20%
10%
0%
0.00
_

Soil Classification: Organic Clay Loam OL

Material	Lakeville Subsoil
Sample Date	June 6, 2012
Sample Loc	Holyoke Rd & 194th St W

Strata: Gray Drift

Geomorphology: Kansan Glaciation

Ecological Province: Eastern Broadleaf Forest

Natural Vegetation: Oak Woodland

Hydrometer (ASTM D422)		100.0 g
Estimated Gs =	2.65	
Gs Corr, a =	0.99	ASTM D-422 Table 1
Lab Temp =	21	deg C
K factor =	0.01328	ASTM D-422 Table 3
Fffective L (cm) =	(10.5 cm - 8.3	2 cm*R / 50 g/L)

 $+ 0.5*(14.0 \text{ cm} - 67.0 \text{ cm}^3/27.8 \text{ cm}^2)$ 

pH (ASTM D4972) = 7.6

Organic Content (ASTM D2974)

Dry Total Mass (g) = 44.1547 Mineral Mass (g) = 43.2905 2.0%

Organic Content =

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	31.5	11.13	0.0313264	31.2%
5	25.5	12.11	0.0206699	25.2%
15	21.5	12.77	0.01225266	21.3%
30	20	13.01	0.008747	19.8%
60	18.5	13.26	0.00624324	18.3%
250	16	13.67	0.00310547	15.8%
1440	14	14.00	0.00130938	13.9%



	Hydrometer An	alysis (ASTM D-42	2)	
			100%	
			90%	
	Lakeville Subsoil		80%	
			70% <u>s</u>	
			Percent Passing 40%	
			50% <b>ent</b>	
			40%	
			30%	
			20%	
			10%	
			0%	
1.00	0.10	0.01	0.00	
	Particle Diameter (mm)			

8

Material	Lakeville Topsoil
Sample Date	June 6, 2012
Sample Loc	Holyoke Rd & 194th St W

Strata: Gray Drift Geomorphology: Kansan Glaciation

Ecological Province: Eastern Broadleaf Forest

pH (ASTM D4972) =

Natural Vegetation: Oak Woodland

Hydrometer (ASTM D422)		100.0 g
Estimated Gs =	2.65	
Gs Corr, a =	0.99	ASTM D-422 Table 1
Lab Temp =	21	deg C
K factor =	0.01328	ASTM D-422 Table 3
Effective L (cm) =	2 cm*R / 50 g/L)	

Organic Content (ASTM D2974)

Dry Total Mass (g) = 36.030

 Ory Total Mass (g) =
 36.0307

 Mineral Mass (g) =
 34.3677

 Organic Content =
 4.6%

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	41	9.57	0.02905099	40.6%
5	31	11.21	0.01988541	30.7%
15	25.5	12.11	0.01193377	25.2%
30	23	12.52	0.00858007	22.8%
60	22	12.69	0.00610663	21.8%
250	18.5	13.26	0.00305855	18.3%
1440	16	13.67	0.00129395	15.8%



	<b>Hydrometer An</b>	alysis (ASTM D-42	.2)
			100%
			90%
	Lakeville Topsoil		80%
			70%
			70% 5 60% 5 50% 5 40% 6
			50%
			40%
			30%
			20%
			10%
			0%
1.00	0.10	0.01	0.00
	Particle I	Diameter (mm)	

8

Material	Lindstrom Subsoil
Sample Date	July 13, 2012
Sample Loc	TH 8 N Lindstrom Lake

Strata: Ground Moraine Geomorphology: Des Moines Lobe

Ecological Province: Laurentian Mixed Forest Natural Vegetation: Maple Basswood Forest

pH (ASTM D4972) =

Hydrometer (ASTM D422)		100.0 g
Estimated Gs =	2.65	
Gs Corr, a =	0.99	ASTM D-422 Table 1
Lab Temp =	21	deg C
K factor =	0.01328	ASTM D-422 Table 3
Effective L (cm) = $(10.5 \text{ cm} - 8.6)$		2 cm*R / 50 g/L)

Organic Content (ASTM D2974)

Dry Total Mass (g) = 45.586

Mineral Mass (g) = 44.9655

Organic Content = 1.4%

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	18	13.34	0.03430119	17.8%
5	16.5	13.59	0.02189305	16.3%
15	15	13.83	0.01275385	14.9%
30	14.5	13.92	0.00904502	14.4%
60	13.5	14.08	0.00643337	13.4%
250	12.5	14.24	0.00317	12.4%
1440	11	14.49	0.00133219	10.9%



# **Hydrometer Analysis (ASTM D-422)** 100% 90% 80% Lindstrom Subsoil 70% 60% 50% 40% 30% 20% 10% 0% 1.00 0.10 0.01 0.00 Particle Diameter (mm)

Soil Classification: Sandy Loam SM

Material	Lindstrom Topsoil
Sample Date	July 13, 2012
Sample Loc	TH 8 N Lindstrom Lake

Strata: Ground Moraine Geomorphology: Des Moines Lobe

Ecological Province: Laurentian Mixed Forest Natural Vegetation: Maple Basswood Forest

Hydrometer (ASTM D422)		100.0 g
Estimated Gs =	2.65	_
Gs Corr, a =	0.99	ASTM D-422 Table 1
Lab Temp =	21	deg C
K factor =	0.01328	ASTM D-422 Table 3
Effective L (cm) =	2 cm*R / 50 g/L)	

pH (ASTM D4972) = 7.9

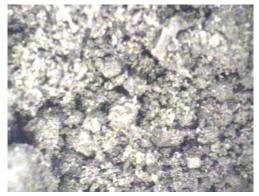
Organic Content (ASTM D2974)

Dry Total Mass (g) = 43.7376 Mineral Mass (g) = 42.214 Organic Content = 3.5%

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	22	12.69	0.03344737	21.8%
5	19	13.18	0.02156025	18.8%
15	16	13.67	0.01267804	15.8%
30	15	13.83	0.00901834	14.9%
60	13.5	14.08	0.00643337	13.4%
250	11.5	14.41	0.00318819	11.4%
1440	10	14.65	0.00133971	9.9%



Hydrometer Analysis (ASTM D-422)			
			100%
			90%
	Lindstrom Topsoil		80%
			70% 💆
			40% 905 60% 609 Percent Passing
			50%
			40% <b>B</b>
			30%
			20%
			10%
			0%
1.00	0.10	0.01	0.00
	Particle I	Diameter (mm)	

Material	Mankato Topsoil
Sample Date	November 20, 2012
Sample Loc	TH 90 & Blue Earth R

Strata: Glacial Lake/Alluvium Geomorphology: Des Moines Lobe Ecological Province: Prairie Parkland

Natural Vegetation: Maple Basswood Forest

pH (ASTM D4972) =

Hydrometer (ASTM D422)		100.0 g
Estimated Gs =	2.65	
Gs Corr, a =	0.99	ASTM D-422 Table 1
Lab Temp =	21	deg C
K factor =	0.01328	ASTM D-422 Table 3
Effective L (cm) = $\frac{10.5 \text{ cm} - 8.2 \text{ cm*R}}{10.5 \text{ cm} - 8.2 \text{ cm*R}} = \frac{10.5 \text{ cm}}{10.5 \text{ cm}} = \frac{10.5 \text{ cm}}{$		

Organic Content (ASTM D2974)

Dry Total Mass (g) = 41.0658

Mineral Mass (g) = 40.3376

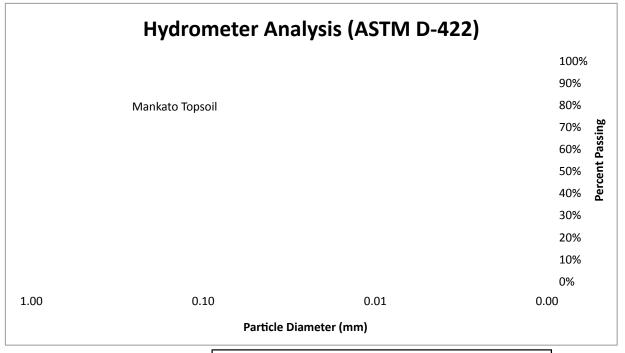
Organic Content = 1.8%

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	18	13.34	0.03430119	17.8%
5	14	14.00	0.02222087	13.9%
15	12	14.33	0.01297865	11.9%
30	11.5	14.41	0.00920352	11.4%
60	11	14.49	0.00652636	10.9%
250	10.5	14.57	0.00320628	10.4%
1440	9.5	14.74	0.00134345	9.4%





Soil Classification: Sandy Loam SM

Material	Mapleton Topsoil
Sample Date	November 4, 2012
Sample Loc	Klein Pond N of TH 30

Strata: Glacial Lake Sediment Geomorphology: Des Moines Lobe Ecological Province: Prairie Parkland Natural Vegetation: Prairie Wetland

pH (ASTM D4972) =

Hydrometer (ASTM D422)		100.0 g	
Estimated Gs =	2.65		
Gs Corr, a =	0.99	ASTM D-422 Table 1	
Lab Temp =	21	deg C	
K factor =	0.01328	ASTM D-422 Table 3	
Effective L (cm) = $(10.5 \text{ cm} - 8.2 \text{ cm*R} / 50 \text{ g/L})$			

Organic Content (ASTM D2974)

Dry Total Mass (g) = 41.9586

Mineral Mass (g) = 40.2246

Organic Content = 4.1%

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	37	10.23	0.03003008	36.6%
5	34.5	10.64	0.01936966	34.2%
15	31	11.21	0.01148085	30.7%
30	29	11.54	0.00823609	28.7%
60	27.5	11.78	0.00588554	27.2%
250	23	12.52	0.00297222	22.8%
1440	18	13.34	0.00127833	17.8%



Hydrometer Analysis (ASTM D-422)					
			100%		
			90%		
	Mapleton Topsoil		80%		
			70% <u>يَّا</u>		
			90% Percent Passing 40%		
			50%		
			40% <b>B</b>		
			30%		
			20%		
			10%		
			0%		
1.00	0.10	0.01	0.00		
	Particle	Diameter (mm)			

Material	Moorhead Subsoil	
Sample Date	July 21, 2012	
Sample Loc	TH 75 & 50th Ave SP#1406	-69

Strata: Glacial Lake Sediment Geomorphology: Des Moines Lobe Ecological Province: Prairie Parkland Natural Vegetation: Floodplain Forest

Hydrometer (ASTM D422)		100.0	g
Estimated Gs =	2.65		-
Gs Corr, a =	0.99	ASTM D-422	Table 1
Lab Temp =	21	deg C	
K factor =	0.01328	ASTM D-422	Table 3
Effective L (cm) =	(10.5 cm - 8.2	2 cm*R / 50 g	/L)

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

Organic Content (ASTM D2974)

Dry Total Mass (g) =	43.4978
Mineral Mass (g) =	42.1036
Organic Content =	3.2%

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	48	8.42	0.02725308	47.5%
5	40	9.73	0.01853021	39.6%
15	34	10.72	0.0112261	33.7%
30	31	11.21	0.00811818	30.7%
60	28.5	11.62	0.00584445	28.2%
250	26.5	11.95	0.00290331	26.2%
1440	22.5	12.60	0.00124247	22.3%



	Moorhead Subsoil		80% 70% 60% 50% 40% 30% 20%	Percent Passing
			10% 0%	_
1.00	0.10	0.01	0.00	
	Particle	Diameter (mm)		

Soil Classification: Slightly Organic Clay Loam CL-ML-OL

Material	Moorhead Topsoil	
Sample Date	July 21, 2012	
Sample Loc	TH 75 & 50th Ave SP#1406	-69

Strata: Glacial Lake Sediment Geomorphology: Des Moines Lobe Ecological Province: Prairie Parkland Natural Vegetation: Floodplain Forest

Hydrometer (ASTM D422)		100.0 g	
Estimated Gs =	2.65	_	
Gs Corr, a =	0.99	ASTM D-422 Table 1	L
Lab Temp =	21	deg C	
K factor =	0.01328	ASTM D-422 Table 3	3
Effective L (cm) = $\frac{10.5 \text{ cm} - 8.2 \text{ cm}^* \text{R} / 50 \text{ g/L}}{10.5 \text{ cm} - 8.2 \text{ cm}^* \text{R}}$			

(10.5 cm - 8.2 cm*R / 50 g/L) + 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

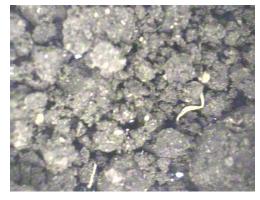
pH (ASTM D4972) =	7.8

Organic Content (ASTM D2974)

Dry Total Mass (g) =	43.5906
Mineral Mass (g) =	40.8702
Organic Content =	6.2%

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	39	9.90	0.02954459	38.6%
5	33.5	10.80	0.01951841	33.2%
15	29	11.54	0.01164758	28.7%
30	27	11.87	0.00835232	26.7%
60	25.5	12.11	0.00596688	25.2%
250	23.5	12.44	0.00296248	23.3%
1440	19.5	13.10	0.00126649	19.3%



**Hydrometer Analysis (ASTM D-422)** 100% 90% 80% Moorhead Topsoil 70% 60% 50% 40% 30% 20% 10% 0% 1.00 0.10 0.01 0.00 Particle Diameter (mm)

Soil Classification: Organic Clay Loam OL

Material	New Sweden Topsoil
Sample Date	November 4, 2012
Sample Loc	TH 22 at CD1

Strata: Stagnation Moraine Geomorphology: Des Moines Lobe **Ecological Province: Prairie Parkland** Natural Vegetation: Upland Prairie

Hydrometer (ASTM D422)		100.0 g
Estimated Gs =	2.65	_
Gs Corr, a =	0.99	ASTM D-422 Table 1
Lab Temp =	21	deg C
K factor =	0.01328	ASTM D-422 Table 3
Effective L (cm) = $(10.5 \text{ cm} - 8.2 \text{ cm*R} / 50 \text{ g/L})$		

 $+ 0.5*(14.0 \text{ cm} - 67.0 \text{ cm}^3/27.8 \text{ cm}^2)$ 

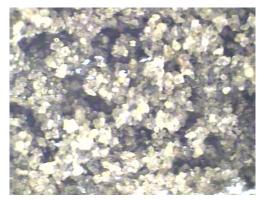
pH (ASTM D4972) =	7.97
p · · ( · · · · · · · · - /	

Organic Content (ASTM D2974)

Dry Total Mass (g) =	44.5727
Mineral Mass (g) =	
Organic Content =	1.6%

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	18	13.34	0.03430119	17.8%
5	16.5	13.59	0.02189305	16.3%
15	15	13.83	0.01275385	14.9%
30	14.5	13.92	0.00904502	14.4%
60	14	14.00	0.00641461	13.9%
250	13	14.16	0.00316086	12.9%
1440	11.5	14.41	0.00132841	11.4%



**Hydrometer Analysis (ASTM D-422)** 100% 90% 80% New Sweden Topsoil 70% 60% 50% 40% 30% 20% 10% 0% 1.00 0.10 0.01 0.00 Particle Diameter (mm)

Soil Classification: Sandy Loam SM

Material	North Mankato Subsoil
Sample Date	August 1, 2012
Sample Loc	TH 14 Sta 518+50 Ravine

Strata: Ground Moraine Geomorphology: Des Moines Lobe Ecological Province: Prairie Parkland Natural Vegetation: Prairie Wetland

Hydrometer (ASTM D422)		100.0 g
Estimated Gs =	2.65	-
Gs Corr, a =	0.99	ASTM D-422 Table 1
Lab Temp =	21	deg C
K factor =	0.01328	ASTM D-422 Table 3
Effective L (cm) = $(10.5 \text{ cm} - 8.2 \text{ cm*R} / 50 \text{ g/L})$		
$\pm 0.5*/14.0$ cm $67.0$ cm ³ /27.9 cm		

Organic Content (ASTM	D2974)
Dry Total Mass (g) =	
Mineral Mass (g) =	39.1343
Organic Content =	1.8%

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

Photomicroscopy (field of view 9 x 12 mm)

pH (ASTM D4972) =

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	31.5	11.13	0.0313264	31.2%
5	29.5	11.46	0.0201024	29.2%
15	27	11.87	0.01181197	26.7%
30	25.5	12.11	0.00843845	25.2%
60	23	12.52	0.00606703	22.8%
250	21	12.85	0.0030109	20.8%
1440	17	13.51	0.00128616	16.8%



	Hydrometer An	alysis (ASTM D-42	.2)
			100%
			90%
N	orth Mankato Subsoil		80%
			70%
			70% <u>.</u> 60% <u>.</u> 50% <u>.</u> 40% <u>.</u>
			50%
			40%
			30%
			20%
			10%
			0%
1.00	0.10	0.01	0.00
	Particle I	Diameter (mm)	

Soil Classification: Sandy Clay Loam SC-SM

Material	North Mankato Topsoil	
Sample Date	August 1, 2012	
Sample Loc	TH 14 Sta 518+50 Ravine	

Strata: Ground Moraine Geomorphology: Des Moines Lobe Ecological Province: Prairie Parkland Natural Vegetation: Prairie Wetland

Hydrometer (ASTM D422)		
2.65	-	
0.99	ASTM D-422 Table 1	
21	deg C	
0.01328	ASTM D-422 Table 3	
Effective L (cm) = $(10.5 \text{ cm} - 8.2)$		
	2.65 0.99 21 0.01328	

Organic Content (ASTM D2974)

Dry Total Mass (g) = 45.1482

Mineral Mass (g) = 42.4729

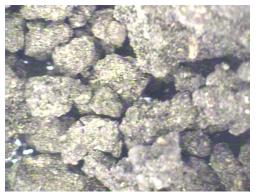
Organic Content = 5.9%

pH (ASTM D4972) =

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	35	10.55	0.03050784	34.7%
5	32	11.05	0.01973943	31.7%
15	28	11.70	0.01173006	27.7%
30	26.5	11.95	0.00838113	26.2%
60	24.5	12.28	0.00600714	24.3%
250	22	12.69	0.00299162	21.8%
1440	18.5	13.26	0.0012744	18.3%



	Hydrometer An	alysis (ASTM D-422	2)	
			100%	
			90%	
	North Mankato Topsoil		80%	
			70%	
			90% Passing 40%	
			50% Fercent	
			40% <b>a</b>	
			30%	
			20%	
			10%	
			0%	
1.00	0.10	0.01	0.00	
Particle Diameter (mm)				

Soil Classification: Organic Sandy Clay Loam SC-SM

Material	Olivia A Subsoil
Sample Date	August 21, 2012
Sample Loc	CR 62 at 1/4 mi W of US 71

Strata: Ground Moraine Geomorphology: Des Moines Lobe Ecological Province: Prairie Parkland Natural Vegetation: Upland Prairie

Hydrometer (ASTM D	100.0 g		
Estimated Gs =	2.65		
Gs Corr, a =	0.99	ASTM D-422 Table 1	
Lab Temp =	21	deg C	
K factor =	0.01328	ASTM D-422 Table 3	
Effective L (cm) = $(10.5 \text{ cm} - 8.2 \text{ cm*R} / 50 \text{ g/L})$			
		2	

Organic Content (ASTM D2974)

Dry Total Mass (g) = 46.0579

Mineral Mass (g) = 45.7711

Organic Content = 0.6%

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

Photomicroscopy (field of view 9 x 12 mm)

pH (ASTM D4972) =

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	24	12.36	0.03301217	23.8%
5	21.5	12.77	0.02122222	21.3%
15	17.5	13.42	0.01256345	17.3%
30	16.5	13.59	0.0089378	16.3%
60	14.5	13.92	0.0063958	14.4%
250	13	14.16	0.00316086	12.9%
1440	11.5	14.41	0.00132841	11.4%



	Hydrometer An	alysis (ASTM D-42	22)
			100%
			90%
	Olivia A Subsoil		80%
			<u>ت</u> 70%
			70% 300 50% 800 Fercent Passing 40%
			50%
			40% <b>A</b>
			30%
			20%
			10%
			0%
1.00	0.10	0.01	0.00
	Particle I	Diameter (mm)	

Soil Classification: Sandy Loam SM

Material	Olivia A Topsoil	
Sample Date	August 21, 2012	
Sample Loc	CR 62 at 1/4 mi W of US 71	

Strata: Ground Moraine Geomorphology: Des Moines Lobe Ecological Province: Prairie Parkland Natural Vegetation: Upland Prairie

Hydrometer (ASTM D	100.0 g		
Estimated Gs =	2.65	-	
Gs Corr, a =	0.99	ASTM D-422 Table 1	
Lab Temp =	21	deg C	
K factor =	0.01328	ASTM D-422 Table 3	
Effective L (cm) = $\frac{10.5 \text{ cm} - 8.2 \text{ cm*R}}{10.5 \text{ cm} - 8.2 \text{ cm*R}}$			

	pH (ASTM D4972) =	7.4
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Organic Content (ASTM D2974)

Dry Total Mass (g) =	44.4135
Mineral Mass (g) =	40.368
Organic Content =	9.1%

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	36.5	10.31	0.03015023	36.1%
5	31	11.21	0.01988541	30.7%
15	25.5	12.11	0.01193377	25.2%
30	22.5	12.60	0.00860812	22.3%
60	20	13.01	0.00618506	19.8%
250	16.5	13.59	0.00309614	16.3%
1440	13	14.16	0.00131702	12.9%



	Hydrometer An	alysis (ASTM D-422	2)
			100%
			90%
	Olivia A Topsoil		80%
			70% <u>s</u>
			90% Passing 40%
			50% Fercent 40%
			40% <b>a</b>
			30%
			20%
			10%
			0%
1.00	0.10	0.01	0.00
	Particle	Diameter (mm)	

Soil Classification: Organic Sandy Loam SM

Material	Olivia B Clay	
Sample Date	August 21, 2012	
Sample Loc	CSAH 4 btwn CSAH 1 & US	71

Strata: Ground Moraine Geomorphology: Des Moines Lobe Ecological Province: Prairie Parkland Natural Vegetation: Prairie Wetland

Hydrometer (ASTM D	422)	100.0 g
Estimated Gs =	2.65	
Gs Corr, a =	0.99	ASTM D-422 Table 1
Lab Temp =	21	deg C
K factor =	0.01328	ASTM D-422 Table 3
Effective L (cm) =	2 cm*R / 50 g/L)	
		2

Organic Content (ASTM D2974)

Dry Total Mass (g) = 44.5888

Mineral Mass (g) = 43.7853

Organic Content = 1.8%

pH (ASTM D4972) =

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	32	11.05	0.03121077	31.7%
5	28.5	11.62	0.02024576	28.2%
15	25.5	12.11	0.01193377	25.2%
30	23.5	12.44	0.00855194	23.3%
60	22	12.69	0.00610663	21.8%
250	20	13.01	0.00303005	19.8%
1440	16	13.67	0.00129395	15.8%



	Hydrometer An	alysis (ASTM D-42	2)
			100%
			90%
	Olivia B Clay		80%
	·		70% <u></u>
			90% Passing 40%
			50% <b>ent</b>
			90% A0% Percent
			30%
			20%
			10%
			0%
1.00	0.10	0.01	0.00
	Particle	Diameter (mm)	

Soil Classification: Sandy Clay Loam SC-SM

Material	Owatonna Clay	
Sample Date	August 21, 2012	
Sample Loc	TH 14 Sta 1569+48 SP#740	1-3

Strata: Ground Moraine Geomorphology: Des Moines Lobe Ecological Province: Eastern Broadleaf Forest

pH (ASTM D4972) =

Natural Vegetation: Oak Woodland

Hydrometer (ASTM D	422)	100.0 g
Estimated Gs =	2.65	•
Gs Corr, a =	0.99	ASTM D-422 Table 1
Lab Temp =	21	deg C
K factor =	0.01328	ASTM D-422 Table 3
Effective L (cm) =	(10.5 cm - 8.2	2 cm*R / 50 g/L)

Organic Content (ASTM D2974)

Dry Total Mass (g) = 42.9901

Mineral Mass (g) = 41.8209

Organic Content = 2.7%

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	30	11.37	0.03167073	29.7%
5	28	11.70	0.02031707	27.7%
15	25.5	12.11	0.01193377	25.2%
30	24	12.36	0.00852371	23.8%
60	22.5	12.60	0.00608686	22.3%
250	20.5	12.93	0.00302049	20.3%
1440	17	13.51	0.00128616	16.8%



Hydrometer Analysis (ASTM D-422)				
			100%	
			90%	
	Owatonna Clay		80%	
			70% <u>د م</u>	
			60% <b>se</b>	
			Percent Passing 40%	
			40%	
			30%	
			20%	
			10%	
			0%	
1.00	0.10	0.01	0.00	
	Particle	Diameter (mm)		

Material	Owatonna Topsoil	
Sample Date	August 21, 2012	
Sample Loc	TH 14 Sta 1569+48 SP#740	1-34

Strata: Ground Moraine Geomorphology: Des Moines Lobe **Ecological Province: Eastern Broadleaf Forest** 

pH (ASTM D4972) =

Natural Vegetation: Oak Woodland

Hydrometer (ASTM D422)		100.0	g
Estimated Gs =	2.65		-
Gs Corr, a =	0.99	ASTM D-422	Table 1
Lab Temp =	21	deg C	
K factor =	0.01328	ASTM D-422	Table 3
Effective L (cm) = $\frac{10.5 \text{ cm} - 8.2 \text{ cm}^*\text{R} / 50 \text{ g/L}}{10.5 \text{ cm} - 8.2 \text{ cm}^*\text{R}}$			

Dry Total Mass (g) = 42.9447 Mineral Mass (g) =

Organic Content =

Organic Content (ASTM D2974)

40.8306 4.9%

7.2

 $+ 0.5*(14.0 \text{ cm} - 67.0 \text{ cm}^3/27.8 \text{ cm}^2)$ 

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	30	11.37	0.03167073	29.7%
5	27.5	11.78	0.02038812	27.2%
15	24	12.36	0.01205434	23.8%
30	22.5	12.60	0.00860812	22.3%
60	21	12.85	0.00614597	20.8%
250	19	13.18	0.00304908	18.8%
1440	16	13.67	0.00129395	15.8%



**Hydrometer Analysis (ASTM D-422)** 100% 90% 80% Owatonna Topsoil 70% 60% 50% 40% 30% 20% 10% 0% 1.00 0.10 0.01 0.00 Particle Diameter (mm)

Material	Perham Subsoil
Sample Date	July 21, 2012
Sample Loc	TH 10 & CR 34 SP#5607-42

Strata: Outwash

Geomorphology: Des Moines Lobe

**Ecological Province: Eastern Broadleaf Forest** 

Natural Vegetation: Oak Woodland

Hydrometer (ASTM D422)		100.0	g
Estimated Gs =	2.65		•
Gs Corr, a =	0.99	ASTM D-422	Table 1
Lab Temp =	21	deg C	
K factor =	0.01328	ASTM D-422	Table 3
Effective L (cm) = $(10.5 \text{ cm} - 8.2 \text{ cm*R} / 50 \text{ g/L})$			

 $+ 0.5*(14.0 \text{ cm} - 67.0 \text{ cm}^3/27.8 \text{ cm}^2)$ 

Organic Content (ASTM D2974)

pH (ASTM D4972) =

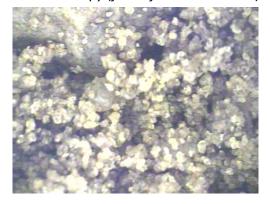
Dry Total Mass (g) = 44.8753 Mineral Mass (g) = 44.5782

Organic Content =

0.7%

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	9	14.82	0.03614864	8.9%
5	8.5	14.90	0.02292558	8.4%
15	8.5	14.90	0.01323609	8.4%
30	8.5	14.90	0.00935933	8.4%
60	8	14.98	0.00663623	7.9%
250	8	14.98	0.00325107	7.9%
1440	7	15.15	0.00136201	6.9%



**Hydrometer Analysis (ASTM D-422)** 100% 90% 80% Perham Subsoil 70% 60% 50% 40% 30% 20% 10% 0% 1.00 0.10 0.01 0.00 Particle Diameter (mm)

Soil Classification: Sandy Loam SM

Material	Perham Topsoil
Sample Date	July 21, 2012
Sample Loc	TH 10 & CR 34 SP#5607-42

Strata: Outwash

Geomorphology: Des Moines Lobe

Ecological Province: Eastern Broadleaf Forest

pH (ASTM D4972) =

Natural Vegetation: Oak Woodland

Hydrometer (ASTM D422)		100.0 g	
Estimated Gs =	2.65	_	
Gs Corr, a =	0.99	ASTM D-422 Table 1	
Lab Temp =	21	deg C	
K factor =	0.01328	ASTM D-422 Table 3	
Effective L (cm) = $\frac{10.5 \text{ cm} - 8.2 \text{ cm*R}}{10.5 \text{ cm}} = \frac{10.5 \text{ cm}}{10.5 \text{ cm}} = $			

 $+ 0.5*(14.0 \text{ cm} - 67.0 \text{ cm}^3/27.8 \text{ cm}^2)$ 

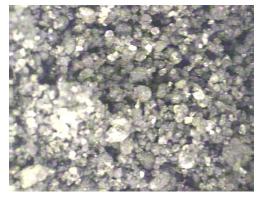
Organic Content (ASTM D2974)

Dry Total Mass (g) = 44.8498 Mineral Mass (g) = 43.736 2.5%

Organic Content =

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	12	14.33	0.0355435	11.9%
5	11.5	14.41	0.02254392	11.4%
15	10.5	14.57	0.0130896	10.4%
30	10	14.65	0.00928175	9.9%
60	9.5	14.74	0.00658152	9.4%
250	9	14.82	0.00323323	8.9%
1440	8	14.98	0.00135461	7.9%



## **Hydrometer Analysis (ASTM D-422)** 100% 90% 80% Perham Topsoil 70% 60% 50% 40% 30% 20% 10% 0% 1.00 0.10 0.01 0.00 Particle Diameter (mm)

Material Pipestone Subsoil		
Sample Date	not stated	
Sample Loc	CSAH 10 at 1.5 mi F of TH	75

Strata: Gray Drift Geomorphology: Kansan Glaciation Ecological Province: Prairie Parkland Natural Vegetation: Upland Prairie

pH (ASTM D4972) =

Hydrometer (ASTM D422)		100.0 g
Estimated Gs =	2.65	-
Gs Corr, a =	0.99	ASTM D-422 Table 1
Lab Temp =	21	deg C
K factor =	0.01328	ASTM D-422 Table 3
Effective L (cm) = $\frac{10.5 \text{ cm} - 8.2 \text{ cm*R}}{10.5 \text{ cm} - 8.2 \text{ cm*R}}$		

Organic Content (ASTM D2974)

Dry Total Mass (g) = 40.6169

Mineral Mass (g) = 39.8541

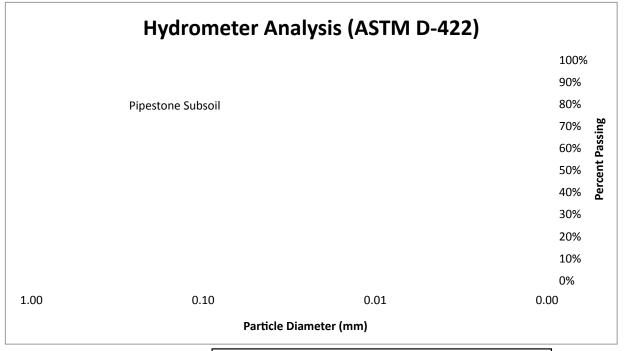
Organic Content = 1.9%

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	23	12.52	0.03323048	22.8%
5	21	12.85	0.02129026	20.8%
15	18.5	13.26	0.01248648	18.3%
30	17	13.51	0.00891079	16.8%
60	16	13.67	0.00633902	15.8%
250	15	13.83	0.00312404	14.9%
1440	13	14.16	0.00131702	12.9%





Soil Classification: Sandy Loam SM

Material	Pipestone Topsoil	
Sample Date	not stated	
Sample Loc	CSAH 10 at 1.5 mi F of TH	75

Strata: Gray Drift Geomorphology: Kansan Glaciation **Ecological Province: Prairie Parkland** Natural Vegetation: Upland Prairie

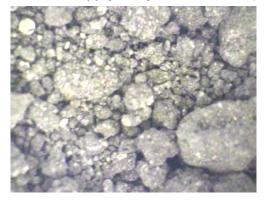
Hydrometer (ASTM D422)		100.0 g	
Estimated Gs =	2.65		
Gs Corr, a =	0.99	ASTM D-422 Table 1	
Lab Temp =	21	deg C	
K factor =	0.01328	ASTM D-422 Table 3	
Effective L (cm) = $(10.5 \text{ cm} - 8.2 \text{ cm*R} / 50 \text{ g/L})$			

pH (ASTM D4972) = Organic Content (ASTM D2974) Dry Total Mass (g) = 39.1515 Mineral Mass (g) = 36.4854 Organic Content = 6.8%

 $+ 0.5*(14.0 \text{ cm} - 67.0 \text{ cm}^3/27.8 \text{ cm}^2)$ 

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	27	11.87	0.03234841	26.7%
5	22.5	12.60	0.0210855	22.3%
15	19	13.18	0.01244781	18.8%
30	17	13.51	0.00891079	16.8%
60	16	13.67	0.00633902	15.8%
250	14	14.00	0.00314251	13.9%
1440	12	14.33	0.00132463	11.9%



	Hydrometer An	alysis (ASTM D-42	2)	
			100%	
			90%	
	Pipestone Topsoil		80%	
			70% . <b>5</b>	
			<b>60% 8</b>	
			40% A09	
			40% <b>a</b>	
			30%	
			20%	
			10%	
			0%	
1.00	0.10	0.01	0.00	
	Particle Diameter (mm)			

Soil Classification: Organic Sandy Loam SM

Material	Ramsey Peat	
Sample Date	not stated	
Sample Loc	35 E & Maryland Ave	

Strata: Glacial Lake Sediment Geomorphology: Des Moines Lobe Ecological Province: Prairie Parkland Natural Vegetation: Upland Prairie

pH (ASTM D4972) =

Hydrometer (ASTM D	422)	100.0 g
Estimated Gs =	2.65	
Gs Corr, a =	0.99	ASTM D-422 Table 1
Lab Temp =	21	deg C
K factor =	0.01328	ASTM D-422 Table 3
Effective L (cm) =	(10.5 cm - 8.2	2 cm*R / 50 g/L)

Organic Content (ASTM D2974)

Dry Total Mass (g) = 45.8799

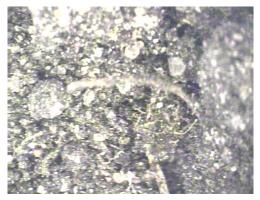
Mineral Mass (g) = 36.8351

Organic Content = 19.7%

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	20.5	12.93	0.03377008	20.3%
5	17.5	13.42	0.02176054	17.3%
15	15	13.83	0.01275385	14.9%
30	13.5	14.08	0.00909816	13.4%
60	12	14.33	0.00648933	11.9%
250	10.5	14.57	0.00320628	10.4%
1440	9	14.82	0.00134718	8.9%



	Hydrometer An	alysis (ASTM D-42	2)
			100%
			90%
	Ramsey Peat		80%
			70%
			90% Passing 40%
			50% Fercent 40%
			40% ع
			30%
			20%
			10%
			0%
1.00	0.10	0.01	0.00
	Particle	Diameter (mm)	

Material	Rockford Cornfield	
Sample Date	not stated	
Sample Loc	TH 55	

Strata: End Moraine Geomorphology: Des Moines Lobe

Ecological Province: Eastern Broadleaf Forest Natural Vegetation: Maple Basswood Forest

pH (ASTM D4972) =

Hydrometer (ASTM D	422)	100.0	g
Estimated Gs =	2.65		-
Gs Corr, a =	0.99	ASTM D-422	Table 1
Lab Temp =	21	deg C	
K factor =	0.01328	ASTM D-422	Table 3
Effective L (cm) =	(10.5 cm - 8.3	2 cm*R / 50 g	/L)

Organic Content (ASTM D2974)

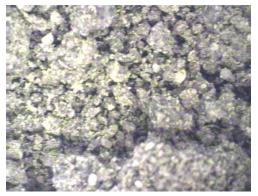
Dry Total Mass (g) = 34.8561

Mineral Mass (g) = 33.306 Organic Content = 4.4%

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	24.5	12.28	0.03290248	24.3%
5	23	12.52	0.0210168	22.8%
15	20	13.01	0.01237012	19.8%
30	18.5	13.26	0.00882927	18.3%
60	17.5	13.42	0.00628173	17.3%
250	16.5	13.59	0.00309614	16.3%
1440	14	14.00	0.00130938	13.9%



	Hydrometer An	alysis (ASTM D-42	2)
			100%
			90%
	Rockford Cornfield		80%
			70%
			Percent Passing %05
			50% <b>5</b> 0%
			40%
			30%
			20%
			10%
			0%
1.00	0.10	0.01	0.00
	Particle I	Diameter (mm)	

Material	Rockford Subsoil	
Sample Date	not stated	
Sample Loc	TH 55	

Strata: End Moraine Geomorphology: Des Moines Lobe

Ecological Province: Eastern Broadleaf Forest Natural Vegetation: Maple Basswood Forest

Hydrometer (ASTM D422)		100.0	g
Estimated Gs =	2.65		
Gs Corr, a =	0.99	ASTM D-422	Table 1
Lab Temp =	21	deg C	
K factor =	0.01328	ASTM D-422	Table 3
Effective L (cm) = $(10.5 \text{ cm} - 8.2 \text{ cm*R} / 50 \text{ g/L})$			
+ 0.5*(14.0 cm - 67.0 cm ³ /27.8 cm ² )			

pH (ASTM D4972) = 6.9

Organic Content (ASTM D2974)

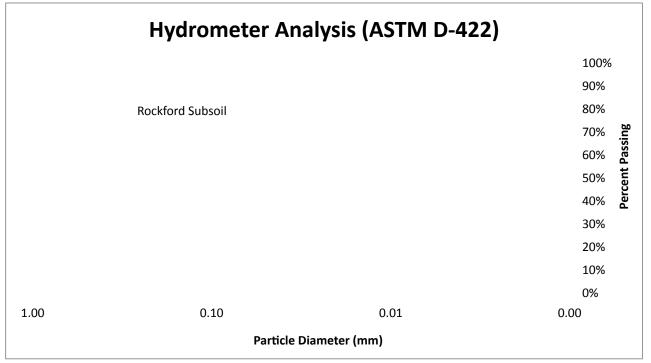
Dry Total Mass (g) = 33.5297

Mineral Mass (g) = 32.7203 Organic Content = 2.4%

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	30	11.37	0.03167073	29.7%
5	27.5	11.78	0.02038812	27.2%
15	25	12.19	0.0119741	24.8%
30	23.5	12.44	0.00855194	23.3%
60	22	12.69	0.00610663	21.8%
250	19.5	13.10	0.00303958	19.3%
1440	16.5	13.59	0.00129006	16.3%





Material	Rockford Topsoil	
Sample Date	not stated	
Sample Loc	TH 55	

Strata: End Moraine Geomorphology: Des Moines Lobe

Ecological Province: Eastern Broadleaf Forest Natural Vegetation: Maple Basswood Forest

Hydrometer (ASTM D422)		100.0 g	
Estimated Gs =	2.65	•	
Gs Corr, a =	0.99	ASTM D-422 Table 1	
Lab Temp =	21	deg C	
K factor =	0.01328	ASTM D-422 Table 3	
Effective L (cm) = $(10.5 \text{ cm} - 8.2 \text{ cm*R} / 50 \text{ g/L})$			
	0 = 4/4 4 0	3/0= 0	

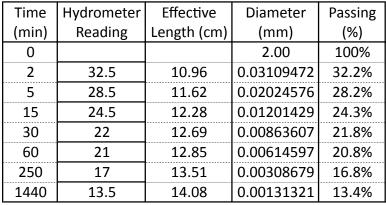
pH (ASTM D4972) = 7.4

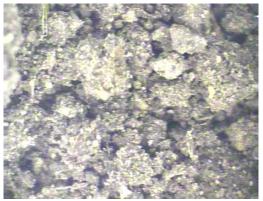
Organic Content (ASTM D2974)

Dry Total Mass (g) = 36.8538 Mineral Mass (g) = 35.5014 Organic Content = 3.7%

 $+ 0.5*(14.0 \text{ cm} - 67.0 \text{ cm}^3/27.8 \text{ cm}^2)$ 

Photomicroscopy (field of view 9 x 12 mm)





Hydrometer Analysis (ASTM D-422)			
			100%
			90%
	Rockford Topsoil		80%
			70%
			90% Passing 40% 20%
			50% <b>J</b>
			40%
			30%
			20%
			10%
			0%
1.00	0.10	0.01	0.00
Particle Diameter (mm)			

8

Material Sandstone Subsoil
Sample Date October 10, 2012
Sample Loc North of TH 18, West of CH 61

Strata: Ground Moraine Geomorphology: Superior Lobe

Ecological Province: Laurentian Mixed Forest Natural Vegetation: Norther Hardwood Forest

pH (ASTM D4972) =

Hydrometer (ASTM D422)		100.0	g
Estimated Gs =	2.65		-
Gs Corr, a =	0.99	ASTM D-422	Table 1
Lab Temp =	21	deg C	
K factor =	0.01328	ASTM D-422	Table 3
Effective L $(cm) = \frac{10.5 \text{ cm}}{10.5 \text{ cm}} = \frac{2.2 \text{ cm*P}}{10.5 \text{ cm}} = \frac{10.5 \text{ cm}}{10.5 \text{ cm}} = \frac{10.5 \text{ cm}}$			

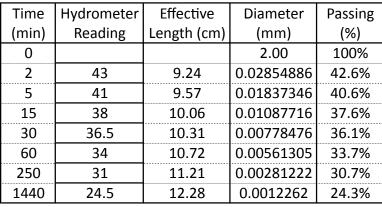
Organic Content (ASTM D2974)

Dry Total Mass (g) = 53.2787 Mineral Mass (g) = 51.9657 Organic Content = 2.5%

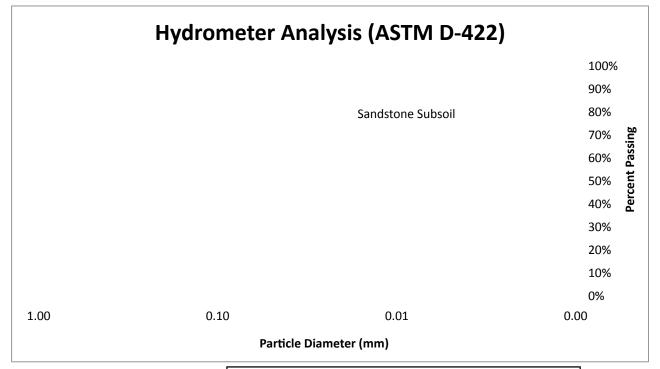
Effective L (cm) = (10.5 cm - 8.2 cm*R / 50 g/L)

 $+ 0.5*(14.0 \text{ cm} - 67.0 \text{ cm}^3/27.8 \text{ cm}^2)$ 

Photomicroscopy (field of view 9 x 12 mm)







Soil Classification: Slightly Organic Clay Loam CL-ML-OL

Material	Sandstone Topsoil	
Sample Date	October 10, 2012	
Sample Loc	North of CR 32, East of CH	22

Strata: Ground Moraine Geomorphology: Superior Lobe

Ecological Province: Laurentian Mixed Forest Natural Vegetation: Norther Hardwood Forest

Hydrometer (ASTM D422)		100.0	g
Estimated Gs =	2.65		-
Gs Corr, a =	0.99	ASTM D-422	Table 1
Lab Temp =	21	deg C	
K factor =	0.01328	ASTM D-422	Table 3
Effective L (cm) = $(10.5 \text{ cm} - 8.2 \text{ cm*R} / 50 \text{ g/L})$			/L)

pH (ASTM D4972) = 7.9

Organic Content (ASTM D2974)

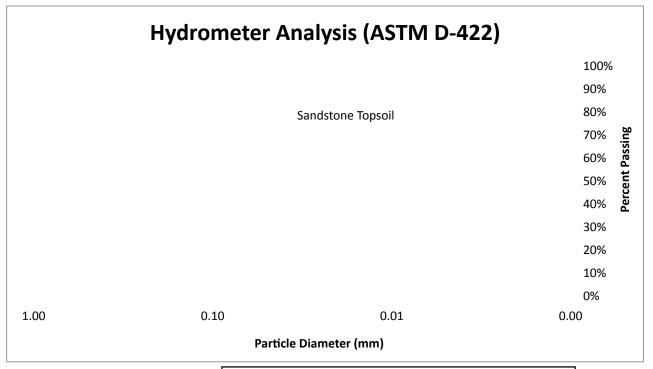
Dry Total Mass (g) =	55.5695
Mineral Mass (g) =	53.274
Organic Content =	4.1%

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	21	12.85	0.03366285	20.8%
5	17.5	13.42	0.02176054	17.3%
15	15.5	13.75	0.012716	15.3%
30	14	14.00	0.00907163	13.9%
60	13	14.16	0.00645208	12.9%
250	11.5	14.41	0.00318819	11.4%
1440	10.5	14.57	0.00133595	10.4%





Soil Classification: Slightly Organic Sandy Loam SM

Material	Stillwater Sediment
Sample Date	July 27, 2012
Sample Loc	Bridge Pier River Boring

Strata: End Moraine Geomorphology: Superior Lobe

Ecological Province: Eastern Broadleaf Forest

pH (ASTM D4972) =

Natural Vegetation: Oak Woodland

Hydrometer (ASTM D422)		100.0	g
Estimated Gs =	2.65		-
Gs Corr, a =	0.99	ASTM D-422	Table 1
Lab Temp =	21	deg C	
K factor =	0.01328	ASTM D-422	Table 3
Effective L (cm) = $(10.5 \text{ cm} - 8.2 \text{ cm*R} / 50 \text{ g/L})$			/L)

Organic Content (ASTM D2974)

Dry Total Mass (g) = 35.5434 Mineral Mass (g) = 32.1189

Organic Content =

9.6%

7.5

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	44.5	9.00	0.02816638	44.1%
5	34	10.72	0.01944417	33.7%
15	29	11.54	0.01164758	28.7%
30	24	12.36	0.00852371	23.8%
60	19.5	13.10	0.00620451	19.3%
250	14	14.00	0.00314251	13.9%
1440	10	14.65	0.00133971	9.9%



### **Hydrometer Analysis (ASTM D-422)** 100% 90% 80% Stillwater Sediment **Percent Passing** 70% 60% 50% 40% 30% 20% 10% 0% 0.01 1.00 0.10 0.00 Particle Diameter (mm)

Soil Classification: Organic Loam OL

Material	Tamarack Clay
Sample Date	September 17, 2012
Sample Loc	CSAH 32 Sta 137+36 40'Lt

Strata: End Moraine Geomorphology: Des Moines Lobe

**Ecological Province: Laurentian Mixed Forest** 

Natural Vegetation: Peatland

Hydrometer (ASTM D422)		100.0 g
Estimated Gs =	2.65	
Gs Corr, a =	0.99	ASTM D-422 Table 1
Lab Temp =	21	deg C
K factor =	0.01328	ASTM D-422 Table 3
Effective L (cm) = $(10.5 \text{ cm} - 8.2 \text{ cm*R} / 50 \text{ g/L})$		

pH (ASTM D4972) = 7.1

Organic Content (ASTM D2974)

Dry Total Mass (g) = 38.2459 Mineral Mass (g) = 37.0922 Organic Content = 3.0%

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	50.5	8.01	0.02658151	50.0%
5	49	8.26	0.01706773	48.5%
15	47	8.59	0.01004783	46.5%
30	44	9.08	0.0073056	43.6%
60	40.5	9.65	0.00532663	40.1%
250	33	10.88	0.00277078	32.7%
1440	23.5	12.44	0.00123437	23.3%



**Hydrometer Analysis (ASTM D-422)** 100% 90% 80% Tamarack Clay 70% 60% 50% 40% 30% 20% 10% 0% 1.00 0.10 0.01 0.00 Particle Diameter (mm)

Soil Classification: Slightly Organic Clay CL-OL

Material Tamarack Peat
Sample Date September 17, 2012
CSAH 32 Sta 152+85 30'Lt

Strata: Peat

Geomorphology: Des Moines Lobe

Ecological Province: Laurentian Mixed Forest

Natural Vegetation: Peatland

Hydrometer (ASTM D422) 100.0 g

Estimated Gs = 2.65

Gs Corr, a = 0.99 ASTM D-422 Table 1

Lab Temp = 21 deg C

K factor = 0.01328 ASTM D-422 Table 3

Effective L (cm) = (10.5 cm - 8.2 cm*R / 50 g/L)

Organic Content (ASTM D2974)

pH (ASTM D4972) =

Dry Total Mass (g) = 7.7385Mineral Mass (g) = 0.7205

Organic Content =

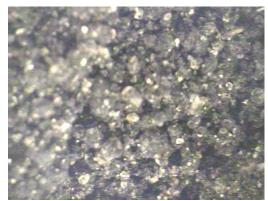
90.7%

6.9

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	29	11.54	0.03189822	28.7%
5	28	11.70	0.02031707	27.7%
15	26.5	11.95	0.01185271	26.2%
30	25	12.19	0.00846696	24.8%
60	18	13.34	0.00626251	17.8%
250	13	14.16	0.00316086	12.9%
1440	14	14.00	0.00130938	13.9%



**Hydrometer Analysis (ASTM D-422)** 100% 90% 80% Tamarack Peat 70% 60% 50% 40% 30% 20% 10% 0% 1.00 0.10 0.01 0.00 Particle Diameter (mm)

Soil Classification: Peat PT

Material Tamarack Topsoil
Sample Date September 17, 2012
Sample Loc CSAH 32 Sta 120+70 34'Lt

Strata: End Moraine Geomorphology: Des Moines Lobe

Ecological Province: Laurentian Mixed Forest

Natural Vegetation: Peatland

Hydrometer (ASTM D422) 100.0 g

Estimated Gs = 2.65

Gs Corr, a = 0.99 ASTM D-422 Table 1

Lab Temp = 21 deg C

K factor = 0.01328 ASTM D-422 Table 3

Effective L (cm) = (10.5 cm - 8.2 cm*R / 50 g/L)

 $+ 0.5*(14.0 \text{ cm} - 67.0 \text{ cm}^3/27.8 \text{ cm}^2)$ 

pH (ASTM D4972) =	6.9

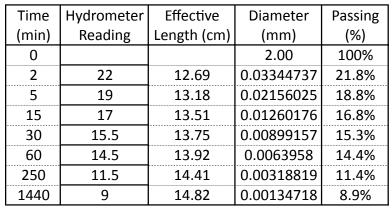
Organic Content (ASTM D2974)

Dry Total Mass (g) = 37.4814

Mineral Mass (g) = 35.828

Organic Content = 4.4%

Photomicroscopy (field of view 9 x 12 mm)





Hydrometer Analysis (ASTM D-422)						
			100%			
			90%			
	Tamarack Topsoil		80%			
			70%			
			60%			
			90% Bercen 40% A0% A0%			
			40% <b>a</b>			
			30%			
			20%			
			10%			
			0%			
1.00	0.10	0.01	0.00			
Particle Diameter (mm)						

Soil Classification: Slightly Organic Sandy Loam SM

Wabasha Subsoil (Box) Material Sample Date September 21, 2012 Sample Loc | CR 4 @ Handshaw Coulee (?)

Strata: Colluvuum Geomorphology: Weathered Residual **Ecological Province: Eastern Broadleaf Forest** 

Natural Vegetation: Oak Woodland

Hydrometer (ASTM D422)		100.0	g
Estimated Gs =	2.65		
Gs Corr, a =	0.99	ASTM D-422	Table 1
Lab Temp =	21	deg C	
K factor =	0.01328	ASTM D-422	Table 3
Effective L (cm) =	(10.5 cm - 8.2	2 cm*R / 50 g	/L)

+ 0.5*(14.0 cm - 67.0 cm³/27.8 cm²)

pH (ASTM D4972) =	7.8

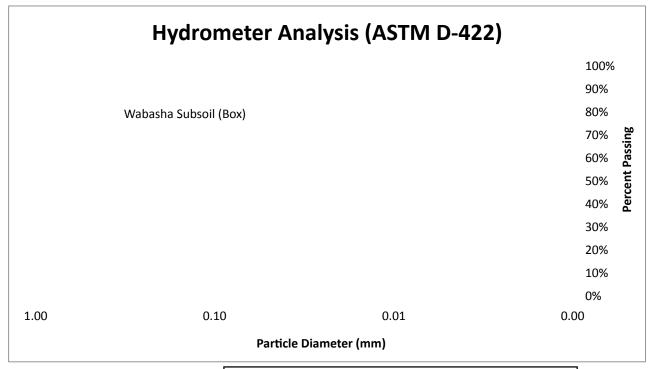
#### Organic Content (ASTM D2974)

· · · · · · · · · · · · · · · · · · ·	
Dry Total Mass (g) =	40.738
Mineral Mass (g) =	39.5298
Organic Content =	3.0%

Photomicroscopy (field of view 9 x 12 mm)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	27.5	11.78	0.03223645	27.2%
5	20.5	12.93	0.02135808	20.3%
15	16	13.67	0.01267804	15.8%
30	15	13.83	0.00901834	14.9%
60	13.5	14.08	0.00643337	13.4%
250	12.5	14.24	0.00317	12.4%
1440	11	14.49	0.00133219	10.9%





Soil Classification: Slightly Organic Sandy Loam SM

8

Material Wabasha Subsoil (Rock) Sample Date September 21, 2012 Sample Loc | CR 4 @ Handshaw Coulee (?)

Strata: Colluvuum Geomorphology: Weathered Residual **Ecological Province: Eastern Broadleaf Forest** 

Natural Vegetation: Oak Woodland

Hydrometer (ASTM D422)		100.0 g
Estimated Gs =	2.65	
Gs Corr, a =	0.99	ASTM D-422 Table 1
Lab Temp =	21	deg C
K factor =	0.01328	ASTM D-422 Table 3
Fffective L (cm) =	(10.5 cm - 8.1	- 2 cm*R / 50 g/L)

Organic Content (ASTM D2974)

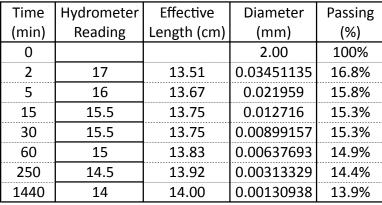
pH (ASTM D4972) =

Dry Total Mass (g) = 48.1498 Mineral Mass (g) = 47.2421

Organic Content = 1.9%

 $+ 0.5*(14.0 \text{ cm} - 67.0 \text{ cm}^3/27.8 \text{ cm}^2)$ 

Photomicroscopy (field of view 9 x 12 mm)





### **Hydrometer Analysis (ASTM D-422)** 100% 90% 80% Wabasha Subsoil (Rock) Percent Passing 70% 60% 50% 40% 30% 20% 10% 0% 1.00 0.10 0.01 0.00 Particle Diameter (mm)

Soil Classification: Sandy Loam SM

7.5

Material Wabasha Topsoil Sample Date September 21, 2012 Sample Loc | CR 4 @ Handshaw Coulee (?)

Strata: Colluvuum

Geomorphology: Weathered Residual **Ecological Province: Eastern Broadleaf Forest** 

Natural Vegetation: Oak Woodland

Hydrometer (ASTM D422)		100.0	g
Estimated Gs =	2.65		-
Gs Corr, a =	0.99	ASTM D-422	Table 1
Lab Temp =	21	deg C	
K factor =	0.01328	ASTM D-422	Table 3
Effective L (cm) = $\frac{10.5 \text{ cm} - 8.2 \text{ cm} \text{ R}}{10.5 \text{ cm}}$			

Organic Content (ASTM D2974)

pH (ASTM D4972) =

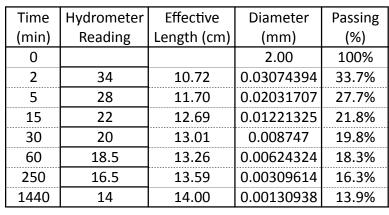
Dry Total Mass (g) = 43.8146 Mineral Mass (g) = 41.9462

Organic Content =

4.3%

 $+ 0.5*(14.0 \text{ cm} - 67.0 \text{ cm}^3/27.8 \text{ cm}^2)$ 

Photomicroscopy (field of view 9 x 12 mm)





**Hydrometer Analysis (ASTM D-422)** 100% 90% 80% Wabasha Topsoil Percent Passing 70% 60% 50% 40% 30% 20% 10% 0% 1.00 0.10 0.01 0.00 Particle Diameter (mm)

Soil Classification: Slightly Organic Sandy Loam SM

Material **Worthington Subsoil** Sample Date July 16, 2012 Sample Loc | TH 60 SP#5305-58

Strata: Ground Moraine Geomorphology: Des Moines Lobe Ecological Province: Prairie Parkland Natural Vegetation: Upland Prairie

Hydrometer (ASTM D422)		100.0 g
Estimated Gs =	2.65	_
Gs Corr, a =	0.99	ASTM D-422 Table 1
Lab Temp =	21	deg C
K factor =	0.01328	ASTM D-422 Table 3
Effective L (cm) =	(10.5 cm - 8.7	2 cm*R / 50 g/L)

Dry Total Mass (g) = Mineral Mass (g) =

46.2904 45.2711 Organic Content = 2.2%

7.6

 $+ 0.5*(14.0 \text{ cm} - 67.0 \text{ cm}^3/27.8 \text{ cm}^2)$ 

Photomicroscopy (field of view 9 x 12 mm)

pH (ASTM D4972) =

Organic Content (ASTM D2974)

Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	28	11.70	0.0321241	27.7%
5	25.5	12.11	0.0206699	25.2%
15	22.5	12.60	0.01217372	22.3%
30	21	12.85	0.00869171	20.8%
60	20	13.01	0.00618506	19.8%
250	18	13.34	0.00306799	17.8%
1440	16.5	13.59	0.00129006	16.3%



**Hydrometer Analysis (ASTM D-422)** 100% 90% 80% Worthington Subsoil Percent Passing 70% 60% 50% 40% 30% 20% 10% 0% 1.00 0.10 0.01 0.00 Particle Diameter (mm)

> Soil Classification: Slightly Organic Sandy Loam SM

7.4

Material	Worthington Topsoil	
Sample Date	July 16, 2012	
Sample Loc	TH 60 SP#5305-58	

Strata: Ground Moraine Geomorphology: Des Moines Lobe Ecological Province: Prairie Parkland Natural Vegetation: Upland Prairie

Hydrometer (ASTM D422)		100.0 g
Estimated Gs =	2.65	
Gs Corr, a =	0.99	ASTM D-422 Table 1
Lab Temp =	21	deg C
K factor =	0.01328	ASTM D-422 Table 3
Effective L (cm) = $(10.5 \text{ cm} - 8.2 \text{ cm*R} / 50 \text{ g/L})$		
	. 0 5*/1/0	one C7 0 one 3/27 0 one

Organic Content (ASTM D2974) Dry Total Mass (g) = 45.0201 Mineral Mass (g) = 42.4371 Organic Content = 5.7%

 $+ 0.5*(14.0 \text{ cm} - 67.0 \text{ cm}^3/27.8 \text{ cm}^2)$ 

Photomicroscopy (field of view 9 x 12 mm)

pH (ASTM D4972) =

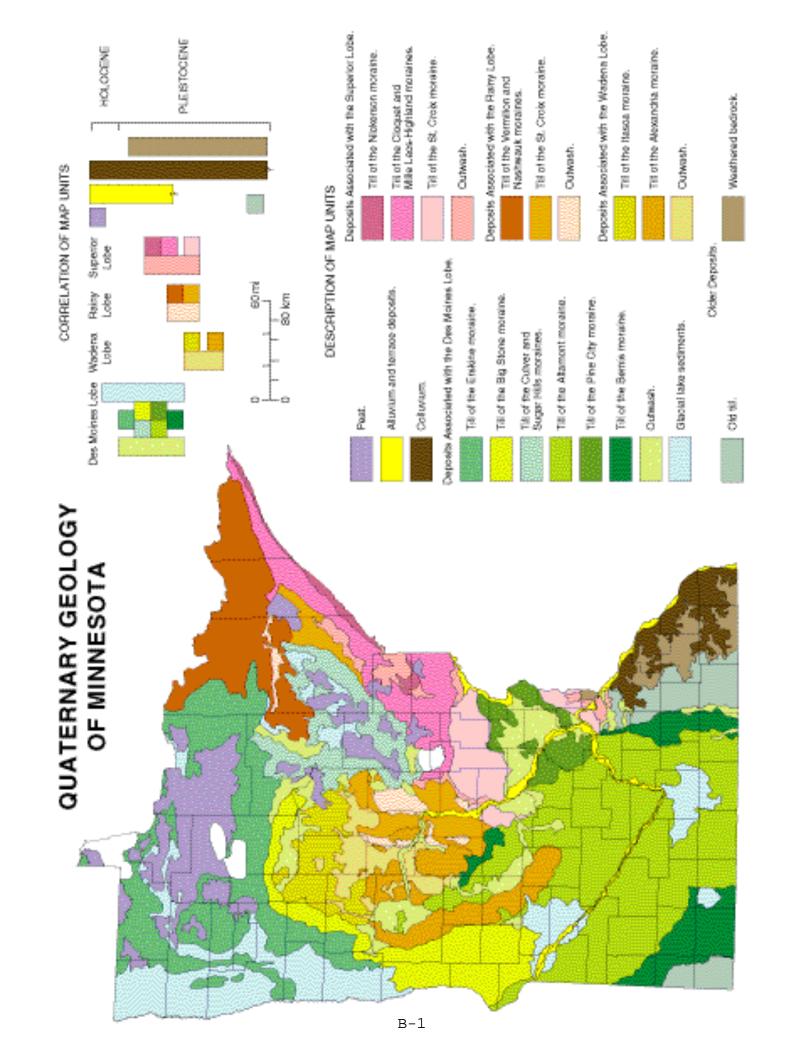
Time	Hydrometer	Effective	Diameter	Passing
(min)	Reading	Length (cm)	(mm)	(%)
0			2.00	100%
2	26	12.03	0.03257116	25.7%
5	23	12.52	0.0210168	22.8%
15	20	13.01	0.01237012	19.8%
30	18.5	13.26	0.00882927	18.3%
60	17.5	13.42	0.00628173	17.3%
250	15	13.83	0.00312404	14.9%
1440	13	14.16	0.00131702	12.9%

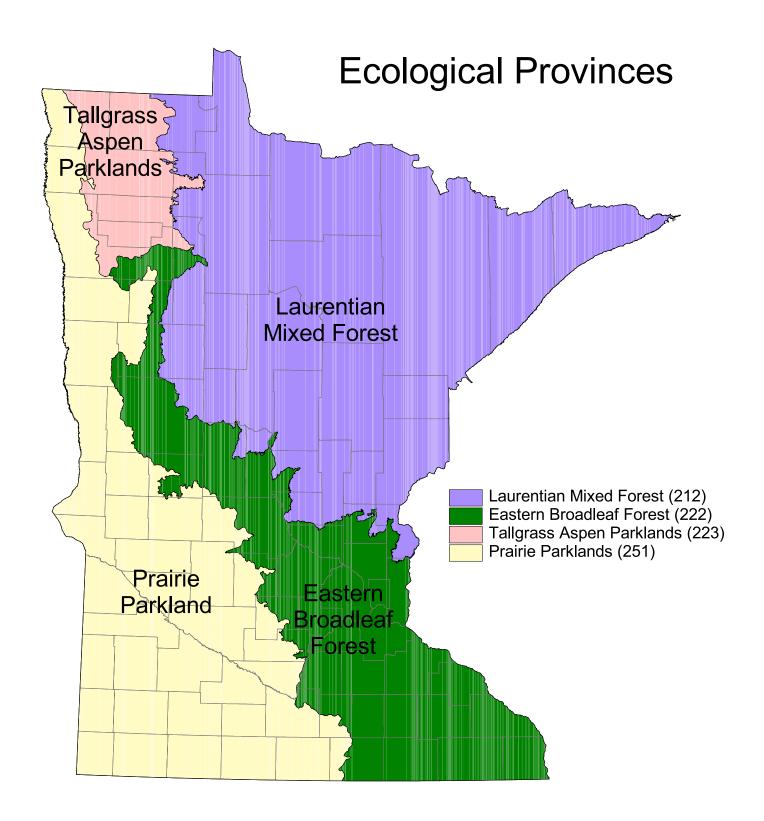


**Hydrometer Analysis (ASTM D-422)** 100% 90% 80% **Worthington Topsoil Percent Passing** 70% 60% 50% 40% 30% 20% 10% 0% 1.00 0.10 0.01 0.00 Particle Diameter (mm)

Soil Classification: Organic Sandy Loam SM

Appendix B	– Geological,	Geomorphol	logical, Ecolo	ogical and Bo	tanical Maps





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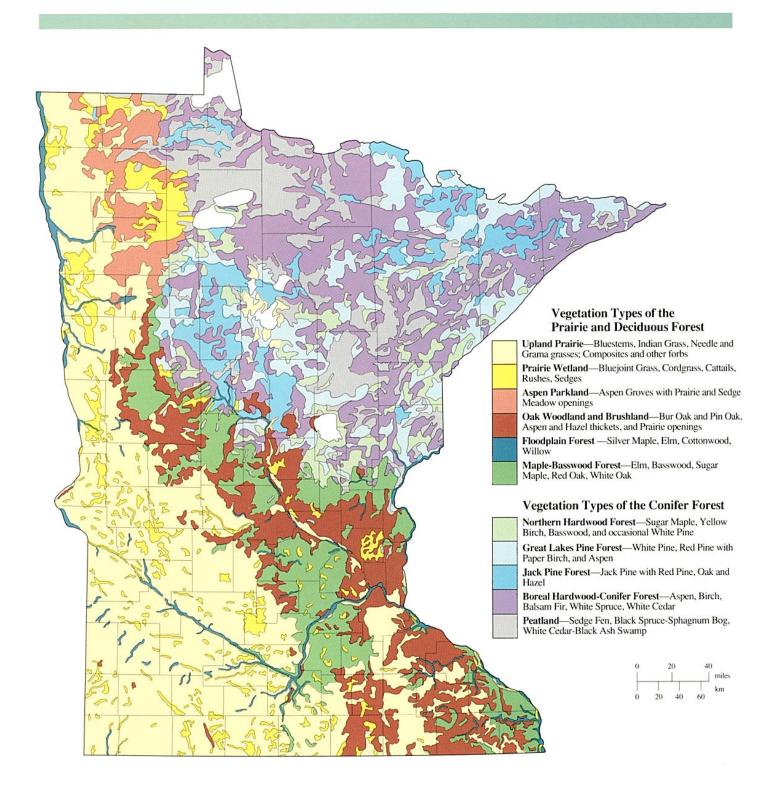
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MN Center for Environmental Advocacy
MN Department of Agriculture
MN Department of Natural Resources
Natural Resources Conservation Service
Potlatch Corporation
USDA Forest Service
U.S. Fish and Wildlife Service

For more information contact: ECS Specialist MN DNR, Division of Forestry Resource Assessment Program 413 SE 13 Street Grand Rapids, MN 55744 (218) 327-4449 ext 239

September, 2000



Division of Forestry Ecological Land Classification Program

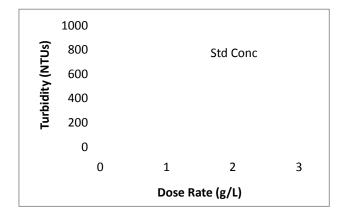


## The Natural Vegetation of Minnesota at the Time of the Public Land Survey: 1847-1907

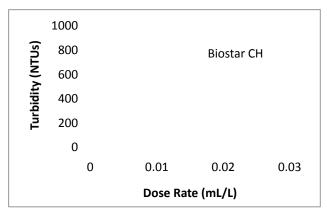
This map was adapted by Barbara Coffin of the DNR, Natural Heritage Program from *The Original Vegetation of Minnesota*, a map compiled in 1930 by F. J. Marschner from the U. S. General Land Office Survey Notes and published in 1974 under the direction of M. L. Heinselman of the U. S. Forest Service. It was produced by the Cartography Laboratory of the Department of Geography, University of Minnesota.

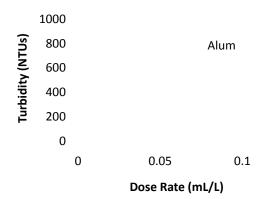
Published by the Natural Heritage Program, Minnesota Department of Natural Resources, 1988[©]

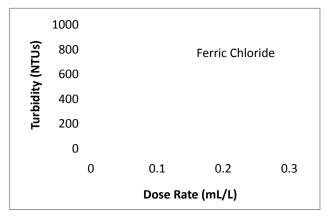
# **Appendix C – Flocculant Dose Testing Results**

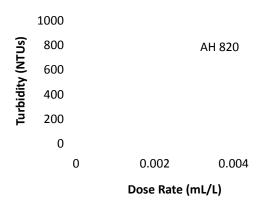


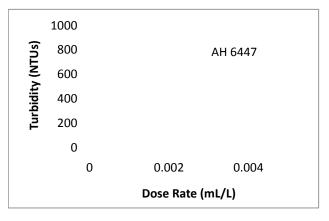
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Classification	Sandy Loam
USCS Symbol(s)	SM
Strata	Glacial Lake Sediment
Geomorphology	Des Moines Lobe

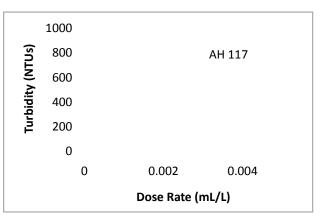


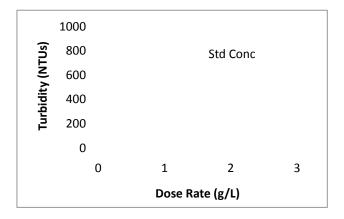




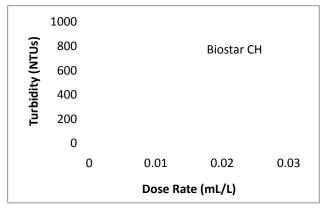


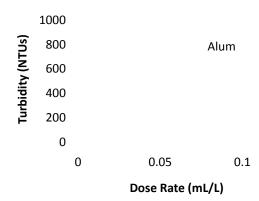


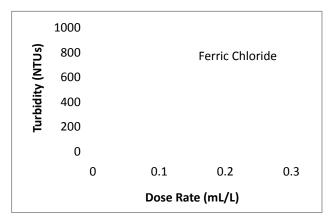


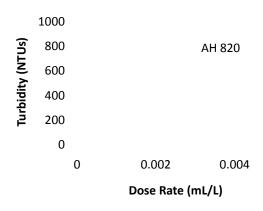


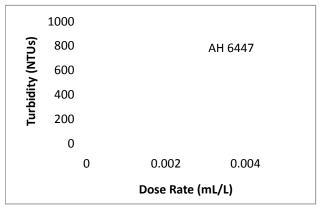
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USCS Symbol(s)	SM
Strata	Glacial Lake Sediment
Geomorphology	Des Moines Lobe

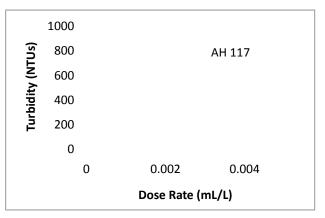


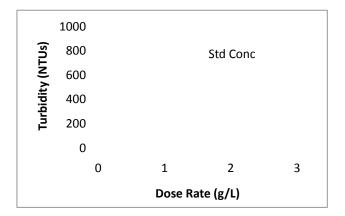




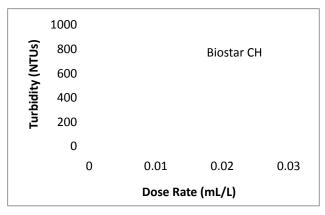


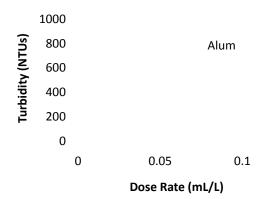


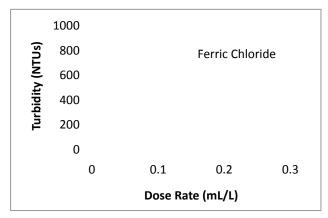


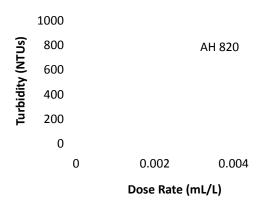


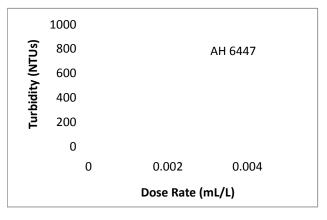
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Strata	Stagnation Moraine
Geomorphology	Wadena Lobe

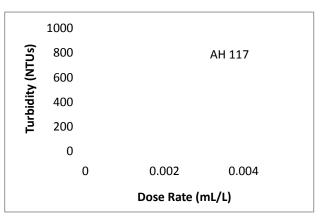


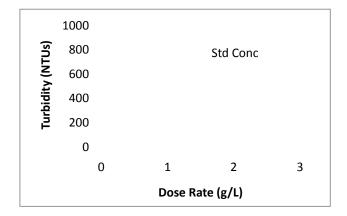




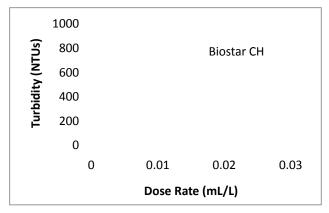


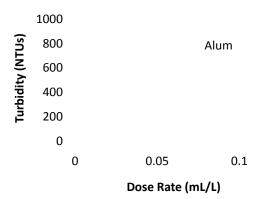


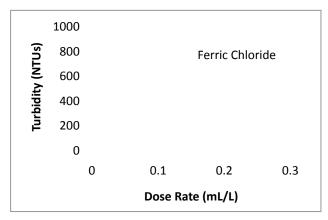


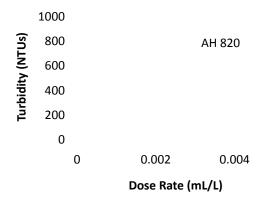


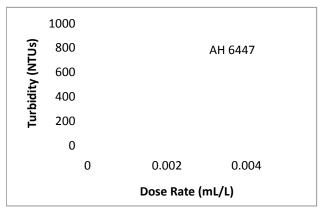
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Classification	Organic Sandy Clay Loam
USCS Symbol(s)	SC-SM
Strata	Stagnation Moraine
Geomorphology	Wadena Lobe

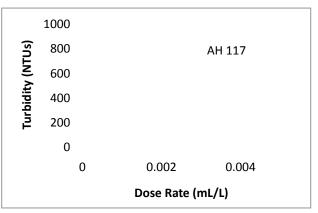


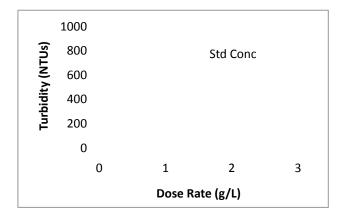




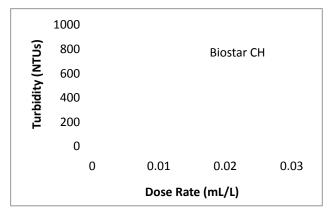


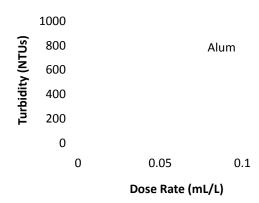


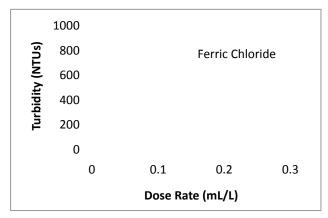


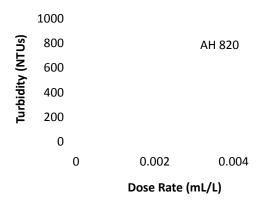


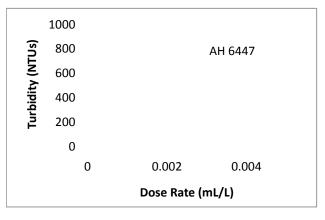
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Strata	Glacial Lake Sediment	
Geomorphology Superior Lobe		

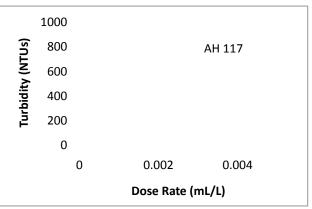


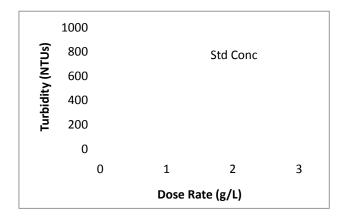




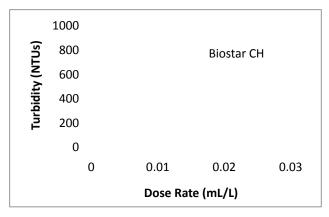


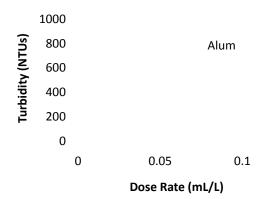


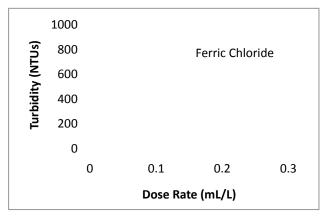


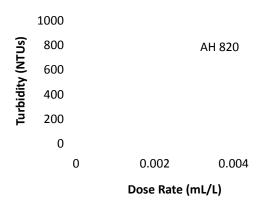


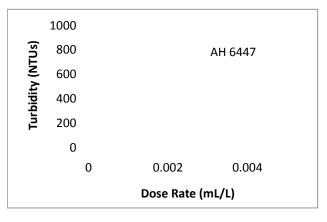
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Classification	Sandy Loam	
USCS Symbol(s)	SM	
Strata	Glacial Lake Sediment	
Geomorphology Superior Lobe		

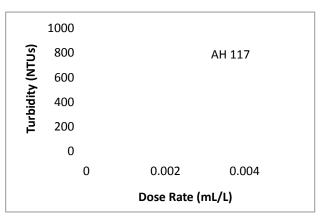


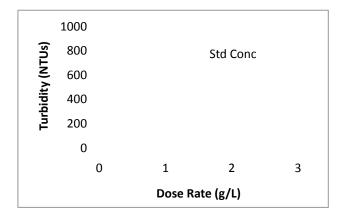




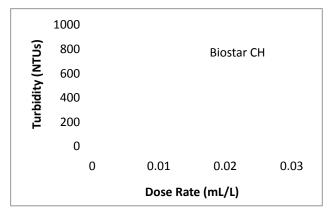


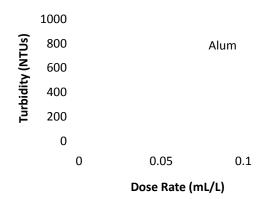


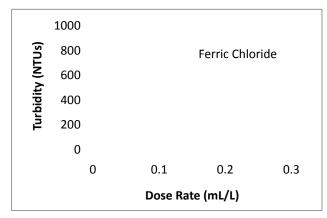


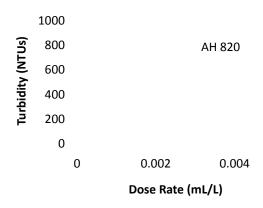


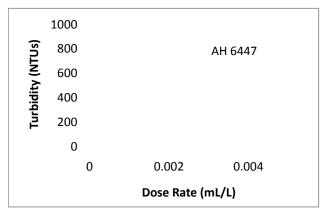
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Classification	Sandy Loam
USCS Symbol(s)	SM
Strata	Ground Moraine
Geomorphology	Superior Lobe

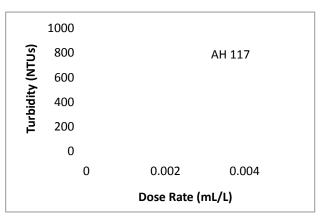


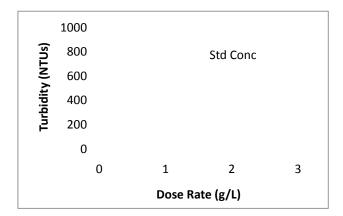




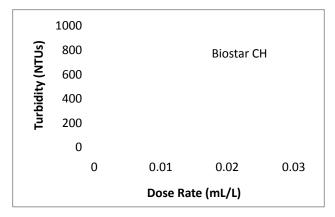


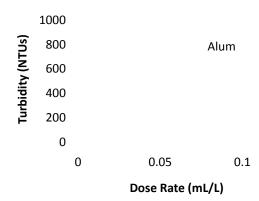


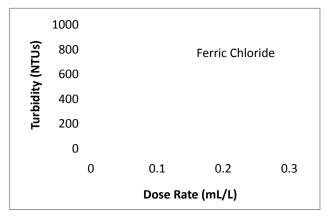


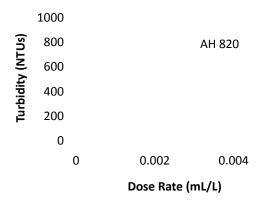


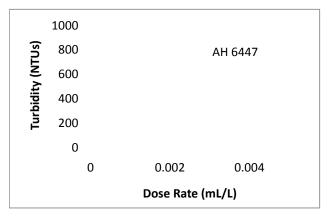
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Classification	Sandy Loam
USCS Symbol(s)	SM
Strata	Ground Moraine
Geomorphology	Superior Lobe

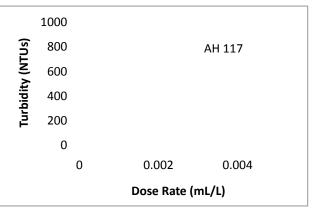


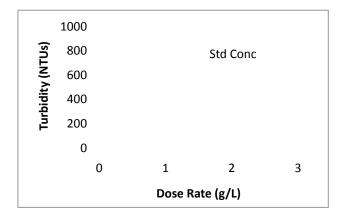




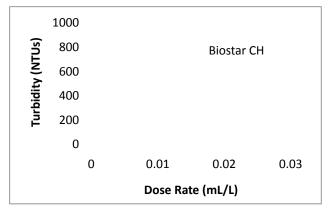


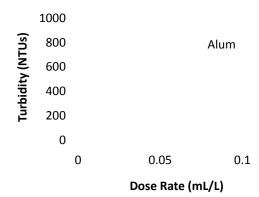


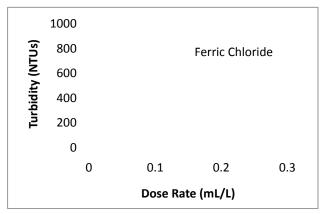


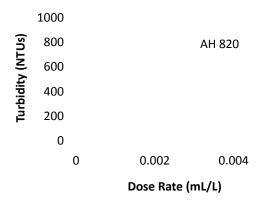


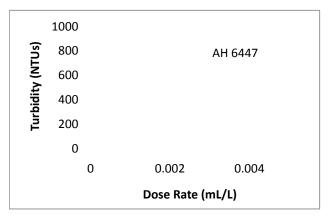
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USCS Symbol(s)	SM
Strata	Outwash
Geomorphology	Des Moines Lobe

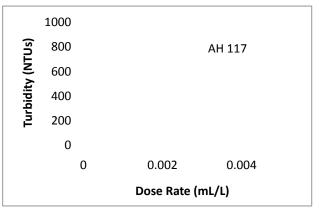


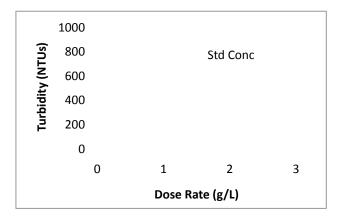




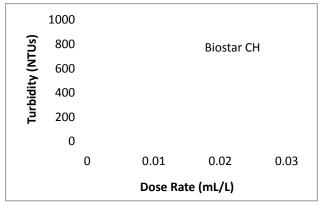


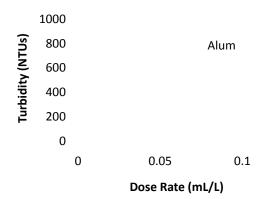


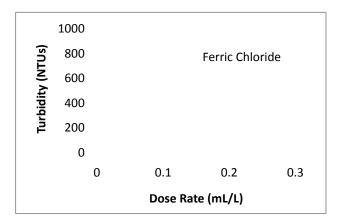


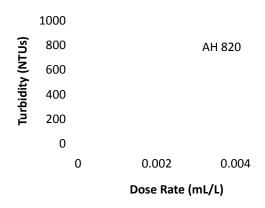


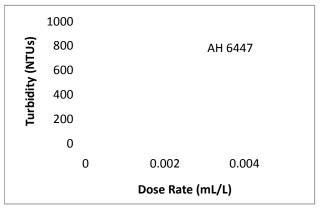
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Classification	Slightly Organic Sandy Loam
USCS Symbol(s)	SM
Strata	Outwash
Geomorphology	Des Moines Lobe

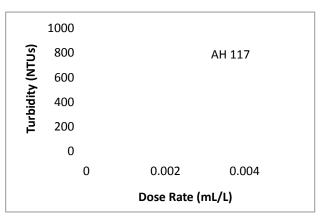


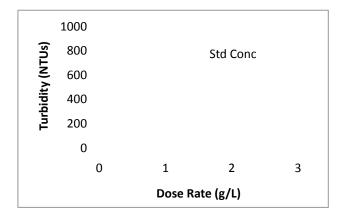




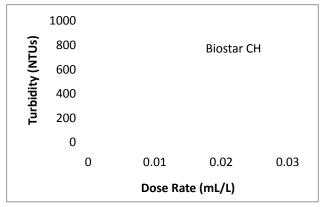


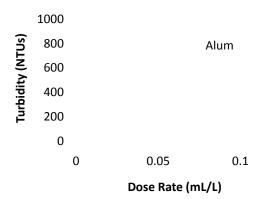


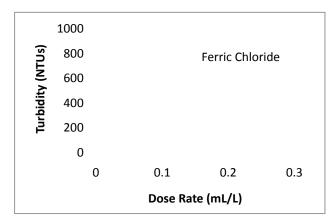


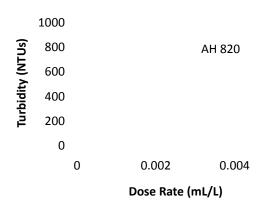


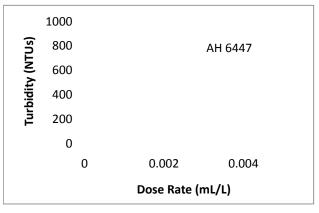
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Strata	Glacial Lake Sediment
Geomorphology	Des Moines Lobe

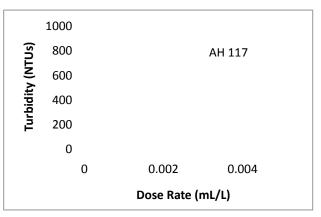


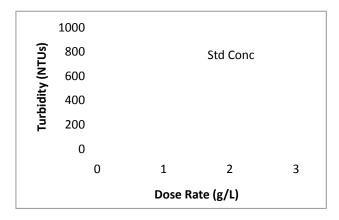




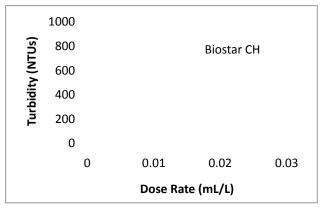


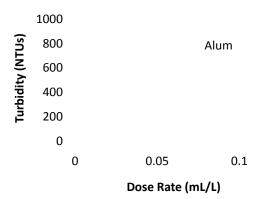


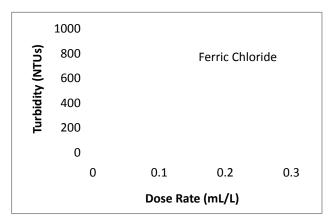


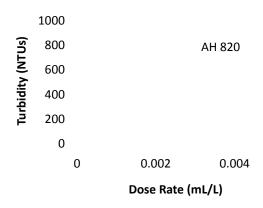


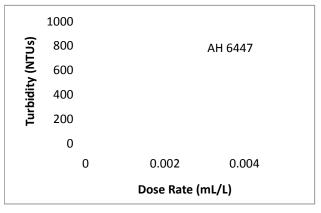
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Classification	Organic Clay Loam
USCS Symbol(s)	OL
Strata	Glacial Lake Sediment
Geomorphology	Des Moines Lobe

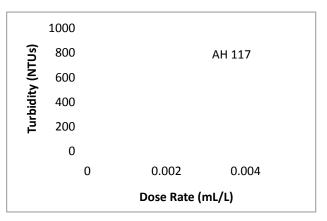


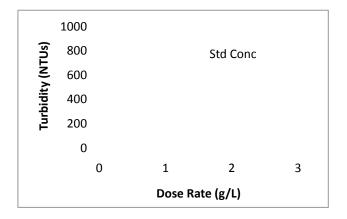




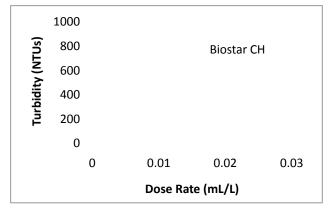


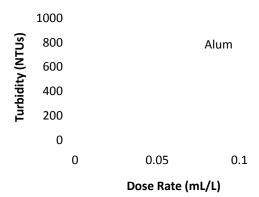


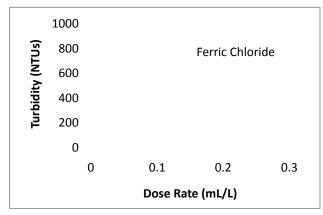


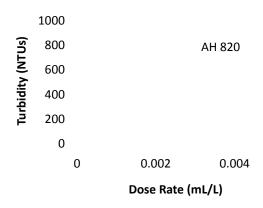


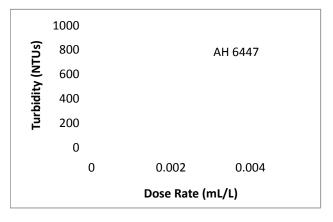
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Classification	Sandy Loam
USCS Symbol(s)	SM
Strata	Ground Moraine
Geomorphology	Superior Lobe

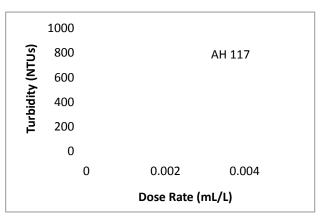


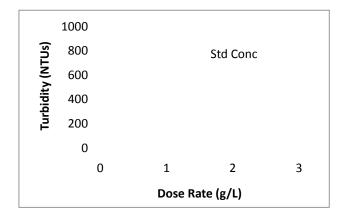




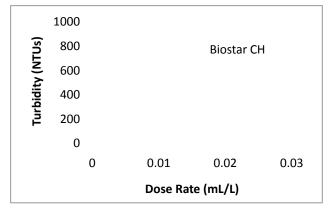


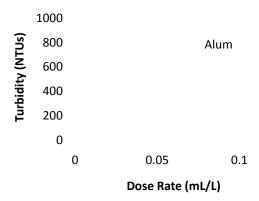


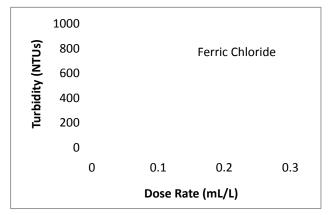


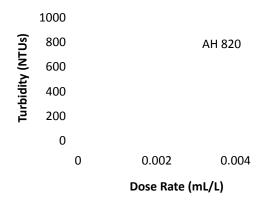


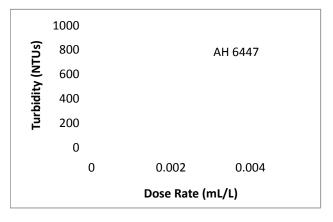
Material	Foley Topsoil
Classification	Slightly Organic Sandy Loam
USCS Symbol(s)	SM
Strata	Ground Moraine
Geomorphology	Superior Lobe

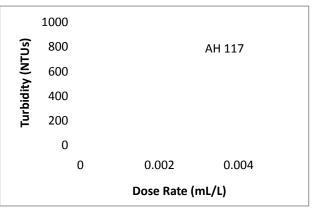


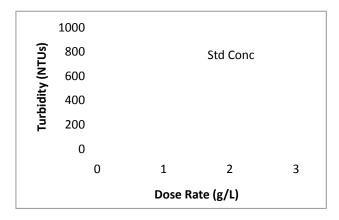




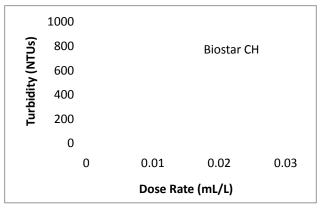


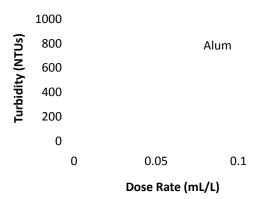


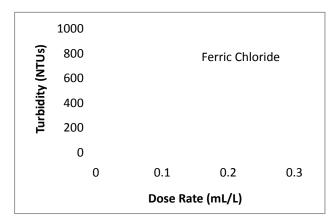


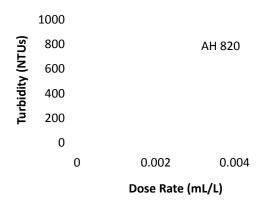


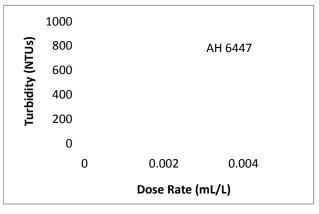
Material	Grand Rapids Subsoil
Classification	Sandy Loam
USCS Symbol(s)	SM
Strata	Outwash
Geomorphology	Des Moines Lobe

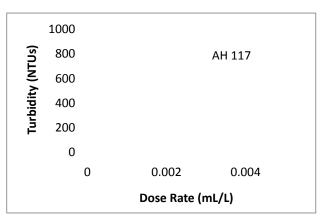


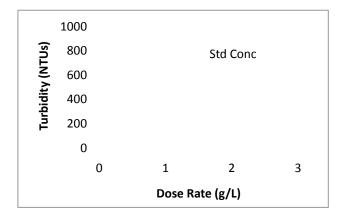




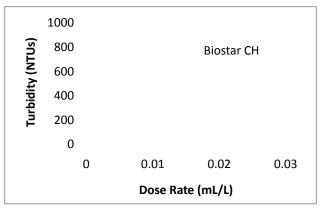


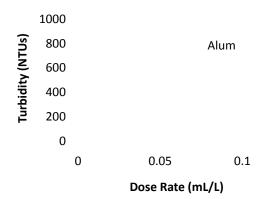


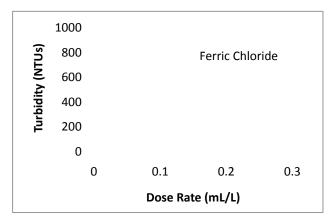


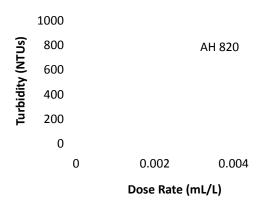


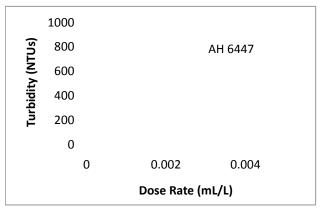
Material	Grand Rapids Topsoil
Classification	Sandy Loam
USCS Symbol(s)	SM
Strata	Outwash
Geomorphology	Des Moines Lobe

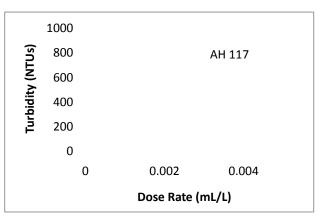






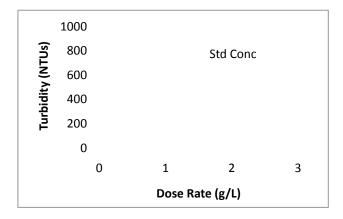




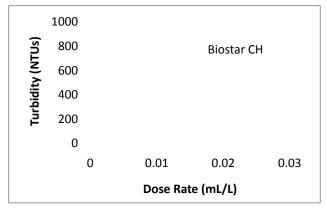


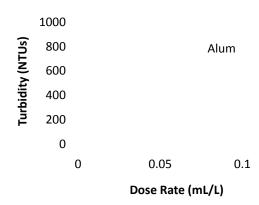
MnDOT Floc

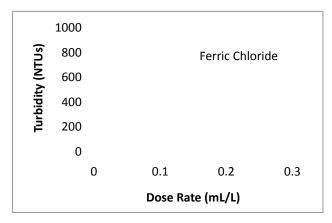
April 5, 2013

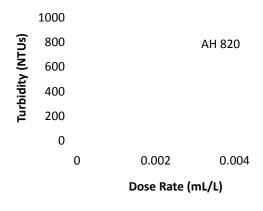


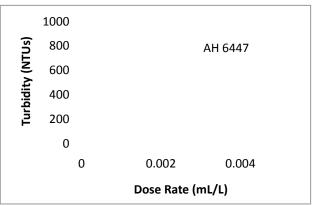
Material	Hampton Subsoil
Classification	Slightly Organic Sandy Loam
USCS Symbol(s)	SM
Strata	Red Drift
Geomorphology	Illinoisian Glaciation

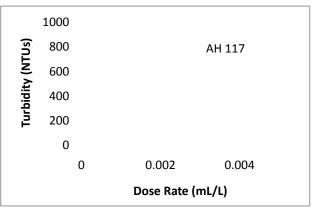


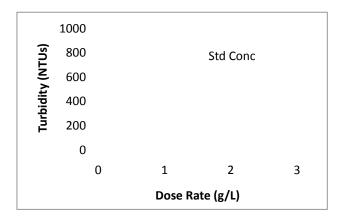




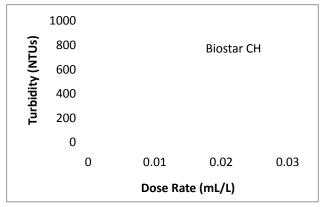


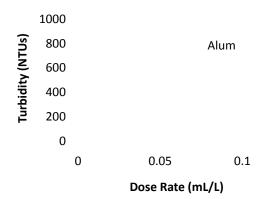


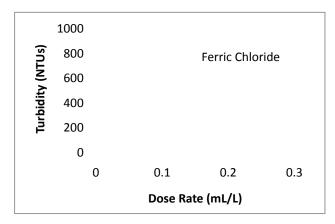


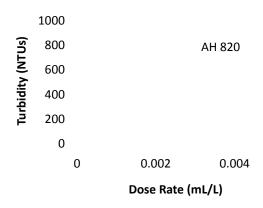


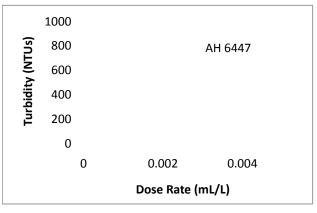
Material	Hampton Topsoil
Classification	Organic Sandy Loam
USCS Symbol(s)	SM
Strata	Red Drift
Geomorphology	Illinoisian Glaciation

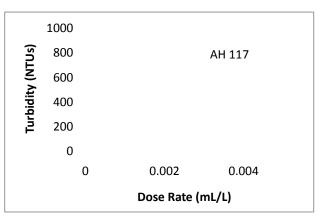


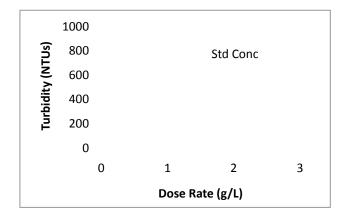




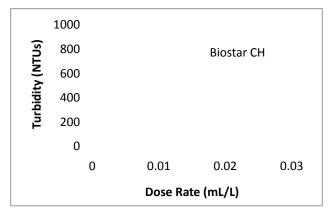


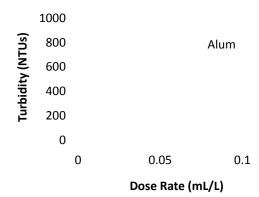


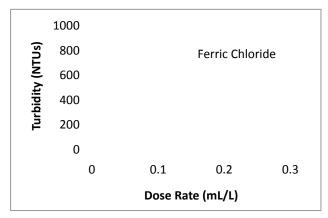


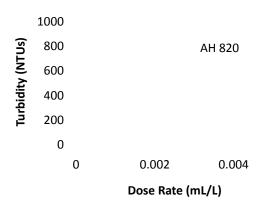


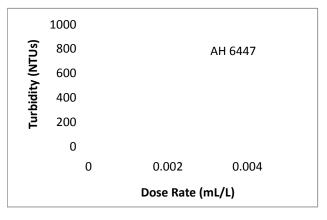
Material	Houston Riverbed
Classification	Sandy Loam
USCS Symbol(s)	SM
Strata	Colluvuum
Geomorphology	Weathered Residual

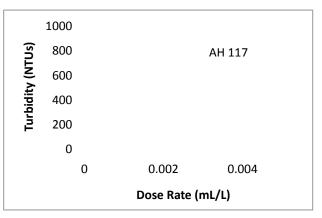


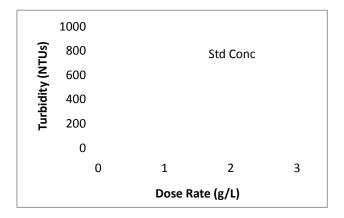




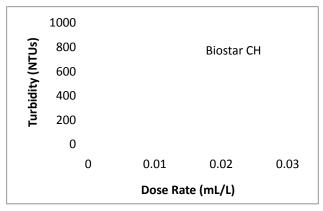


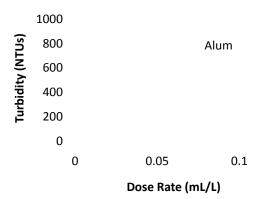


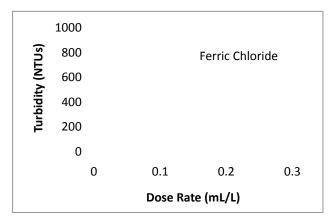


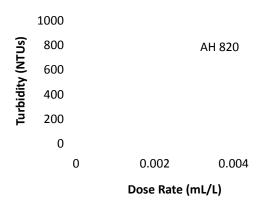


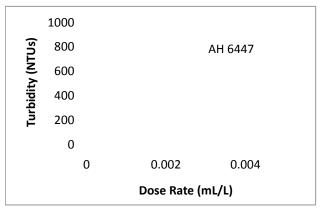
Material	Houston Topsoil
Classification	Organic Sandy Loam
USCS Symbol(s)	SM
Strata	Colluvuum
Geomorphology	Weathered Residual

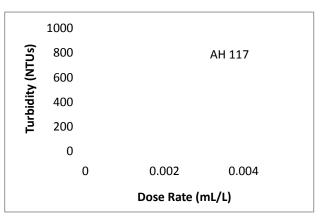


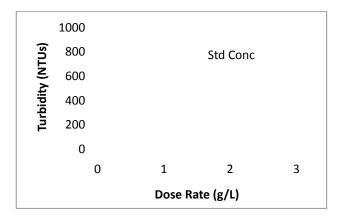




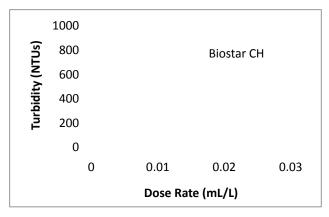


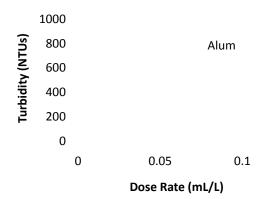


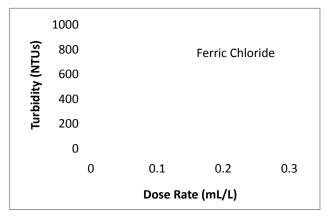


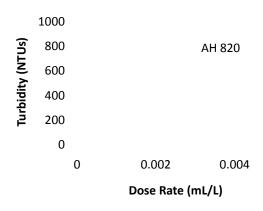


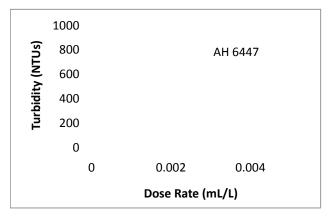
Material	Kandiyohi Subsoil
Classification	Sandy Loam
USCS Symbol(s)	SM
Strata	Stagnation Moraine
Geomorphology	Wadena Lobe

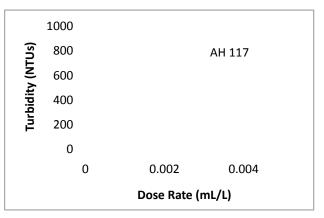


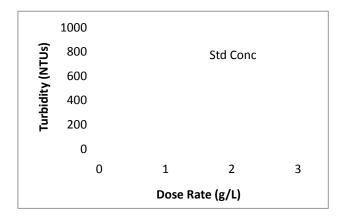




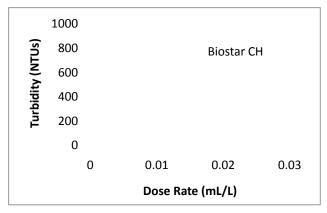


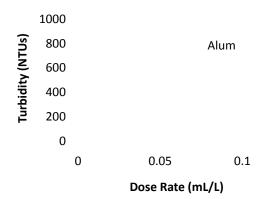


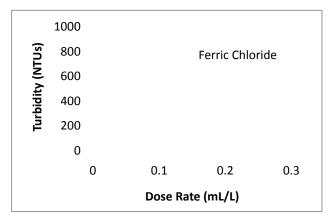


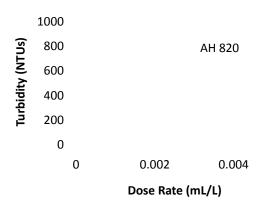


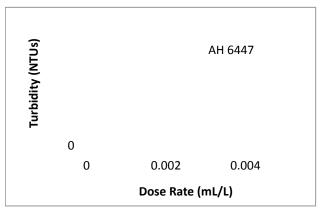
Material	Kandiyohi Topsoil
Classification	Organic Clay Loam
USCS Symbol(s)	OL
Strata	Stagnation Moraine
Juata	Jugnation Moralic

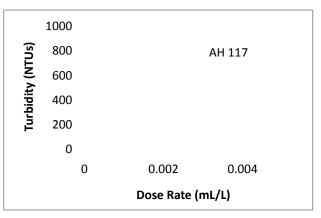


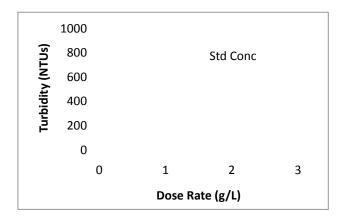




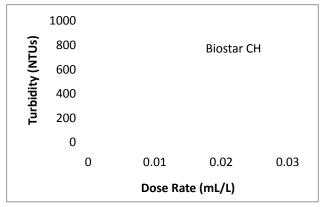


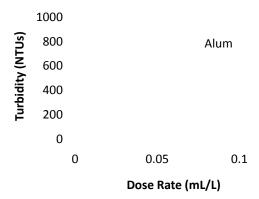


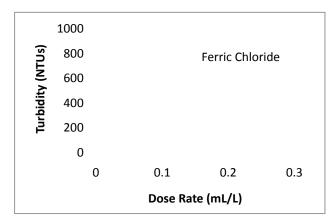


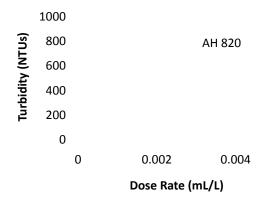


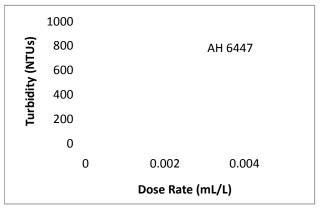
Material	Lakeville Subsoil
Classification	Slightly Organic Sandy Loam
USCS Symbol(s)	SM
Strata	Gray Drift
Geomorphology	Kansan Glaciation



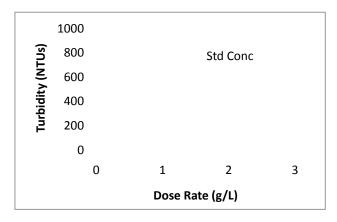




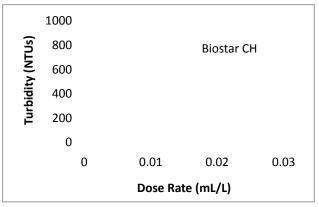


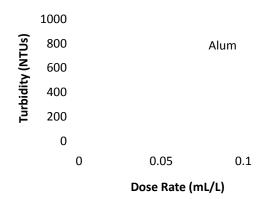


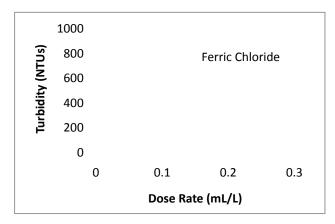


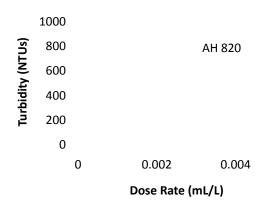


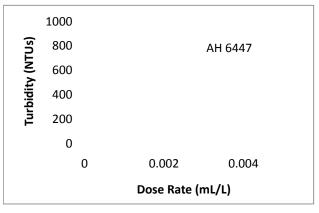
Material	Lakeville Topsoil
Classification	Slightly Organic Clay Loam
USCS Symbol(s)	ML-OL
Strata	Gray Drift
Geomorphology	Kansan Glaciation

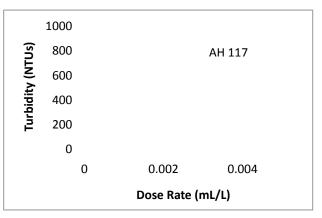


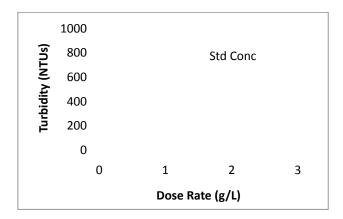




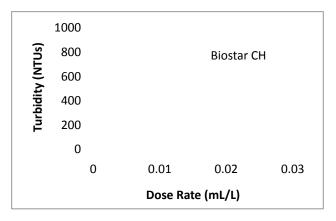


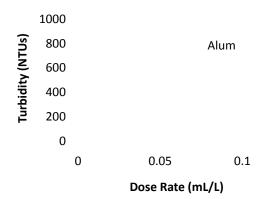


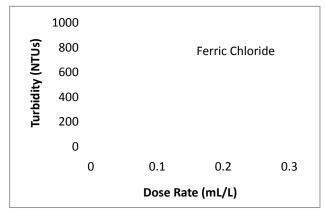


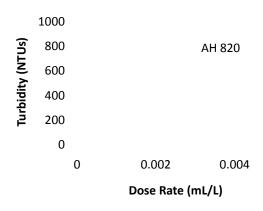


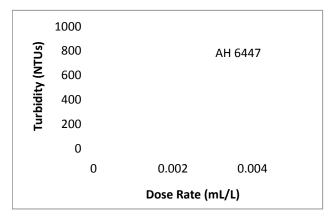
Material	Lindstrom Subsoil
Classification	Sandy Loam
USCS Symbol(s)	SM
Strata	Ground Moraine
Geomorphology	Des Moines Lobe

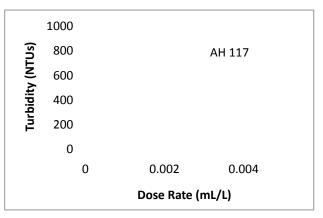


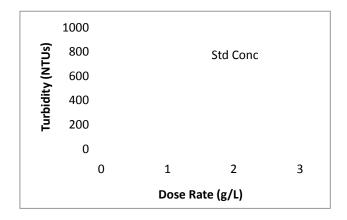




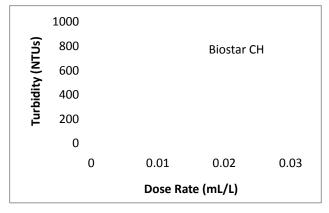


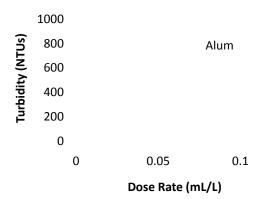


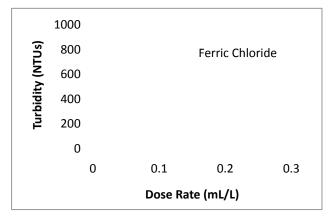


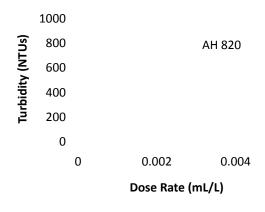


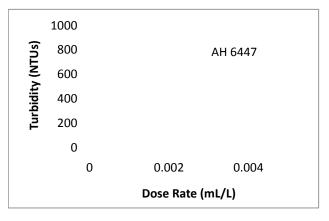
Material	Lindstrom Topsoil
Classification	Slightly Organic Sandy Loam
USCS Symbol(s)	SM
Strata	Ground Moraine
Geomorphology	Des Moines Lobe

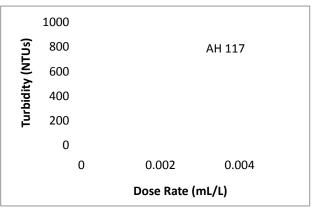


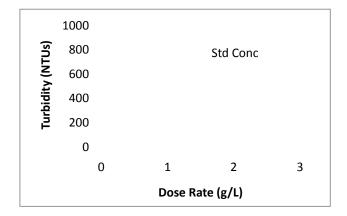




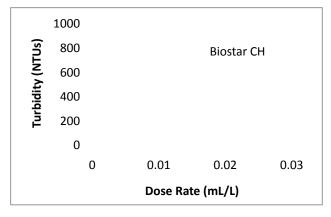


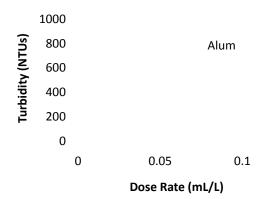


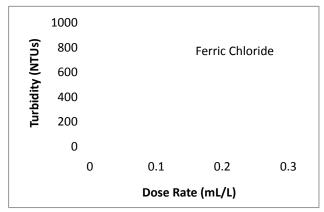


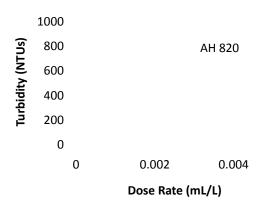


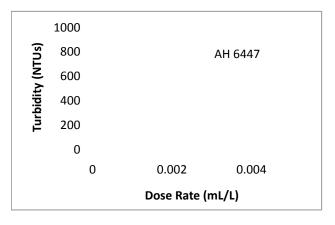
Material	Mankato Topsoil
Classification	Sandy Loam
USCS Symbol(s)	SM
Strata	Glacial Lake/Alluvium
Geomorphology	Des Moines Lobe

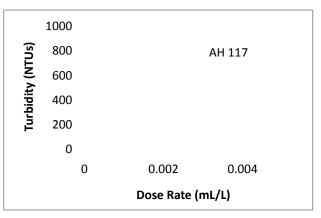


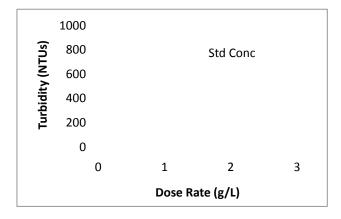




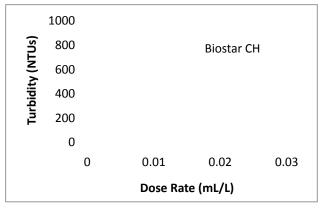


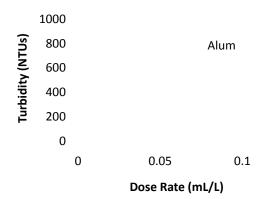


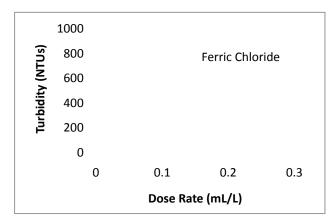


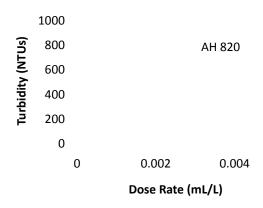


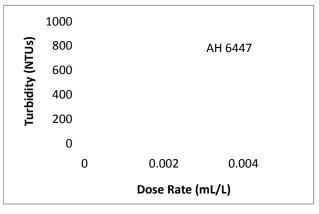
Material	Mapleton Topsoil
Classification	Slightly Organic Sandy Clay Loam
USCS Symbol(s)	SC-SM
Strata	Glacial Lake Sediment
Geomorphology	Des Moines Lobe

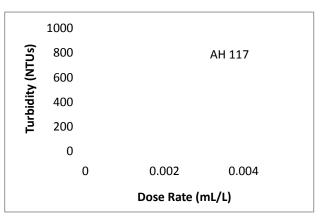


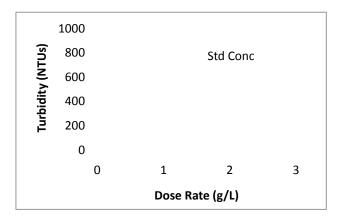




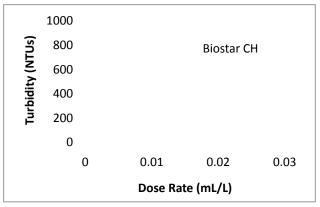


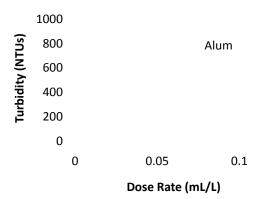


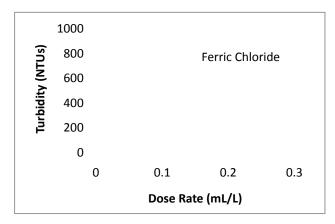


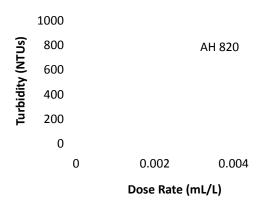


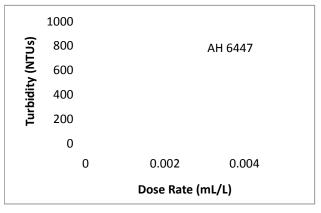
Material	Moorhead Subsoil
Classification	Slightly Organic Clay Loam
USCS Symbol(s)	CL-ML-OL
Strata	Glacial Lake Sediment
Geomorphology	Des Moines Lobe

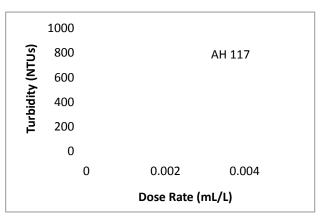


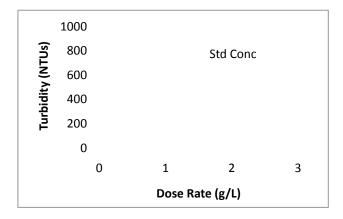




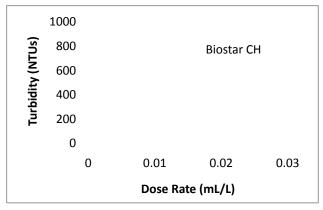


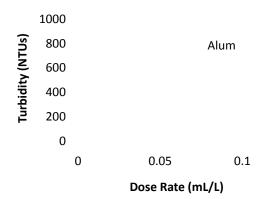


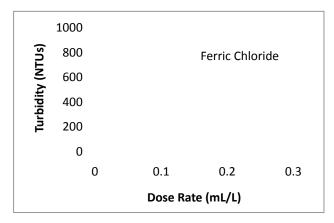


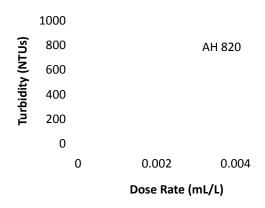


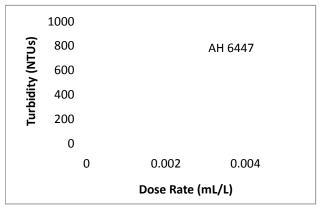
Material	Moorhead Topsoil
Classification	Organic Clay Loam
USCS Symbol(s)	OL
Strata	Glacial Lake Sediment
Geomorphology	Des Moines Lobe

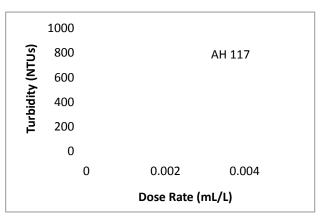


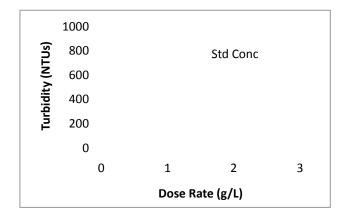




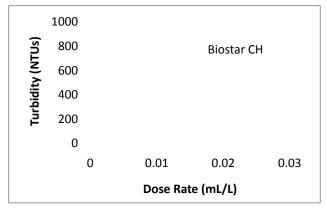


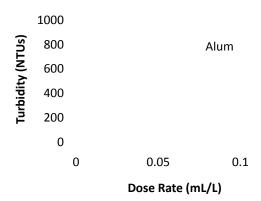


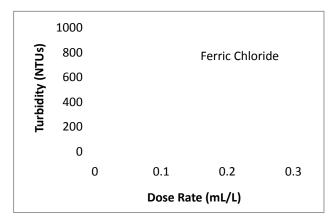


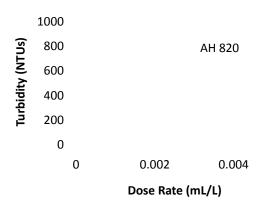


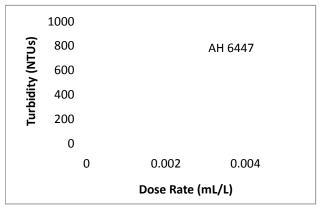
Material	New Sweden Topsoil
Classification	Sandy Loam
USCS Symbol(s)	SM
Strata	Stagnation Moraine
Geomorphology	Des Moines Lobe

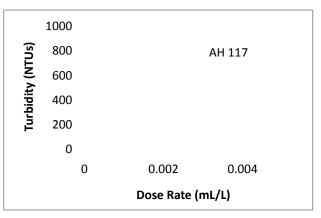


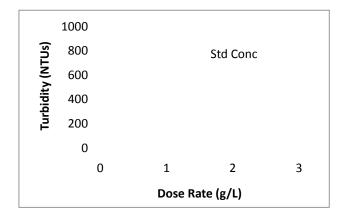




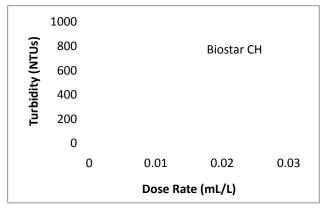


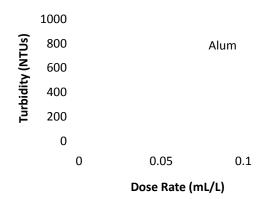


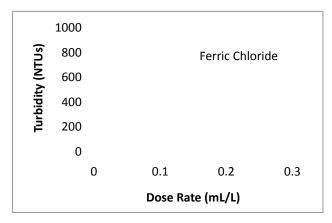


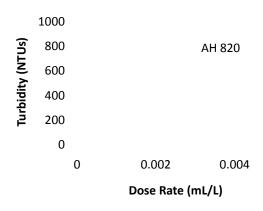


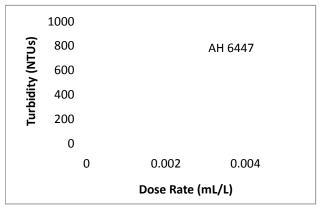
Material	North Mankato Subsoil
Classification	Sandy Clay Loam
USCS Symbol(s)	SC-SM
Strata	Ground Moraine
Geomorphology	Des Moines Lobe

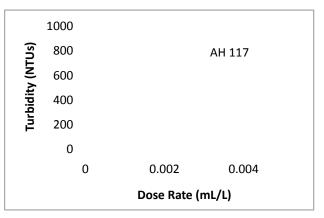


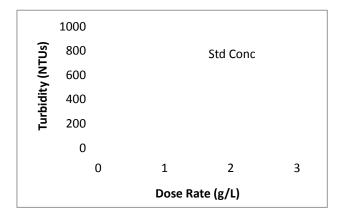




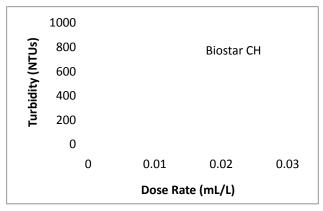


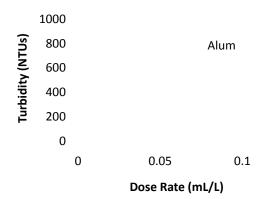


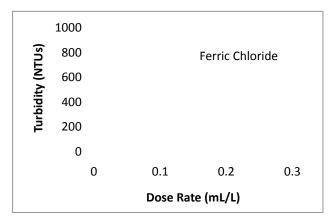


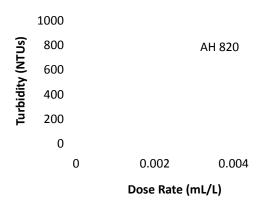


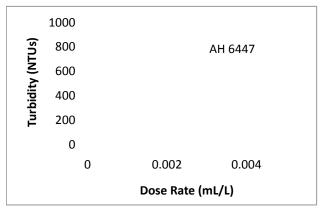
Material	North Mankato Topsoil
Classification	Organic Sandy Clay Loam
USCS Symbol(s)	SC-SM
Strata	Ground Moraine
Geomorphology	Des Moines Lobe

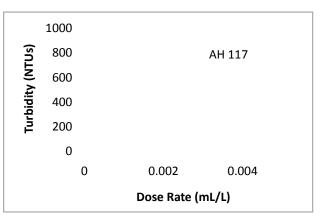


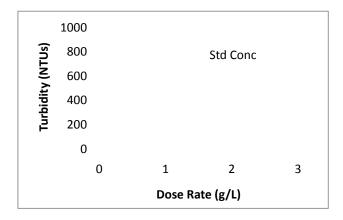




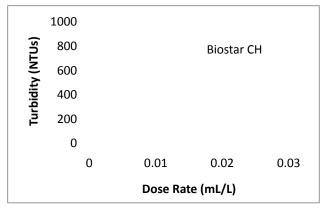


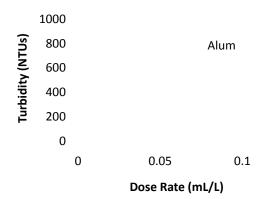


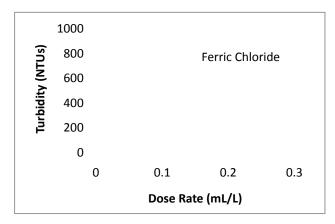


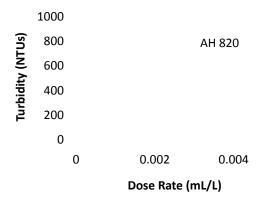


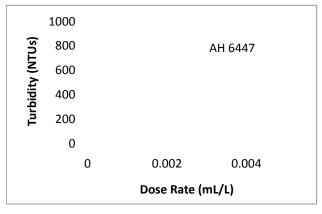
Material	Olivia A Subsoil
Classification	Sandy Loam
USCS Symbol(s)	SM
Strata	Ground Moraine
Geomorphology	Des Moines Lobe

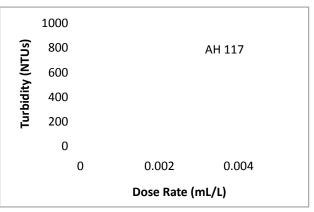


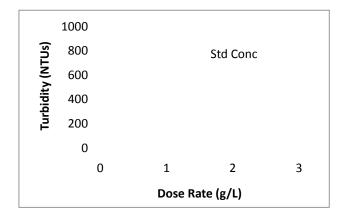




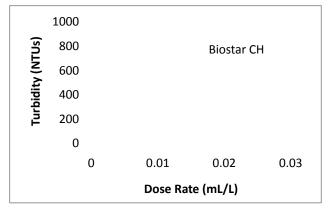


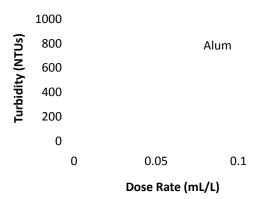


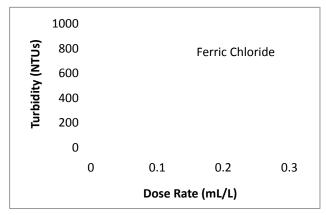


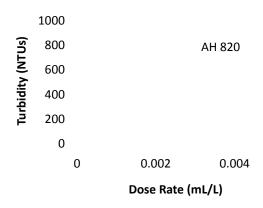


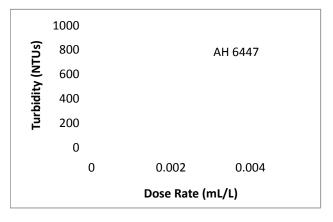
Material	Olivia A Topsoil
Classification	Organic Sandy Loam
USCS Symbol(s)	SM
Strata	Ground Moraine
Geomorphology	Des Moines Lobe

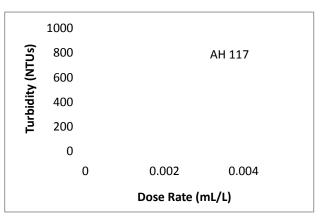


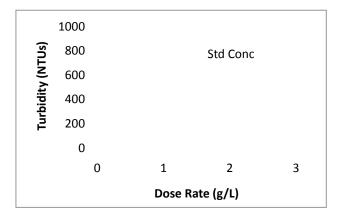




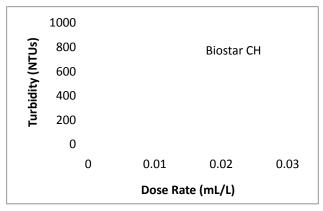


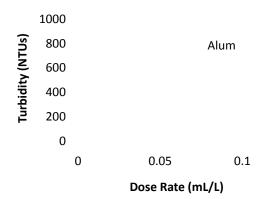


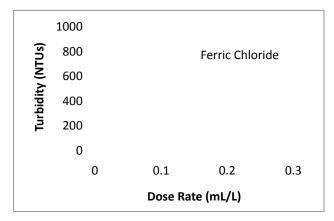


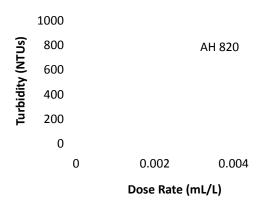


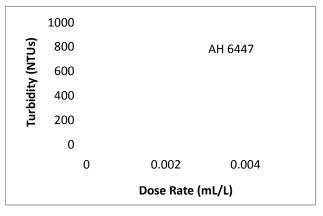
Material	Olivia B Clay
Classification	Sandy Clay Loam
USCS Symbol(s)	SC-SM
Strata	Ground Moraine
Geomorphology	Des Moines Lobe

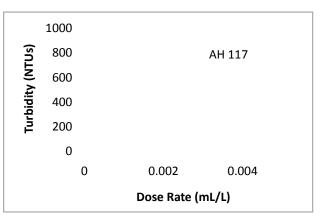


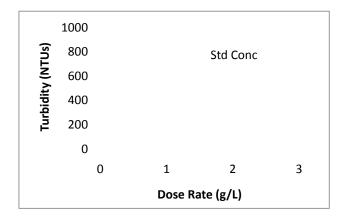




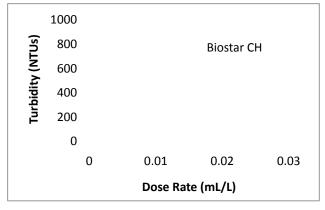


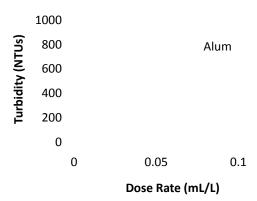


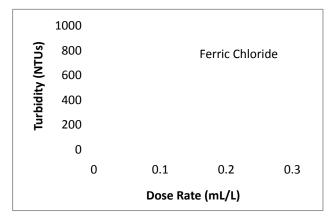


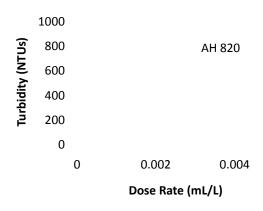


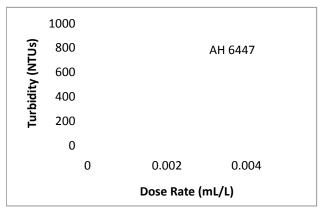
Material	Owatonna Clay
Classification	Slightly Organic Sandy Clay Loam
USCS Symbol(s)	SC-SM
Strata	Ground Moraine
Geomorphology	Des Moines Lobe

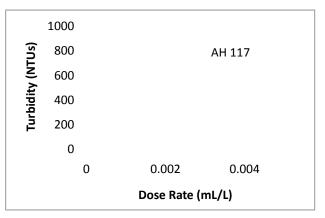


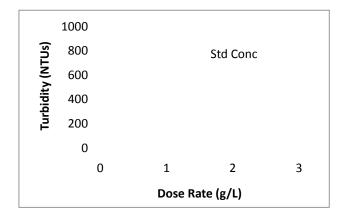




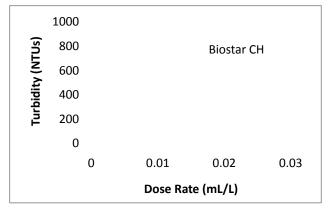


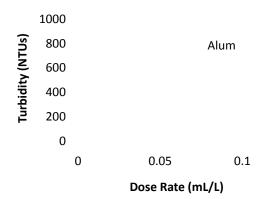


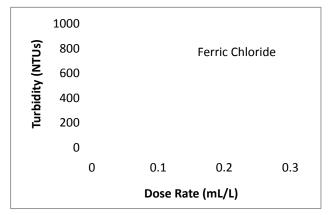


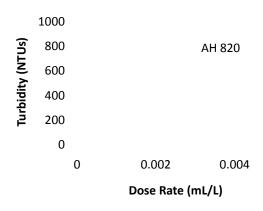


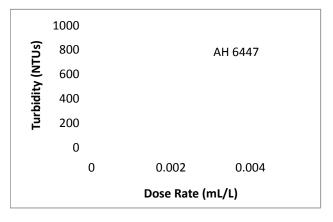
Material	Owatonna Topsoil
Classification	Slightly Organic Sandy Clay Loam
USCS Symbol(s)	SC-SM
Strata	Ground Moraine
Geomorphology	Des Moines Lobe

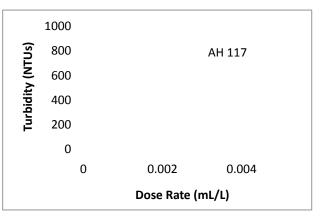


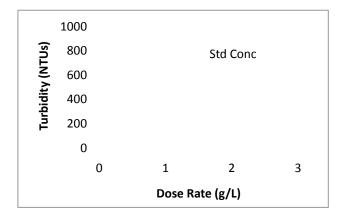




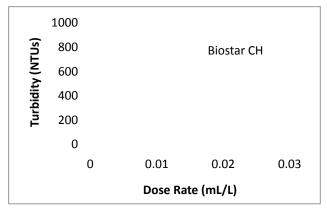


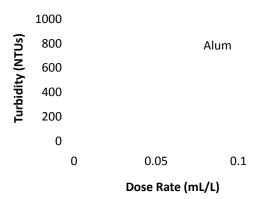


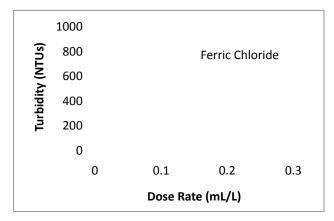


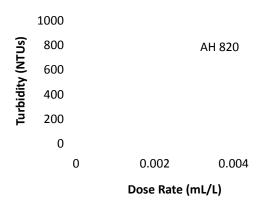


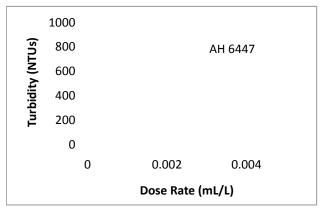
Material	Perham Subsoil
Classification	Sandy Loam
USCS Symbol(s)	SM
Strata	Outwash
Geomorphology	Des Moines Lobe

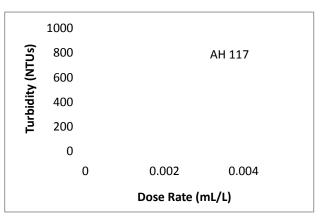


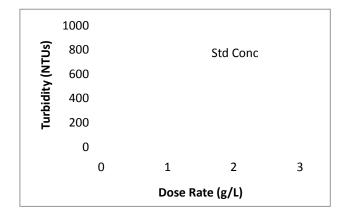




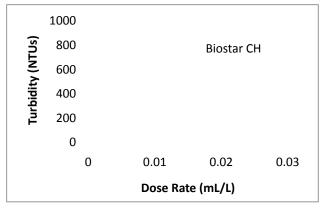


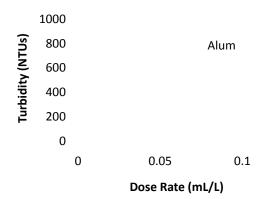


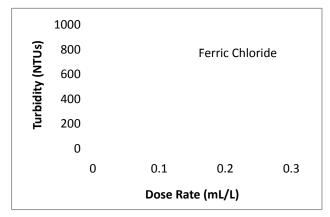


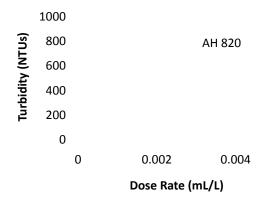


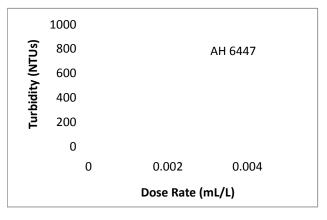
Material	Perham Topsoil
Classification	Slightly Organic Sandy Loam
USCS Symbol(s)	SM
Strata	Outwash
Geomorphology	Des Moines Lobe

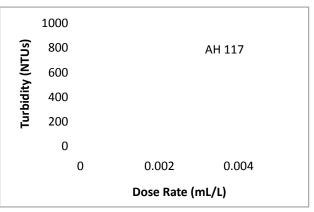


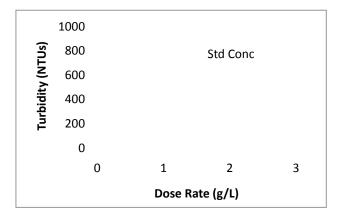




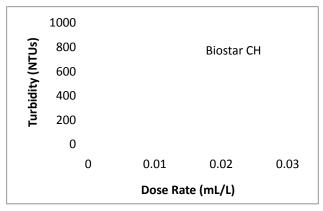


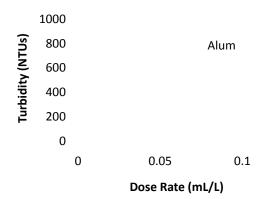


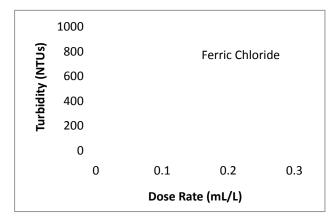


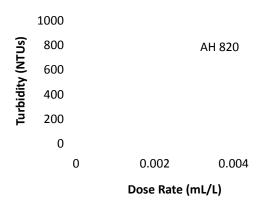


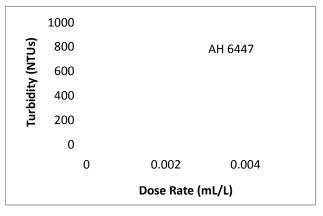
Material	Pipestone Subsoil
Classification	Sandy Loam
USCS Symbol(s)	SM
Strata	Gray Drift
Geomorphology	Kansan Glaciation

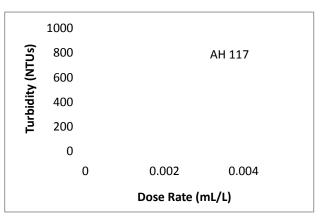


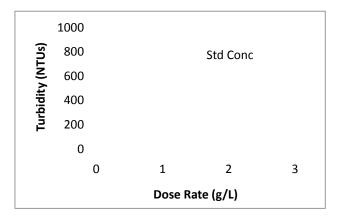




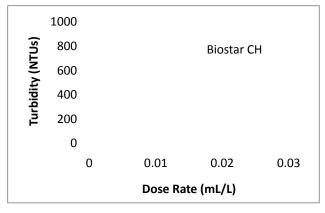


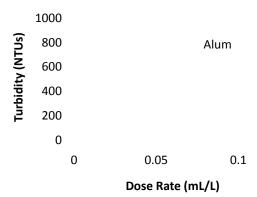


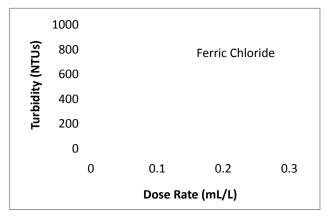


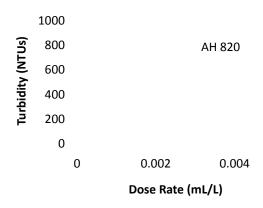


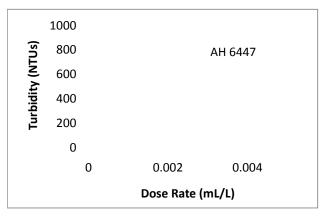
Material	Pipestone Topsoil
Classification	Organic Sandy Loam
USCS Symbol(s)	SM
Strata	Gray Drift
Geomorphology	Kansan Glaciation

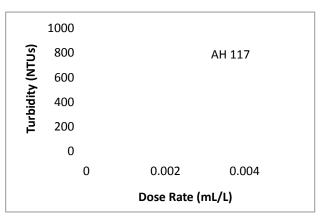


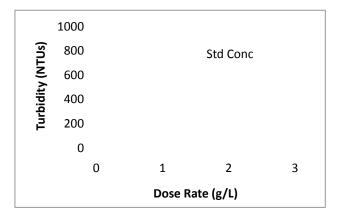




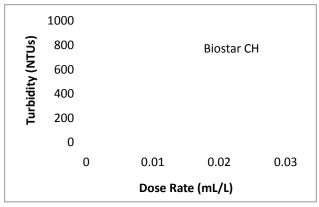


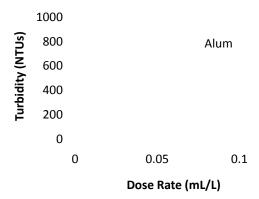


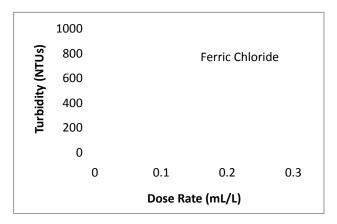


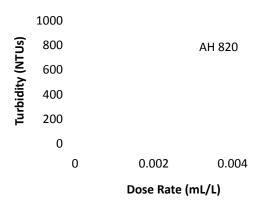


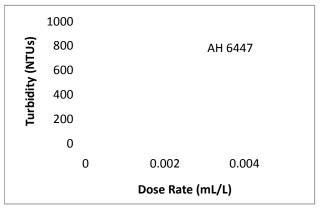
Material	Ramsey Peat
Classification	Highly Organic Sandy Loam
USCS Symbol(s)	SM
Strata	End Moraine
Geomorphology	Des Moines Lobe

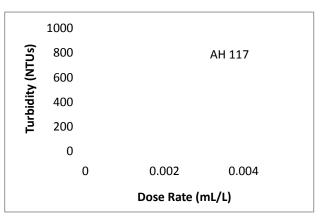


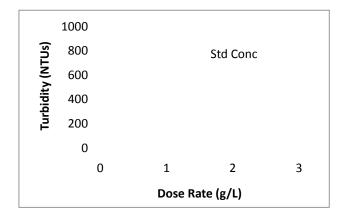




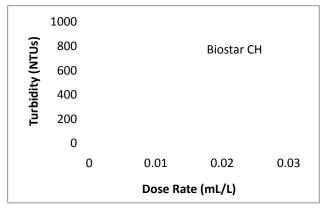


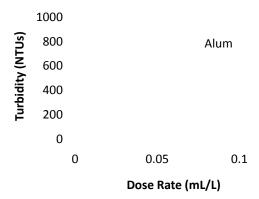


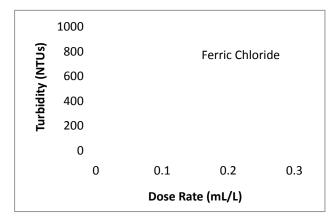


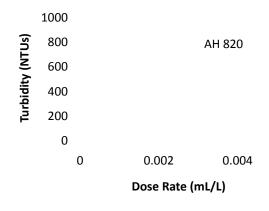


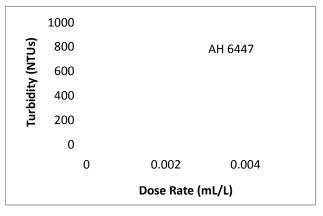
Material	Rockford Cornfield
Classification	Slightly Organic Sandy Loam
USCS Symbol(s)	SM
Strata	End Moraine
Geomorphology	Des Moines Lobe

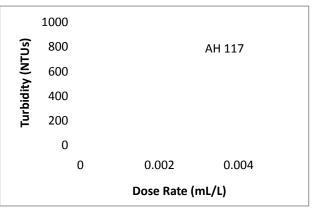


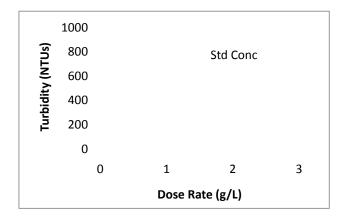




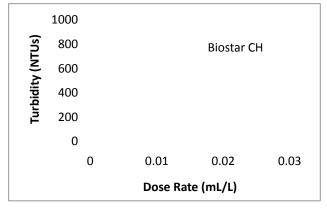


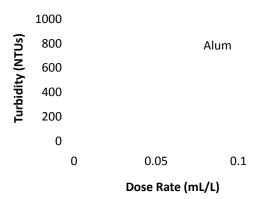


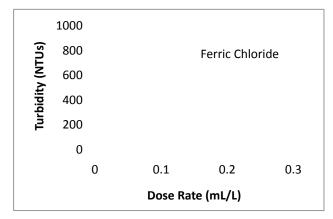


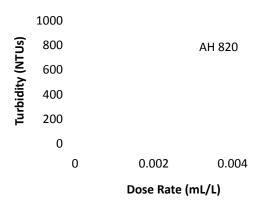


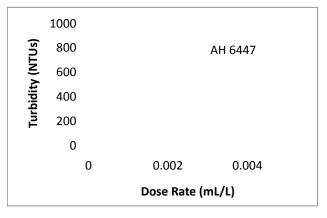
Material	Rockford Subsoil
Classification	Slightly Organic Sandy Loam
USCS Symbol(s)	SM
Strata	End Moraine
Geomorphology	Des Moines Lobe

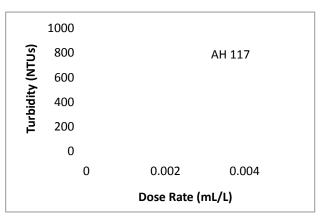


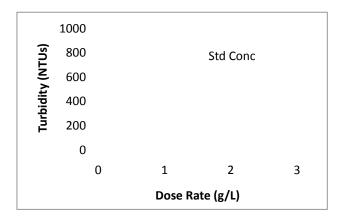




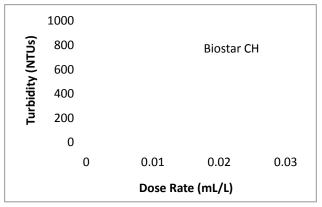


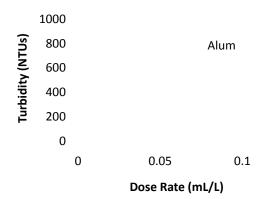


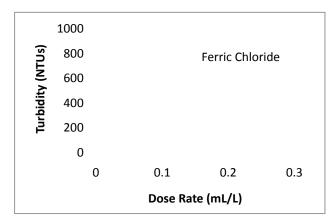


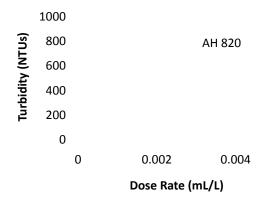


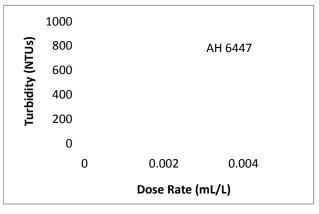
Material	Rockford Topsoil
Classification	Slightly Organic Sandy Loam
USCS Symbol(s)	SM
Strata	End Moraine
Geomorphology	Des Moines Lobe

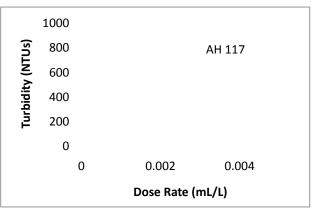


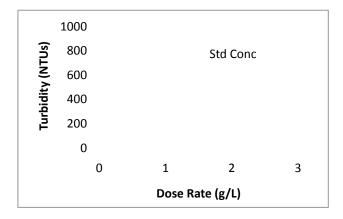




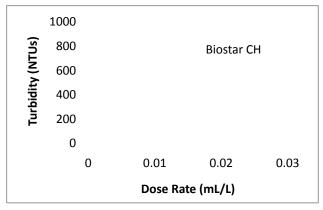


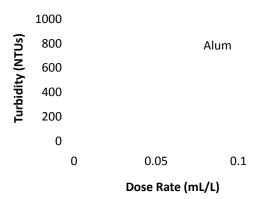


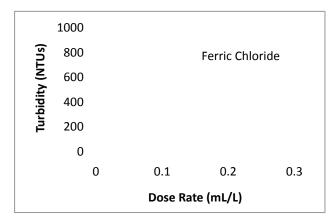


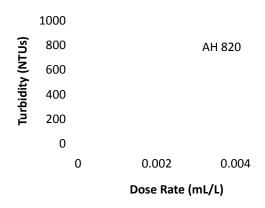


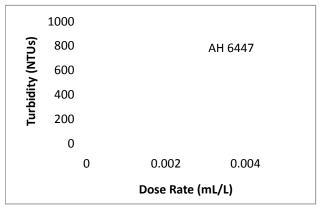
Material	Sandstone Subsoil
Classification	Slightly Organic Clay Loam
USCS Symbol(s)	CL-ML-OL
Strata	Ground Moraine
Geomorphology	Superior Lobe

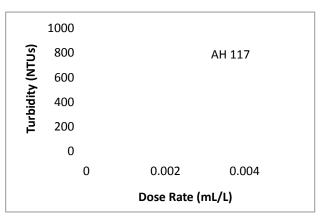


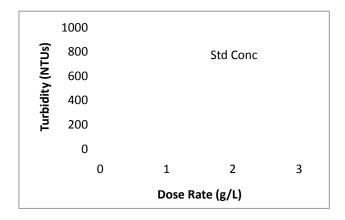




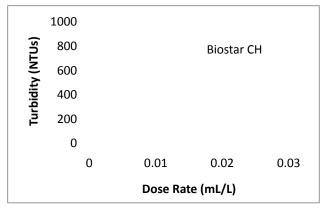


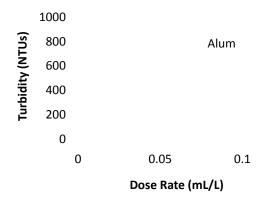


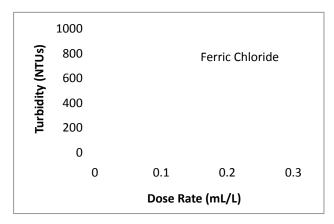


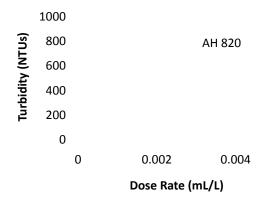


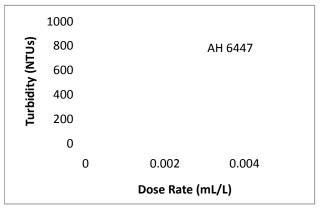
Material	Sandstone Topsoil
Classification	Slightly Organic Sandy Loam
USCS Symbol(s)	SM
Strata	Ground Moraine
Geomorphology	Superior Lobe



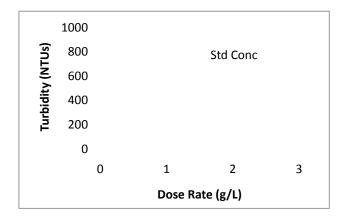




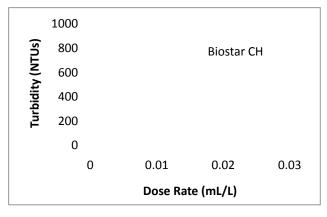


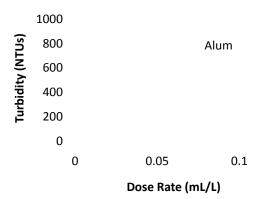


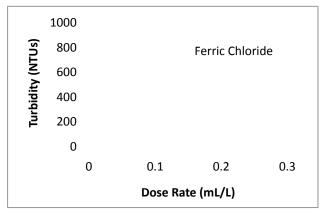


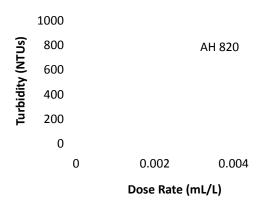


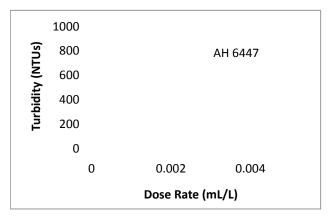
Material	Stillwater Sediment
Classification	Organic Loam
USCS Symbol(s)	OL
Strata	End Moraine
Geomorphology	Superior Lobe

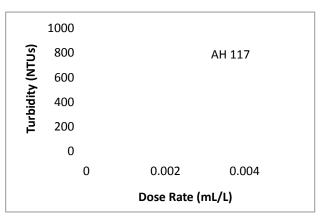






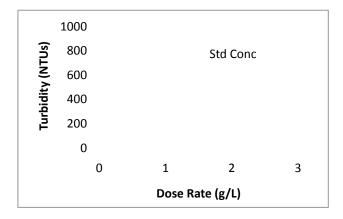




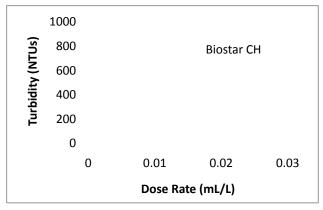


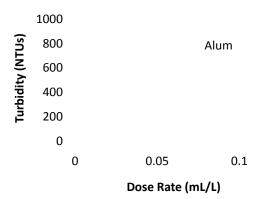
MnDOT Floc

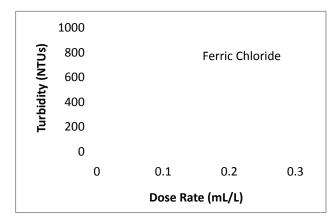
April 5, 2013

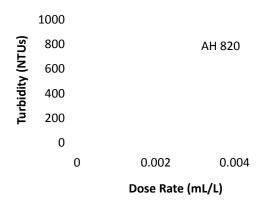


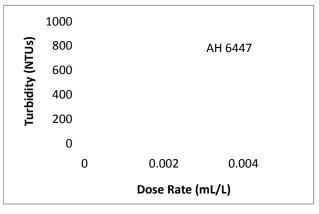
Material	Tamarack Clay
Classification	Slightly Organic Clay
USCS Symbol(s)	CL-OL
Strata	End Moraine
Geomorphology	Des Moines Lobe

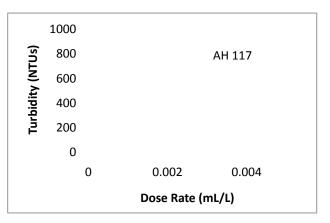


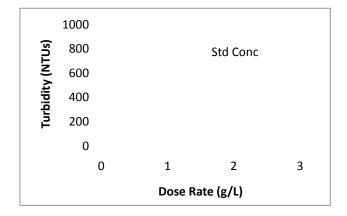




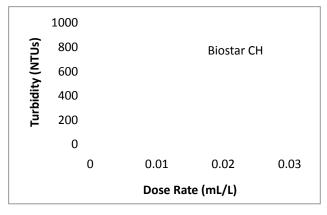


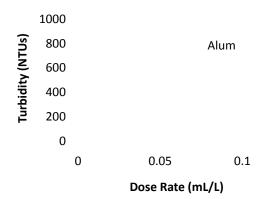


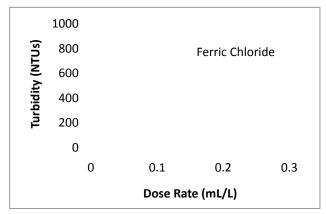


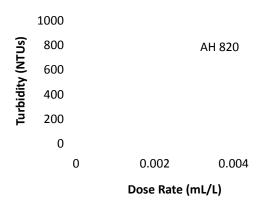


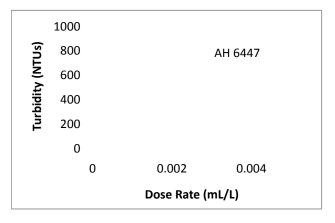
Material	Tamarack Peat
Classification	Peat
USCS Symbol(s)	PT
Strata	Peat
Geomorphology	Des Moines Lobe

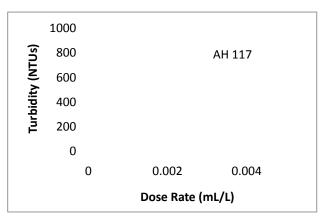


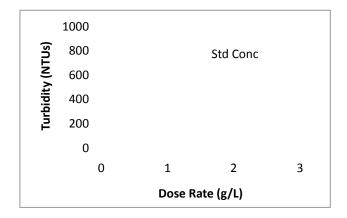




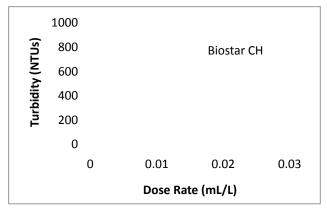


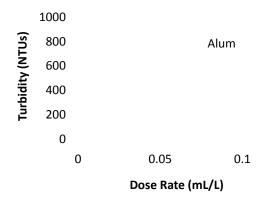


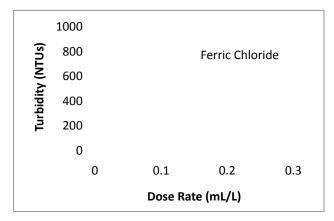


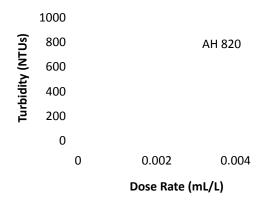


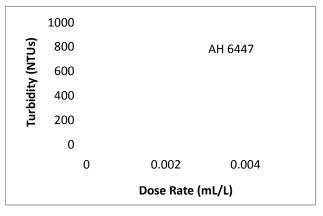
Material	Tamarack Topsoil
Classification	Slightly Organic Sandy Loam
USCS Symbol(s)	SM
Strata	End Moraine
Geomorphology	Des Moines Lobe

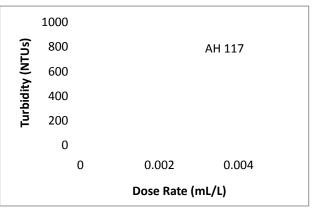


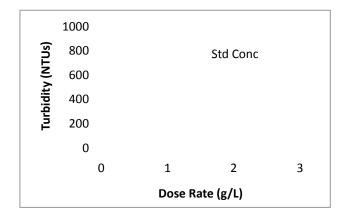




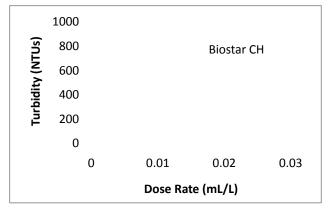


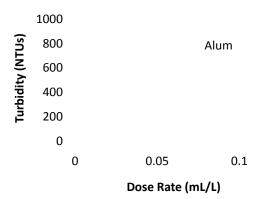


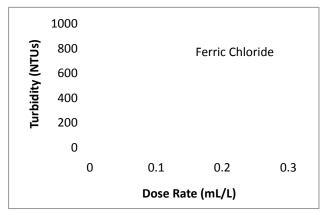


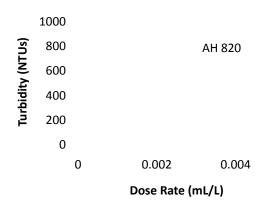


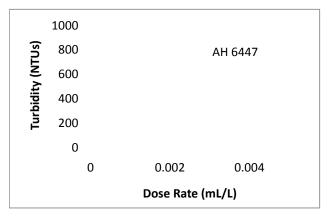
Material	Wabasha Subsoil (Box)
Classification	Slightly Organic Sandy Loam
USCS Symbol(s)	SM
Strata	Colluvuum
Geomorphology	Weathered Residual

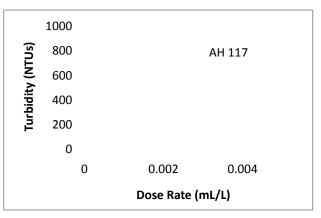


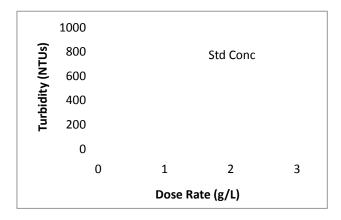




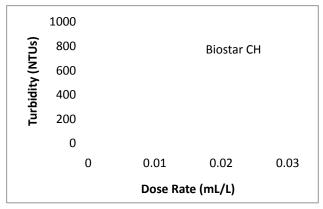


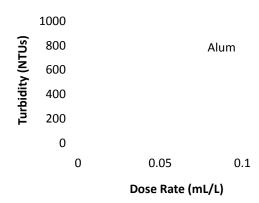


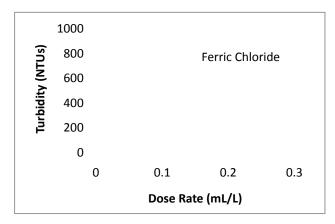


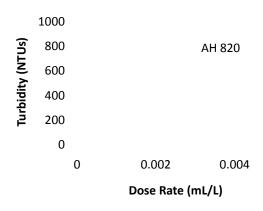


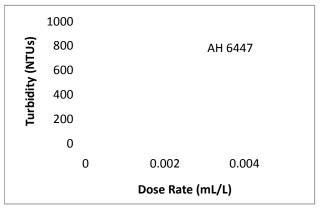
Material	Wabasha Subsoil (Rock)
Classification	Sandy Loam
USCS Symbol(s)	SM
Strata	Colluvuum
Geomorphology	Weathered Residual

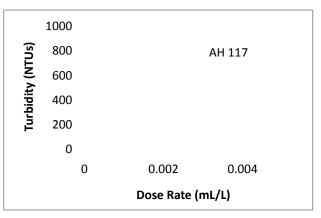


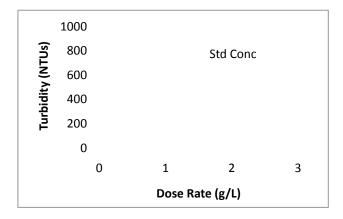




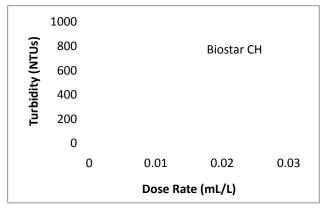


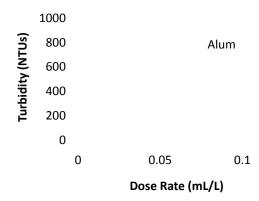


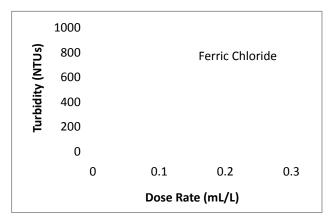


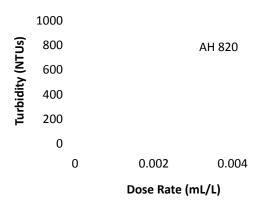


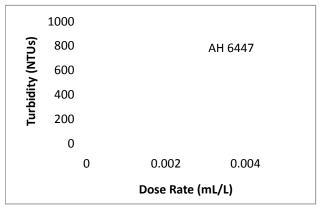
Material	Wabasha Topsoil
Classification	Slightly Organic Sandy Loam
USCS Symbol(s)	SM
Strata	Colluvuum
Geomorphology	Weathered Residual

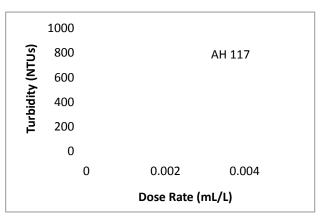


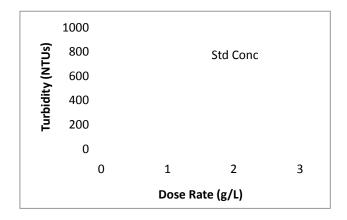




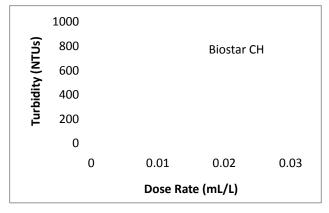


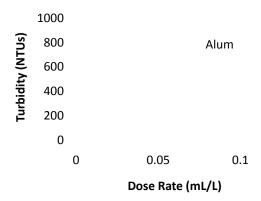


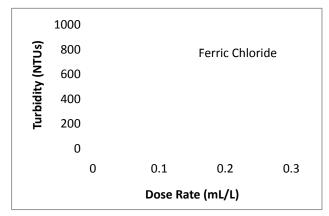


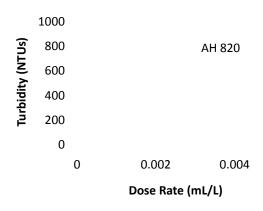


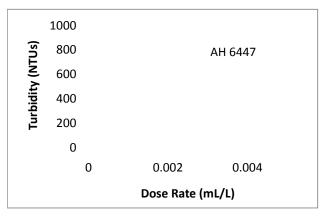
Material	Worthington Subsoil
Classification	Slightly Organic Sandy Loam
USCS Symbol(s)	SM
Strata	Ground Moraine
Geomorphology	Des Moines Lobe

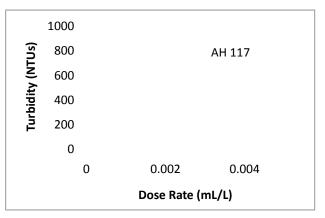


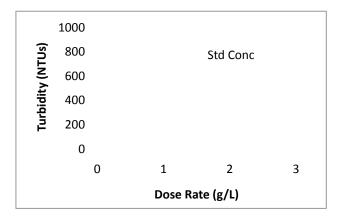




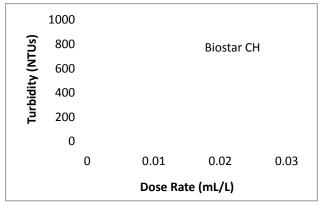


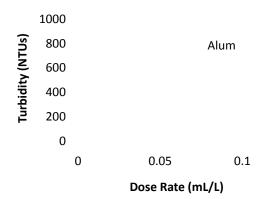


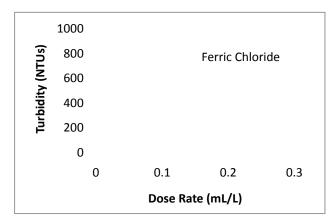


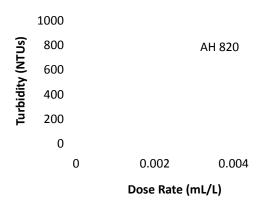


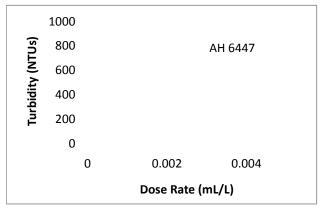
Material	Worthington Topsoil
Classification	Organic Sandy Loam
USCS Symbol(s)	SM
Strata	Ground Moraine
Geomorphology	Des Moines Lobe

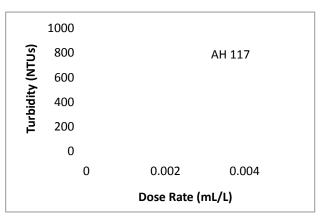












## Appendix D – Mixing Technique Analysis

## Appendix D

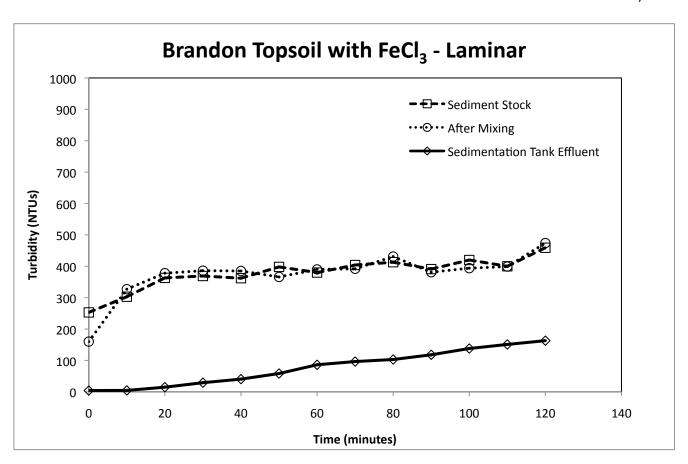
Mixing Technique Analysis

				Average		2-Hour	
		Average	Average Post	Settlement		Settlment	
		Sediment	Mixing	Tank Effluent		Tank Effluent	
		Stock Turbity	Turbidity	Turbidity	Average	Turbidity	2-Hour
Mixing Technique	Soil	(NTUs)	(NTUs)	(NTUs)	Removal	(NTN)	Removal
Laminar	<b>Brandon Topsoil</b>	378	374	77	%9'62	163	26.9%
2 ft Gravel	Perham Topsoil	725	713	212	70.8%	462	36.3%
6 ft Gravel	Perham Topsoil	302	257	19	93.7%	28	90.7%
2 ft Corregated Pipe	Perham Topsoil	487	434	61	87.5%	106	78.2%
8 ft Corregated Pipe	Perham Topsoil	324	288	34	89.5%	47.3	85.4%
Baffle Mixing	<b>Brandon Topsoil</b>	290	258	43	85.2%	104	64.1%
Floc Log	Perham Topsoil	306	181	29	78.1%	55.5	81.9%
Floc Pillow	Perham Topsoil	589	571	97	83.5%	202	65.7%
Drip Line Mixing	Mankato Topsoil	403	364	59	85.4%	108	73.2%
Sump Mixing	Mankato Topsoil	350	317	56	84.0%	143	59.1%
<b>Turbulant Air Mixing</b>	Mankato Topsoil	644	595	30	95.3%	32.8	94.9%
<b>Double Bladed Mixer</b>	Mankato Topsoil	834	807	141	83.1%	203	75.7%
Single Bladed Mixer	Mankato Topsoil	298	250	15	95.0%	15.7	94.7%
No Flocculant	Mankato Topsoil	830	816	378	54.5%	538	35.2%
No Flocculant	<b>Brandon Topsoil</b>	314	297	57	81.8%	114	63.7%
No Flocculant	Perham Topsoil	203	207	22	89.2%	57.5	71.7%

Analyst: Anthony Adderley Date: March 14, 2013

	Elapsed Time	Sediment Stock	After Mixing	Settlement Tank Effluent	Notes
	(min)	(NTUs)	(NTUs)	(NTUs)	
Soil Sample & Strata	0	253	160	4.07	
Brandon Topsoil	10	303	327	4.5	
Soil Mass (g)	20	363	378	14.79	
Water (gal)	30	369	386	29.1	
Sediment Stock Conc.	40	362	385	40.5	
100 g/gal	50	398	366	58.3	
1 200[9/ 90.	60	380	390	86.2	
Flocculant	70	404	392	96.5	
35% Ferric Chloride	80	413	431	103	
	2 90	391	381	118	
Water (L)	1 100	420	394	138	
Floc Stock Conc.	110	400	399	151	
2 mL/L	120	459	474	163	
				100	
Sediment Stock Flow	Averages	378	374	77	57%
500 mL/min	7.1.0.0900	1		80%	2-Hour Removal
Floc Stock Flow Rate					Z 110ul Reliiovul
25 mL/min				-	
Total Flow Rate				-	
525 mL/min				-	
323[1112/11111					
Floc Final Conc.				-	
0.10 mL/L				-	
0.10 2					
Sediment Mixing Method					
Air					
Water Temp 22 deg C				-	
Water remp   22 deg e					
Flash Mixing Method					
Laminar -				-	
straight channel					
no obstructions					
Slow Mixing Method					
Baffle				-	
Dame					
Sedimentation Method		<del> </del>		-	
20 gal Aquarium		<del> </del>		-	
20 gai Aquarium					
				-	
				1	

Analyst: Anthony Adderley Date: March 14, 2013



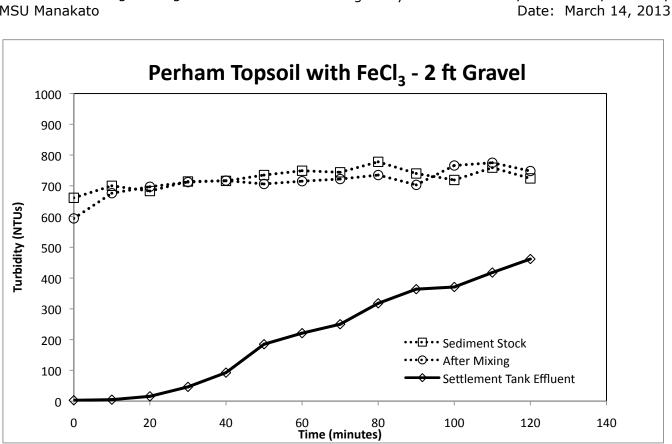




Analyst: Anthony Adderley Date: March 14, 2013

	Elapsed Time	Sediment Stock	After Mixing	Settlement Tank Effluent	Notes
	(min)	(NTUs)	(NTUs)	(NTUs)	
Soil Sample & Strata	0	661	594	2.73	
Perham Topsoil	10	700	676	4.71	
Soil Mass (g)	20	683	697	15.58	
Water (gal)	30	714	712	46.3	
Sediment Stock Conc.	40	716	717	92.6	
100 g/gal	50	735	706	185	
13,3	60	749	715	221	***************************************
Flocculant	70	744	722	250	
35% Ferric Chloride	80	778	735	318	
Floc Vol (mL) 2	90	740	703	364	
Water (L) 1	100	719	766	371	
Floc Stock Conc.	110	759	775	418	
2 mL/L	120	724	748	462	
Sediment Stock Flow	Averages	725	713	212	36%
500 mL/min				71%	2-Hour Removal
Floc Stock Flow Rate					
25 mL/min					
Total Flow Rate				-	
525 mL/min					
323 1112/11111					
Floc Final Conc.					
0.10 mL/L				-	
0110]1112/2					
Sediment Mixing Method				-	
Air				-	
Water Temp 22 deg C					
Water remp [22 deg C					
Flash Mixing Method				-	
Gravel 2 ft bed				-	
Graver 2 it bed				-	
Claw Mixing Mathad				-	
Slow Mixing Method					
Baffle					
Codimontation Mathed					
Sedimentation Method					
20 gal Aquarium					
				ļ	

Analyst: Anthony Adderley







Analyst: Isabelle Race

Date: March 12, 2013

Settlement After Sediment **Elapsed Tank** Time Stock Mixing **Notes Effluent** (NTUs) (NTUs) (min) (NTUs) Soil Sample & Strata 0 414 248 6.77 Perham Topsoil 325 10 203 32.1 Soil Mass (g) 249 5.77 20 311 Water (gal) 30 300 300 9.64 Sediment Stock Conc. 40 301 269 15.84 100 g/gal 50 308 250 19.63 299 60 258 21.1 **Flocculant** 70 269 304 21.2 35% Ferric Chloride 80 294 217 26.6 Floc Vol (mL) 90 283 239 21.5 Water (L) 100 289 263 19.77 Floc Stock Conc. 110 260 309 19.7 2 mL/L 120 274 237 28 257 Sediment Stock Flow **Averages** 302 91% 19 500 mL/min 94% 2-Hour Removal Floc Stock Flow Rate 25 mL/min Total Flow Rate 525 mL/min Floc Final Conc. 0.10 mL/L Sediment Mixing Method Air Water Temp 22 deg C Flash Mixing Method Gravel 6 ft bed Slow Mixing Method Baffle Sedimentation Method 20 gal Aquarium

Date: March 12, 2013 Perham Topsoil with FeCl₃ - 6 ft Gravel 1000 900 **- - Sediment Stock** ···⊙ ·· After Mixing 800 Settlement Tank Effluent 700 Turbidity (NTUs) 600 500 400 300 200 100 0 20 40 60 80 Time (minutes) 100 120 140 0





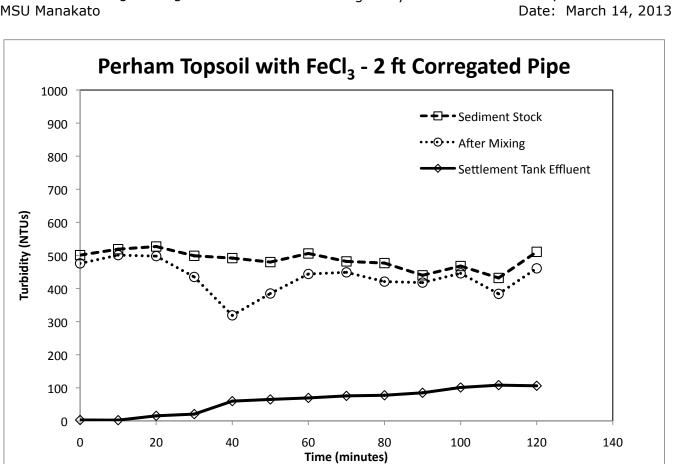
Analyst: Isabelle Race

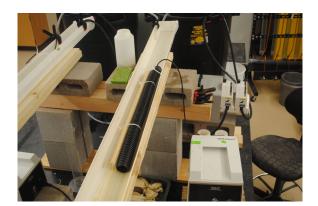
Analyst: Isabelle Race

Date: March 14, 2013

Settlement After Sediment **Elapsed Tank** Time Stock Mixing **Notes Effluent** (NTUs) (NTUs) (min) (NTUs) Soil Sample & Strata 0 501 476 2.74 Perham Topsoil 501 10 519 2.16 Soil Mass (g) 498 20 527 15.29 Water (gal) 30 499 435 20.5 Sediment Stock Conc. 40 492 319 59.7 100 g/gal 50 480 385 64.9 60 506 444 69.4 **Flocculant** 70 482 449 75.6 35% Ferric Chloride 80 477 421 77.4 Floc Vol (mL) 90 440 418 84.9 Water (L) 100 446 101 468 Floc Stock Conc. 110 432 384 108 2 mL/L 120 511 461 106 Sediment Stock Flow **Averages** 487 434 78% 61 500 mL/min 88% 2-Hour Removal Floc Stock Flow Rate 25 mL/min Total Flow Rate 525 mL/min Floc Final Conc. 0.10 mL/L Sediment Mixing Method Air Water Temp 22 deg C Flash Mixing Method Corregated Pipe 2 ft Slow Mixing Method Baffle Sedimentation Method 20 gal Aquarium

Analyst: Isabelle Race



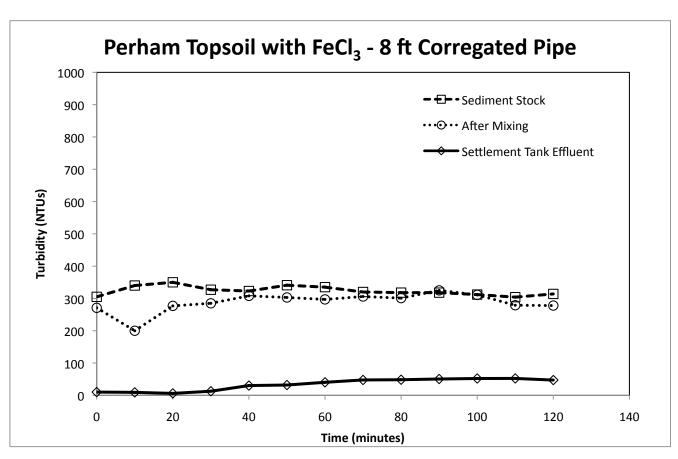


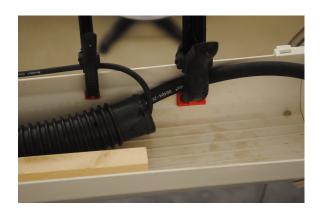


Analyst: Ryan Johnson Date: March 12, 2013

		1			<u> </u>
	Elapsed Time	Sediment Stock	After Mixing	Settlement Tank Effluent	Notes
	(min)	(NTUs)	(NTUs)	(NTUs)	
Soil Sample & Strata	0	305	271	10.17	
Perham Topsoil	10	340	200	9.23	
Soil Mass (g)	20	350	277	6.04	
Water (gal)	30	327	285	12.77	
Sediment Stock Conc.	40	323	308	30.3	
100 g/gal	50	341	303	32	
	60	335	297	40.4	
Flocculant	70	320	306	47.7	
35% Ferric Chloride	80	318	301	48.4	
Floc Vol (mL) 2	90	318	325	50.6	
Water (L) 1	100	312	311	52.2	
Floc Stock Conc.	110	304	279	52.4	
2 mL/L	120	314	278	47.3	
Sediment Stock Flow	Averages	324	288	34	85%
500 mL/min				90%	2-Hour Removal
Floc Stock Flow Rate					
25 mL/min					
Total Flow Rate				-	
525 mL/min					
Floc Final Conc.					
0.10 mL/L					
Sediment Mixing Method					
Air					
Water Temp 22 deg C					
Flash Mixing Method					
Corregated Pipe 8 ft					
Slow Mixing Method					
Baffle					
Sedimentation Method					
20 gal Aquarium					

Analyst: Ryan Johnson Date: March 12, 2013







Analyst: Isabelle Race Date: March 14, 2013

	Elapsed Time	Sediment Stock	After Mixing	Settlement Tank Effluent	Notes
	(min)	(NTUs)	(NTUs)	(NTUs)	
Soil Sample & Strata	0	218	213	2.07	
Brandon Topsoil	10	282	179	5.52	
Soil Mass (g)	20	336	257	3.29	
Water (gal)	30	286	251	6.45	
Sediment Stock Conc.	40	306	263	11.07	
100 g/gal	50	301	284	22.6	
	60	293	286	39.1	
Flocculant	70	288	279	50.5	
35% Ferric Chloride	80	313	277	67.4	
Floc Vol (mL) 2	90	301	279	81	
Water (L) 1	100	266	243	83.1	
Floc Stock Conc.	110	281	272	86.1	
2 mL/L	120	293	270	104	
Sediment Stock Flow	Averages	290	258	43	64%
500 mL/min				85%	2-Hour Removal
Floc Stock Flow Rate					
25 mL/min				-	
Total Flow Rate					
525 mL/min					
Floc Final Conc.					
0.10 mL/L					
Sediment Mixing Method Air					
Water Temp 22 deg C					
, , , , , , , , , , , , , , , , , , , ,	1				
Flash Mixing Method					
Hydraulic Baffles					
Slow Mixing Method					
Baffle					
				I	
Sedimentation Method					
20 gal Aquarium					
				<u> </u>	•
				-	

Analyst: Isabelle Race

Date: March 14, 2013 Brandon Topsoil with FeCl₃ - Baffle Mixing **- - Sediment Stock** ···⊙ ·· After Mixing Settlement Tank Effluent Turbidity (NTUs) Time (minutes)

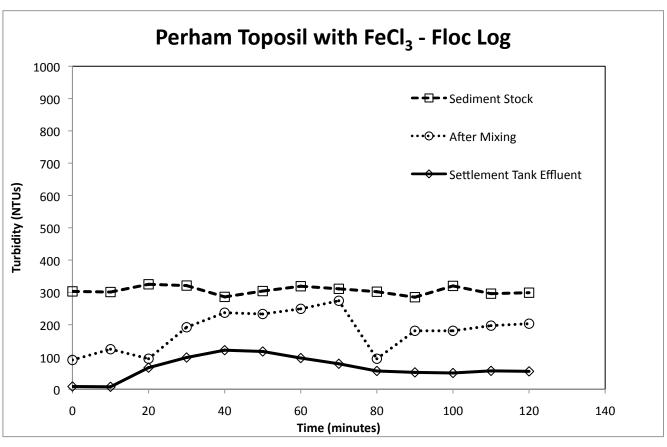




Analyst: Isabelle Race Date: March 12, 2013

				Settlement	
	Elapsed	Sediment	After	Tank	
	Time	Stock	Mixing	Effluent	Notes
	(	(NITH-)	(NITH-)		
	(min)	(NTUs)	(NTUs)	(NTUs)	
Soil Sample & Strata	0	303	90.7	8.6	
Perham Topsoil	10	301	124	7.63	
Soil Mass (g)	20	325	94.5	66.9	
Water (gal)	30	321	192	98.3	
Sediment Stock Conc.	40	286	237	121	
100 g/gal	50	304	233	117	
	60	319	249	96.7	
Flocculant	70	311	274	78.9	
35% Ferric Chloride	80	302	94.5	56.6	
Floc Vol (mL) 2	90	285	181	52.3	
Water (L) 1	100	320	181	50.4	
Floc Stock Conc.	110	296	197	57	
2 mL/L	120	299	203	55.5	
Sediment Stock Flow	Averages	306	181	67	82%
500 mL/min				78%	2-Hour Removal
Floc Stock Flow Rate					
25 mL/min					
Total Flow Rate					
525 mL/min					
Floc Final Conc.					
0.10 mL/L					
Sediment Mixing Method					
Air					
Water Temp 22 deg C					
Flash Mixing Method					
Floc Log					
1.100 2.09				<b>-</b>	
Slow Mixing Method					
Baffle	•				
Dame					
Sedimentation Method					
20 gal Aquarium					
				<b></b>	

Analyst: Isabelle Race Date: March 12, 2013

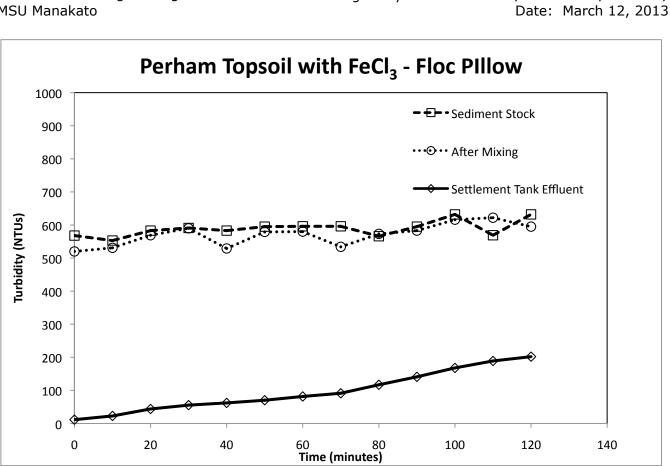




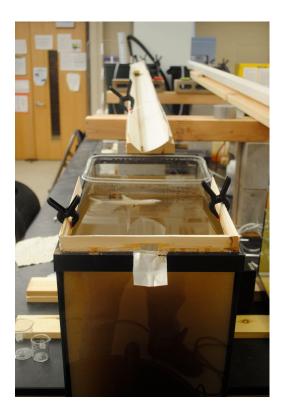


Analyst: Anthony Adderley Date: March 12, 2013

	Elapsed	Sediment	After	Settlement Tank	Natas
	Time	Stock	Mixing	Effluent	Notes
	(min)	(NTUs)	(NTUs)	(NTUs)	
Soil Sample & Strata	<b>a</b> 0	568	520	11.57	
Perham Topsoil	10	553	531	22.8	
Soil Mass (g)	20	583	569	43.9	
Water (gal)	30	591	590	55.5	
Sediment Stock Conc.	40	583	529	62.2	
100 g/gal	50	595	579	70.3	
	60	596	580	81.7	
Flocculant	70	596	534	91.4	
35% Ferric Chloric	de 80	566	573	117	
Floc Vol (mL)	2 90	595	583	141	
Water (L)	1 100	632	616	168	
Floc Stock Conc.	110	569	622	189	
2 mL/L	120	632	595	202	
Sediment Stock Flow	Averages	589	571	97	66%
500 mL/min				84%	2-Hour Removal
Floc Stock Flow Rate					
25 mL/min					
Total Flow Rate					
525 mL/min					
Floc Final Conc.					
0.10 mL/L				-	
Sediment Mixing Meth	od			-	
Air					
Water Temp 22 deg	С				
F L 325					
Flash Mixing Method					
Floc Pillow					
Slow Mixing Method				-	
Baffle					
Sedimentation Method	1			<del> </del>	
20 gal Aquarium	<del>-</del>	·		-	
20 30.7.400.70111					
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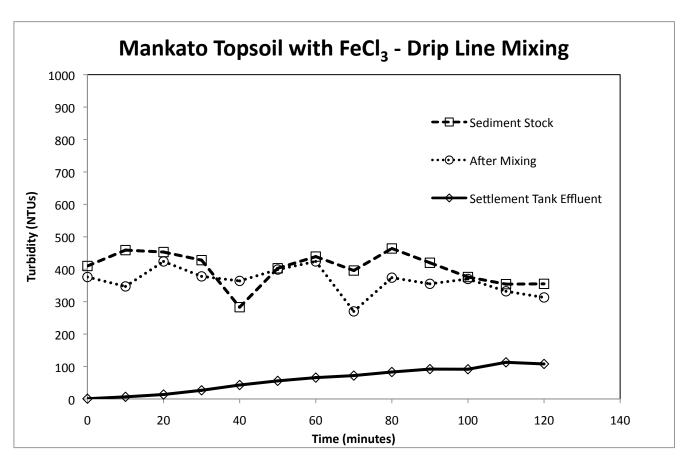


Analyst: Anthony Adderley

Analyst: Anthony Adderley Date: March 15, 2013

	Elapsed Time	Sediment Stock	After Mixing	Settlement Tank Effluent	Notes
	(min)	(NTUs)	(NTUs)	(NTUs)	
Soil Sample & Strata	0	410	376	0.78	
Mankato Topsoil	10	459	347	6.53	
Soil Mass (g)	20	453	424	13.76	
Water (gal)	30	428	378	26.7	
Sediment Stock Conc.	40	283	364	43	
100 g/gal	50	403	399	55.8	
100 9/ 90.	60	439	424	65.8	
Flocculant	70	396	270	72.1	
35% Ferric Chloride		464	374	83.1	
Floc Vol (mL)	1 90	420	355	92.1	
Water (L)	1 100	376	370	91.7	
Floc Stock Conc.	110	354	332	113	
1 mL/L	120	355	313	108	
11110/			313	100	
Sediment Stock Flow	Averages	403	364	59	73%
500 mL/min	7.10.4900	1		85%	2-Hour Removal
Floc Stock Flow Rate					Z mour removar
50 mL/min					
Total Flow Rate				-	
550 mL/min					
<u></u>					
Floc Final Conc.					
0.09 mL/L				-	
0.05]1112/2					
Sediment Mixing Metho					
Air	<u>u</u>				
Water Temp 22 deg C					
Water remp  22 deg e					
Flash Mixing Method					
Drip Line					
Drip Line					
Slow Mixing Method					
Baffle				-	
Darrie					
Sedimentation Method		<del> </del>			
20 gal Aquarium		<del> </del>		-	
Zo gai Aquanuili					
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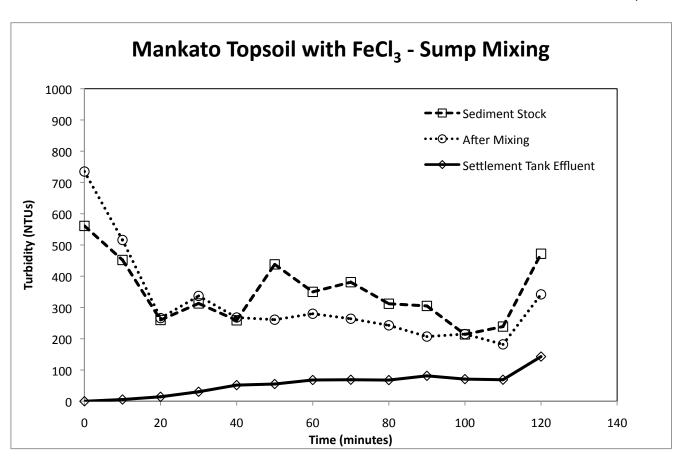
Analyst: Anthony Adderley Date: March 15, 2013



Analyst: Alex Raymond Date: March 15, 2013

	Elapsed Time	Sediment Stock	After Mixing	Settlement Tank	Notes
				Effluent	
	(min)	(NTUs)	(NTUs)	(NTUs)	
Soil Sample & Strata	0	561	735	0	
Mankato Topsoil	10	452	516	5.59	
Soil Mass (g)	20	260	266	14.38	
Water (gal)	30	313	337	30.4	
Sediment Stock Conc.	40	259	268	51.7	
100 g/gal	50	438	261	55.2	
	60	350	280	68.1	
Flocculant	70	381	264	68.9	
35% Ferric Chloride	80	312	243	67.8	
Floc Vol (mL) 2	90	305	207	81.4	
Water (L) 1	100	214	215	70.9	
Floc Stock Conc.	110	239	182	68.9	
2 mL/L	120	472	342	143	
Sediment Stock Flow	Averages	350	317	56	59%
500 mL/min				84%	2-Hour Removal
Floc Stock Flow Rate					
25 mL/min					
Total Flow Rate				-	
550 mL/min					
<u></u>					
Floc Final Conc.				-	
0.09 mL/L				-	
Codiment Mixing Method				-	
Sediment Mixing Method Air					
				-	
Water Temp 22 deg C					
Clash Mississ Marth 1					
Flash Mixing Method				<b>.</b>	
Sump Mixing					
w/20 gal/hr					
centrifugal pump					
Slow Mixing Method					
Baffle					
Sedimentation Method					
20 gal Aquarium					
		1			

Analyst: Alex Raymond Date: March 15, 2013



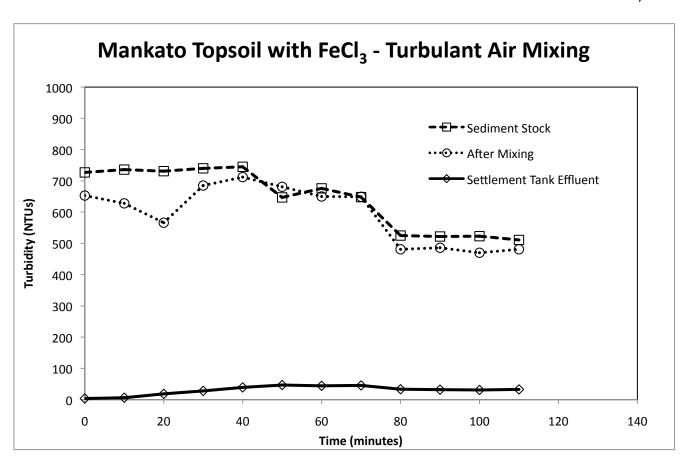


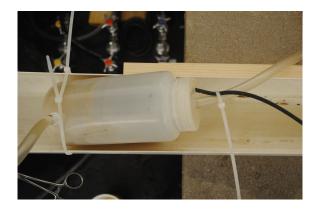


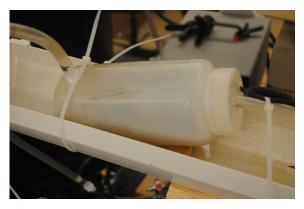
Analyst: Alex Raymond Date: March 23, 2013

	Elapsed Time	Sediment Stock	After Mixing	Settlement Tank Effluent	Notes
	(min)	(NTUs)	(NTUs)	(NTUs)	
Soil Sample & Strata	0	727	653	3.65	
Mankato Topsoil	10	736	628	6.12	
Soil Mass (g)	20	731	566	18.59	
Water (gal)	30	740	685	28	
Sediment Stock Conc.	40	745	712	39.2	
100 g/gal	50	647	681	46.8	
	60	676	650	44.3	
Flocculant	70	648	649	45.6	
35% Ferric Chloride	80	525	481	33.4	
Floc Vol (mL) 2	90	522	486	32	
Water (L) 1	100	523	470	31	
Floc Stock Conc.	110	511	481	32.8	
2 mL/L					
	Averages	644	595	30	95%
Sediment Stock Flow				95%	2-Hour Removal
500 mL/min					
Floc Stock Flow Rate				-	
25 mL/min					
Total Flow Rate				-	
550 mL/min					
Floc Final Conc.					
0.09 mL/L					
Sediment Mixing Method					
Air					
Water Temp 22 deg C				-	
, , , , , , , , , , , , , , , , , , ,					
Flash Mixing Method					
Turbulant air injection	<u>1</u>				
Slow Mixing Method					
Baffle					
Sedimentation Method					
20 gal Aquarium					

Analyst: Alex Raymond Date: March 23, 2013







Analyst: Alex Raymond Date: March 23, 2013

		<u> </u>		6-141	I
	Elapsed Time	Sediment Stock	After Mixing	Settlement Tank Effluent	Notes
	(min)	(NTUs)	(NTUs)	(NTUs)	
Soil Sample & Strata	0	835	782	2.89	
Mankato Topsoil	10	860	760	14.17	
Soil Mass (g)	20	831	824	39.5	
Water (gal)	30	848	839	82.4	
Sediment Stock Conc.	40	842	790	100	Floc stock nozzle
100 g/gal	50	829	760	129	noted as clogged
1,5,1,5	60	824	793	164	
Flocculant	70	813	813	185	
35% Ferric Chloride	80	835	817	221	insufficient floc stock
Floc Vol (mL) 2	90	889	847	230	noted. Not enough
Water (L) 1	100	852	822	223	has been injected.
Floc Stock Conc.	110	755	823	235	(approx half dose)
2 mL/L	120	835	819	203	
Sediment Stock Flow	Averages	834	807	141	76%
500 mL/min				83%	2-Hour Removal
Floc Stock Flow Rate					
25 mL/min				-	
Total Flow Rate				-	
550 mL/min				-	
<u>[ 330</u> [IIIL/IIIII					
Floc Final Conc.				-	
0.09 mL/L				-	
Sediment Mixing Method					
Air				-	
Water Temp 22 deg C				-	
Water remp [22 deg C					
Flach Mixing Mothod				-	
Flash Mixing Method  Double Blade Mixer					
Double Blade Mixel				-	
Class Missing Mathed					
Slow Mixing Method					
Baffle					
Coding and this is Markey				-	
Sedimentation Method		ļ			
20 gal Aquarium					
				ļ	

 Analyst: Alex Raymond

Date: March 23, 2013 Mankato Topsoil with FeCl₃ - Double Bladed Mixer Turbidity (NTUs) **- ← -** • Sediment Stock ···⊙ ·· After Mixing Settlement Tank Effluent 

Time (minutes)

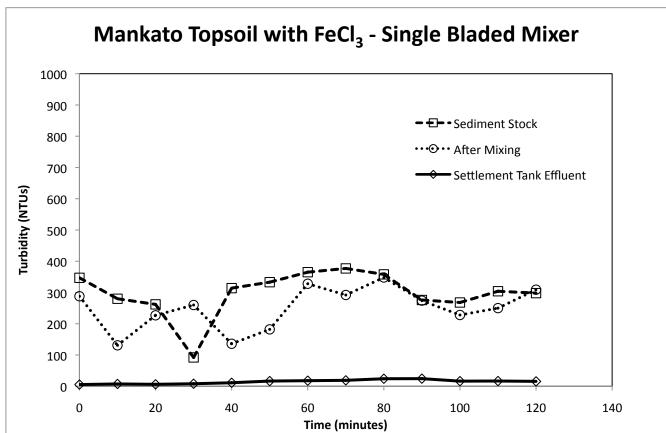


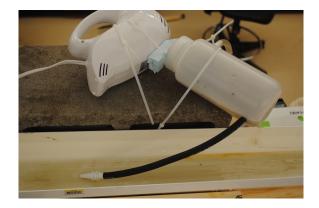


Analyst: Sam Stoffels Date: March 23, 2013

	Elapsed Time	Sediment Stock	After	Settlement Tank	Notes
	Time	Stock	Mixing	Effluent	Notes
	(min)	(NTUs)	(NTUs)	(NTUs)	
Soil Sample & Strata	0	347	288	5.43	
Mankato Topsoil	10	280	131	7.43	
Soil Mass (g)	20	262	227	6.26	
Water (gal)	30	92.2	260	8.09	
Sediment Stock Conc.	40	314	136	11.25	
100 g/gal	50	333	182	16.67	
	60	365	328	17.88	
Flocculant	70	377	292	19	
35% Ferric Chloride	80	358	348	24.1	
Floc Vol (mL)	2 90	276	274	24.4	
Water (L)	1 100	268	228	16.39	
Floc Stock Conc.	110	304	250	16.78	
2 mL/L	120	298	309	15.7	
Sediment Stock Flow	Averages	298	250	15	95%
500 mL/min				95%	2-Hour Removal
Floc Stock Flow Rate					
25 mL/min					
Total Flow Rate					
550 mL/min					
Floc Final Conc.					
0.09 mL/L				-	
Sediment Mixing Method	<b> </b>				
Air	1			-	
Water Temp 22 deg C					
Tracer remp [22 deg C		·		-	
Flash Mixing Method		<del> </del>		-	
Single Blade Mixer		<del> </del>		-	
Siligle blade Mixel					
Claw Mixing Mathad					
Slow Mixing Method Baffle					
Daille					
Codimontation Mothed			***************************************	-	
Sedimentation Method		ļ		-	
20 gal Aquarium					
		ļ			
		ļ		<b>-</b>	
				ļ	

Analyst: Sam Stoffels Date: March 23, 2013



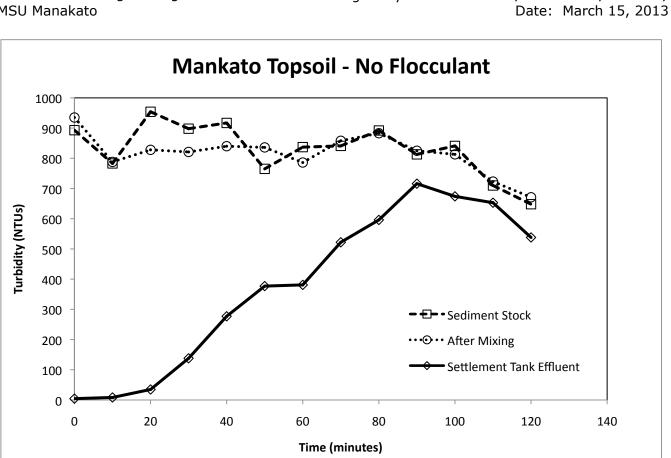




Analyst: Anthony Adderley Date: March 15, 2013

	Elapsed Time	Sediment Stock	After Mixing	Settlement Tank Effluent	Notes
	(min)	(NTUS)	(NITUS)		
Soil Sample & Strata	<b>(min)</b> 0	(NTUs) 893	(NTUs) 935	(NTUs) 4.49	
Mankato Topsoil	10	783	788	8.38	
Soil Mass (g)	20	954	828	34.8	
Water (gal)	30	898	821	138	
Sediment Stock Conc.	40	917	840	277	
				377	
100 g/gal	50	765	836		
	60	837	786	381	
Flocculant	70	841	858	522	
none	80	892	883	596	
Floc Vol (mL) 0		813	825	716	
Water (L) 0		841	813	674	
Floc Stock Conc.	110	710	723	653	
0 mL/L	120	648	671	538	
Sediment Stock Flow	Averages	830	816	378	35%
500 mL/min				54%	2-Hour Removal
Floc Stock Flow Rate					
0 mL/min					
Total Flow Rate					
500 mL/min					
Floc Final Conc.				-	
0.00 mL/L	-				
Sediment Mixing Method					
Air				-	
Water Temp 22 deg C	_				
Water remp [22 deg C	-				
Elach Mixing Mothed	-				
Flash Mixing Method					
ivone					
CL. Mi M	-	ļ			
Slow Mixing Method					
Baffle					
Sedimentation Method					
20 gal Aquarium					
					•
	-	<b>†</b>			

Analyst: Anthony Adderley

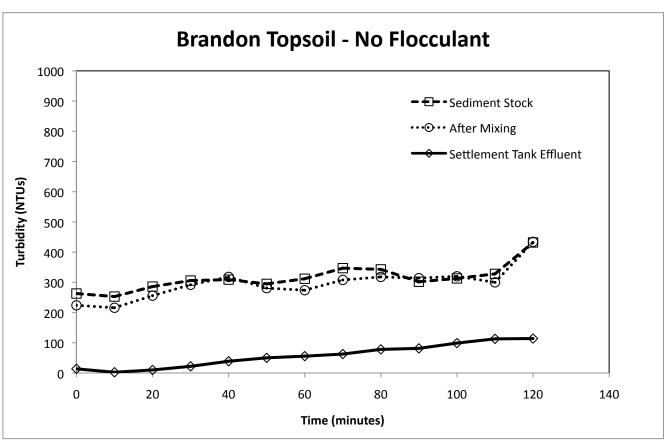


Analyst: Sam Stoffels

Date: March 23, 2013

Settlement After Sediment **Elapsed Tank** Time Stock Mixing **Notes Effluent** (NTUs) (NTUs) (NTUs) (min) Soil Sample & Strata 0 263 224 13.73 Brandon Topsoil 10 253 216 2.4 Soil Mass (g) 2880 10.08 20 286 256 Water (gal) 28,86 292 22.1 30 306 Sediment Stock Conc. 40 309 318 38.9 100 g/gal 50 295 281 50.2 60 312 274 55.7 **Flocculant** 70 347 308 62.7 none 80 343 318 78.1 Floc Vol (mL) 0 90 302 314 81.5 Water (L) 100 320 99 313 Floc Stock Conc. 110 328 300 113 0 mL/L 120 432 434 114 315 Sediment Stock Flow **Averages** 297 57 64% 500 mL/min 82% 2-Hour Removal Floc Stock Flow Rate 0 mL/min Total Flow Rate 500 mL/min Floc Final Conc. 0.00 mL/L Sediment Mixing Method Air Water Temp 22 deg C Flash Mixing Method None Slow Mixing Method Baffle Sedimentation Method 20 gal Aquarium

Analyst: Sam Stoffels Date: March 23, 2013

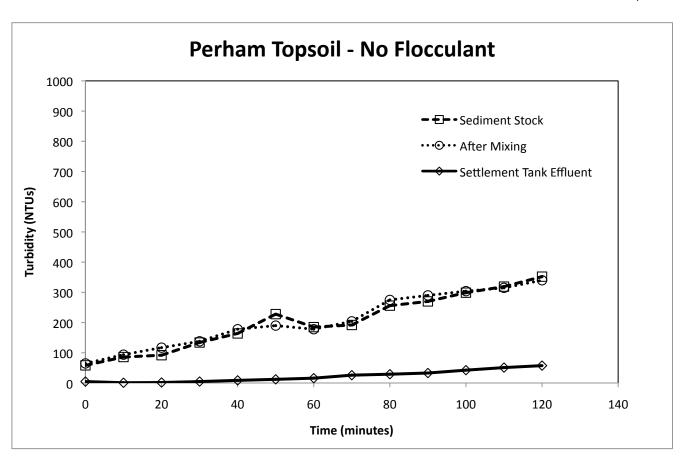


Analyst: Sam Stoffels

Date: March 23, 2013

Settlement After Sediment **Elapsed Tank Time** Stock Mixing **Notes Effluent** (NTUs) (NTUs) (NTUs) (min) Soil Sample & Strata 0 58.3 64.7 4.72 Perham Topsoil 85.9 10 93.8 0.82 Soil Mass (g) 1.59 1150 20 92 117 Water (gal) 28.86 134 138 4.72 30 Sediment Stock Conc. 40 164 178 8.69 39.8 g/gal 50 228 190 12.13 60 185 178 15.99 **Flocculant** 70 192 204 25.9 none 80 256 275 29.1 Floc Vol (mL) 0 90 270 290 33.0 Water (L) 100 299 304 42.5 Floc Stock Conc. 110 319 315 51.0 0 mL/L 120 352 340 57.5 Sediment Stock Flow **Averages** 203 207 72% 22 500 mL/min 89% 2-Hour Removal Floc Stock Flow Rate 0 mL/min Total Flow Rate 500 mL/min Floc Final Conc. 0.00 mL/L Sediment Mixing Method Air Water Temp 22 deg C Flash Mixing Method None Slow Mixing Method Baffle Sedimentation Method 20 gal Aquarium

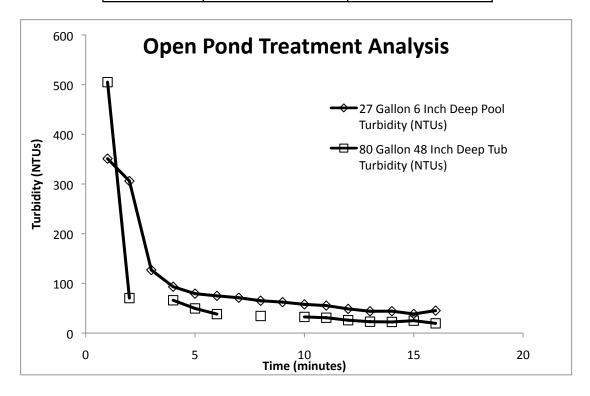
Analyst: Sam Stoffels Date: March 23, 2013



## Appendix E – Open Pond Treatment Model Analysis

25 g/L Mankato Topsoil 2 minute rapid mix
0.2 mL/L Ferric Chloride 35% Pool: Turbine Mixer Tub: Air Mix
Direct apply Turbidity measured 3 inches below surface

Time (minutes)	27 Gallon 6 Inch Deep Pool Turbidity (NTUs)	80 Gallon 48 Inch Deep Tub Turbidity (NTUs)
1.0	351	505
2.0	306	70.4
3.0	127	
4.0	93.3	66
5.0	79.3	49.6
6.0	74.9	38.2
7.0	71.1	
8.0	65.1	34.4
9.0	62.3	
10.0	57.8	32.5
11.0	55.3	30.9
12.0	48.7	25.9
13.0	43.9	22.9
14.0	44.1	22.4
15.0	38.5	25
16.0	45.2	19.6



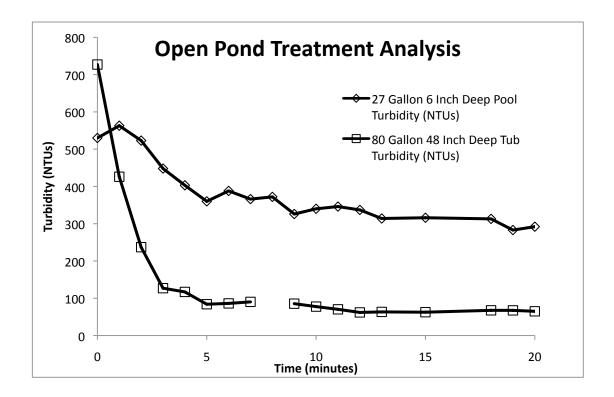
25 g/L Mankato Topsoil 2 minute rapid mix

0.2 mL/L Ferric Chloride 35% Pool: Air Mix Tub: Air Mix

Direct apply

Turbidity measured 3 inches below surface

Time	27 Gallon 6 Inch	80 Gallon 48 Inch	1
(minutes)	Deep Pool Turbidity	Deep Tub Turbidity	
(minutes)	(NTUs)	(NTUs)	
0.0	530	727	]
1.0	563	426	
2.0	523	237	
3.0	448	127	
4.0	403	117	
5.0	360	84	
6.0	388	86.2	
7.0	366	90.3	
8.0	372		
9.0	326	85.5	
10.0	340	77.6	
11.0	346	70.3	
12.0	337	62	
13.0	314	63.4	
15.0	316	62.6	
18.0	313	67.6	air mix
19.0	283	67.5	
20.0	292	65	
21.0	292	55.1	
22.0	281	56.1	
24.0	277	56.8	
25.0	283	51.7	
26.0	278		





			Τι	urbidity [NT	U]			
					Time [min]			
Group	Depth[ft]	0	10	20	30	40	50	60
		330	72					
	0	276	81.3					
		285						
est		252	146	22.1	17.04	17.17	17.54	15.43
≥	2	243	151	22.7	18.57	17.24	21.7	16.86
р		236						
J Burns and D West		248	192	45.1	16.5	14.41	16.73	12.87
lrns	4	216	201	39.7	16.35	15.47	17.19	16.19
Bu Bu		231						
		339	305	69.8	12.71	12.04	13.49	12.15
	6	318	322	90	15.72	13.01	16.07	11.96
		313						
		232	88					
	0	376	107					
چ								
aite		488	280	24.4	39.6	24.3	21.8	17
ЕJ	2	620	249	34.9	23.5	20.2	21.2	15.93
pue								
u e		620	160	220	55	25.1	37.1	41.7
sqc	4	612	469	187	32.9	21	31.2	34.4
C Jacobsen and E Jaiteh								
- C		560	586	296	99	15.01	14.14	12.73
	6	586	588	209	154	23.5	12.1	12.69
		551						
	0	456						
gel		492	275	109	188	36.8	25.5	16.34
En	2	428	218	73.4	137	33.5	25.6	15.85
Αp								
Z Lingl and A Engel		369	623	233	203	32.2	22.9	23.9
ngl	4	382	622	354	160	34.9	21.4	23.4
Z Li								
		410	512	189	212	70.1	20.4	15.45
	6	535	596	110	164	46.8	16.87	13.67

			Τι	urbidity [NT	 Ul			
				, ,	Time [min]			
Group	Depth[ft]	0	10	20	30	40	50	60
	0							
K Fischer and A Stanek		276	85.3	43.2	46.4	26.9	21.4	16.2
Sta	2	309	87.2	42.7	34.6	35.6	21.5	18.94
Αρ		284	82.2	47.1	39.0	32.8	22.6	12.24
an .		587	213	71.2	90.4	20.9	17.72	19.18
her	4	571	226	64.4	42.1	32.6	20.9	14.97
Fisc		579	218	68.4		18.7	17.95	16.8
<u> </u>		362	328	38.3	64	33.3	26.6	20.5
	6	429	345	64.1	55.9	28.8	21.8	24.4
		462	339	60.4	66.1	19.77	23.3	24.9
		650	330					
	0	635	340					
<u>_</u>		625	338					
S Bulfer and R Johnson		640	492	159	82.5	39.4	57.8	36.4
	2	633	443	210	48.6	70	64.3	39.6
~		650	622	168	80.3	90	54.5	35
anc		497	529	605	106	33.3	39.8	22.4
fer	4	550	478	605	66.7	36	38.2	32.2
Bul		311	466	558	112	33.3	37.6	33.2
S		680	644	666	43.9	37.1	28.5	21.1
	6	578	621	790	57.4	49.6	33.5	25.1
		578	639	765	47	54	32.1	20.9
		329	84.1					
	0	337						
er		349						
rap		386	156	47	33.9	28.8	21.8	20.5
0 0	2	391	150	45.1	33.8	28.7	22.7	20.5
l pu		393						
G Gomez and D Draper		372	171	47.2	37.1	28.9	25.4	22.2
me	4	379	167	49.5	35.6	29	26.4	22.9
69		389						
٥		309	331	62.8	26.2	23.9	20.5	20.6
	6	350	297	58.7	28.2	23.1	20.3	22
		369						

			Tu	urbidity [NT	 U]			
				, -	Time [min]			
Group	Depth[ft]	0	10	20	30	40	50	60
		388	180	99.5	86.7	71.9	78.6	
	0	425	185	86.1	91.2	81	117	
συ		383	183	77.5	94.2	62.8	58.2	
T Elbert and S Huneke		371	285	87.2	179	46	27.8	24.1
Η	2	371	426	96	315	44.1	26.6	24.8
S P		351	287	64.4	308	40.7	31.3	23
ano		376	716	221	66.5	44.7	22.2	25
ert	4	307	579	193	55.1	32.9	22.8	28.3
Elb		396	561	265	63.5	31.1	14.84	25.6
-		251	621	277	155	41	15.4	16.53
	6	275	546	293	124	23.4	15.89	13.21
		240	671	278	96.7	22.8	15.63	13.69
		505	176					
	0	497	209					
р		468	217					
nor		604	310	156	119	185	61.2	52.1
ayn	2 [	427	403	182	134	202	50.9	40.6
D Tipp and A Raymond		662	297	175	146	144	48.7	58.3
, bn	4	497	731	302	212	160	51.8	57.5
ра		495	659	342	201	192	54.5	48.1
Тір		521	765	311	164	154	54	44.9
Q		608	776	386	108	58.1	37.1	29.2
	6	549	655	327	75.3	40.3	43.5	38.6
		587	771	286	73.1	48	33.4	28.3
		539						
	0	681						
_		734						
igeı		494	214	150	53.2	38.2	58.2	41.2
Or	2 [	508	202	129	60.5	42.3	45.9	41.2
Σ		429	280	105	55.6	46	63.5	40.7
S Muir and M Origer		294	326	49.4	23.6	18.2	14.7	13.2
uir	4	278	338	43.2	22.8	17.1	14.8	12.1
Σ		290	366	43.2	21.7	18.1	16	12.9
01		557	348	103	48.4	19.6	22.4	14.3
	6	449	332	109	33	24	22.6	16.5
		610	270	117	38.1	22	21.3	15.8

		Chautina	Removals	s (% based c	n average st	arting turbi	dity reading	at depth)
		Starting Turbidity			Time	[min]		
Group	Depth[ft]	Turblatty	10	20	30	40	50	60
		330	76%					
	0	276	73%					
		285						
est		252	40%	91%	93%	93%	93%	94%
>	2	243	38%	91%	92%	93%	91%	93%
J Burns and D West		236						
s an		248	17%	81%	93%	94%	93%	94%
ırı	4	216	13%	83%	93%	93%	93%	93%
J Br		231						
		339	6%	78%	96%	96%	96%	96%
	6	318	0%	72%	95%	96%	95%	96%
1		313						
		232	71%					
	0	376	65%					
-i								
aite		488	49%	96%	93%	96%	96%	97%
EJ	2	620	55%	94%	96%	96%	96%	97%
and								
en 🤅		620	74%	64%	91%	96%	94%	93%
sqc	4	612	24%	70%	95%	97%	95%	94%
C Jacobsen and E Jaiteh								
ပ		560	-2%	48%	83%	97%	98%	98%
	6	586	-3%	64%	73%	96%	98%	98%
1								
		551						
	0	456						
gel		492	40%	76%	59%	92%	94%	96%
En	2	428	53%	84%	70%	93%	94%	97%
φ								
l an		369	-66%	38%	46%	91%	94%	94%
Z Lingl and A	4	382	-66%	6%	57%	91%	94%	94%
ZL								
		410	-8%	60%	55%	85%	96%	97%
	6	535	-26%	77%	65%	90%	96%	97%

		Classi's a	Removals	s (% based c	n average st	arting turbi	dity reading	at depth)
		Starting			Time			
Group	Depth[ft]	Turbidity	10	20	30	40	50	60
	0							
*								
ane		276	71%	85%	84%	91%	93%	94%
۱ St	2	309	70%	85%	88%	88%	93%	93%
K Fischer and A Stanek		284	72%	84%	87%	89%	92%	96%
r ar		587	63%	88%	84%	96%	97%	97%
iþei	4	571	61%	89%	93%	94%	96%	97%
Fisc		579	62%	88%	100%	97%	97%	97%
$\preceq$		362	21%	91%	85%	92%	94%	95%
	6	429	17%	85%	87%	93%	95%	94%
		462	19%	86%	84%	95%	94%	94%
		650	48%					
S Bulfer and R Johnson	0	635	47%					
		625	47%					
		640	23%	75%	87%	94%	91%	94%
	2	633	31%	67%	92%	89%	90%	94%
R.		650	3%	74%	87%	86%	91%	95%
pue		497	-17%	-34%	77%	93%	91%	95%
er 3	4	550	-6%	-34%	85%	92%	92%	93%
3ulf		311	-3%	-23%	75%	93%	92%	93%
SE		680	-5%	-9%	93%	94%	95%	97%
	6	578	-1%	-29%	91%	92%	95%	96%
		578	-4%	-25%	92%	91%	95%	97%
		329	75%					
	0	337						
_		349						
Draper		386	60%	88%	91%	93%	94%	95%
ے	2	391	62%	88%	91%	93%	94%	95%
дρ		393						
G Gomez and D		372	55%	88%	90%	92%	93%	94%
nez	4	379	56%	87%	91%	92%	93%	94%
Gor		389						
9		309	3%	82%	92%	93%	94%	94%
	6	350	13%	83%	92%	93%	94%	94%
		369						

		Chautina	Removals	(% based o	n average st	arting turbi	dity reading	at depth)	
			Removals (% based on average starting turbidity reading at Time [min]						
Group	Depth[ft]	Turbluity	10	20	30	40	50	60	
		388	55%	75%	78%	82%	80%		
	0	425	54%	78%	77%	80%	71%		
D Tipp and A Raymond		383	54%	81%	76%	84%	85%		
Jek		371	22%	76%	51%	87%	92%	93%	
] E	2	371	-17%	74%	14%	88%	93%	93%	
S		351	21%	82%	15%	89%	91%	94%	
ano		376	-99%	39%	82%	88%	94%	93%	
ert	4	307	-61%	46%	85%	91%	94%	92%	
Elb		396	-56%	26%	82%	91%	96%	93%	
<b>⊢</b>		251	-143%	-8%	39%	84%	94%	94%	
	6	275	-114%	-15%	51%	91%	94%	95%	
		240	-163%	-9%	62%	91%	94%	95%	
		505	64%						
	0	497	57%						
D		468	56%						
00		604	45%	72%	79%	67%	89%	91%	
ayn	2	427	29%	68%	76%	64%	91%	93%	
A R		662	47%	69%	74%	74%	91%	90%	
nd ,	4	497	-45%	40%	58%	68%	90%	89%	
ра		495	-31%	32%	60%	62%	89%	90%	
l di		521	-52%	38%	67%	69%	89%	91%	
۵		608	-33%	34%	81%	90%	94%	95%	
	6	549	-13%	44%	87%	93%	93%	93%	
		587	-33%	51%	87%	92%	94%	95%	
		539							
	0	681							
		734							
iger		494	55%	69%	89%	92%	88%	91%	
o	2	508	58%	73%	87%	91%	90%	91%	
Σ 7		429	41%	78%	88%	90%	87%	91%	
S Muir and M Or		294	-13%	83%	92%	94%	95%	95%	
ui.	4	278	-18%	85%	92%	94%	95%	96%	
Σ		290	-27%	85%	92%	94%	94%	96%	
",		557	35%	81%	91%	96%	96%	97%	
	6	449	38%	80%	94%	96%	96%	97%	
		610	50%	78%	93%	96%	96%	97%	

Average Starting	438
Turbidity (NTUs)	438

	10 min	20 min	30 min	40 min	50 min	60 min
Average Removal All Tests & Depths	16%	60%	80%	90%	93%	94%
Average Turbidity All Tests & Depths (NTUs)	361	181	85	46	31	25

## Minnesota State University, Mankato Department of Mechanical and Civil Engineering

Spring 2013

### CIVE 436 – Civil Engineering Experimentation

## Experiment 10 Sedimentation Lab

Objective: Evaluate water quality for the design of primary sedimentation basin.

Method: Perform evaluations of water quality during sedimentation performed after flocculation

#### **Evaluations:**

Turbidity

Note: perform all analyses <u>in triplicate</u> and report all results as MEAN ± STANDARD DEVIATION units (%RSD), where % RSD = relative standard deviation = standard deviation/mean (expressed as %).

#### Materials Available:

8 gallons water (obtain from eye wash station) 0.8 kg soil (divide in two) 6 mL of Ferric Chloride

## **Equipment Available:**

Two 5-gal buckets
Turbidity meter
Drill & mixer paddle
Column testing apparatus
Four 250 mL Erlenmeyer flasks

## Experiment 10 Sedimentation Lab

#### LAB METHOD & MINIMUM REQUIREMENTS:

- 1. Plan out whole lab and set up a table of measurements and observations for before and after all planned actions. Label the four Erlenmeyer flasks with the depth of measurement (dedicate one flask to each measurement level).
- 2. TURBIDITY: Calibrate turbidity meter using provided standards, and note meter manufacture and model. Mix sample. Place ~20 mL sample water into turbidity vial and cap. Place vial into turbidity meter, cap with shroud and press read. Record value. Dump out sample and rinse vial with DI water.
- 3. Take two buckets of water and the soil onto the testing platform. Homogenize the two buckets of sample water (4 gal) and 0.4 kg of soil using the drill and mixer paddle. Add 3 mL of Ferric Chloride to each bucket and mix well for one minute alternating between buckets. Lift the buckets one by one and pour into the column set for your test. Begin the time measurement.
- 4. Take turbidity samples (about 100 mL is plenty) from each sampling port and test for turbidity in triplicate. Sample at 0, 10, 20, 30, 40, 50 and 60 minutes.
- 5. After the 60 minute sampling, place a bucket over the clean out to redirect the water flow and raise the pool to its' full water holding level. Blocking behind the bucket firmly, CAREFULLY remove the lower cleanout cap and drain the column. Assist the instructor in pumping off the water and scraping up the sludge (sediment) from the pool until the pool is emptied.
- 6. Prepare a report of all methods and observations. Make use of tables as needed, particularly for the evaluation of the triplicate turbidity results. Include photographs. Prepare a graph of turbidity level by depth and time (use average of turbidity but do not include standard deviation information on graph). Interpret results and make conclusions. Assess potential for error.

## Appendix G – Flocculant Aided Filtration Analysis

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	Filter Rate (vol/time)	1.07	1.10	1.16	0.59	1.01	2.39	1.00	1.02	1.06	0.59	1.05	1.04
Garnet	Filtrate Tubidity (UTN)	4.70	3.62	9.55	18.13	32.7	11.11	3.92	37.6	22.6	38.3	41.3	46.2
GAC + 54	Filtrate Tubidity (UTN)	5.64	6.16	155	26.3	78.9	126	6.71	94.4	112	72.7	124	202
#4	Filtrate Tubidity (UTN)	23.7	25.7	229	78.3	73.8	181	22.5	102	179	96.1	211	254
Filter #6:	Turbidity in Beaker Tank (NTU)	6.36	8.44	268	27.0	87.9	237	7.56	124	319	95.8	204	332
	Filter Rate (vol/time)	1.04	1.10	1.29	1.02	1.05	1.06	1.04	1.05	1.04	1.03	1.00	2.11
20 Sand	Filtrate Tubidity (UTN)	2.93	4.40	13.76	9.91	32.5	12.98	4.05	64.8	20.1	74.1	39.40	54.0
+ 10-	Filtrate Tubidity (UTU)	5.42	19.94	141	25.0	75.6	149	10.72	157	208	110	108	276
: #4 GAC	Filtrate Tubidity (UTU)	26.0	85.4	217	65.2	91.0	382	39.7	162	478	122	139	318
Filter #5:	Turbidity in Beaker Tank (NTU)	7.90	5.75	350	9.77	83.3	341	8.03	174	312	187	159	349
	Filter Rate (vol/time)	1.06	1.06	1.14	1.01	1.06	N/A	1.02	1.03	1.15	0.59	1.16	0.58
	Filtrate Tubidity (UTU)	3.27	3.17	18.15	13.57	42.3	15.82	3.95	47.1	16.55	36.5	45.6	50.3
net	Filtrate Tubidity (UTN)	5.39	5.13	159	19.3	9.62	135	10.12	89.3	318	67.1	104	256
54 Gar	Filtrate Tubidity (UTU)	22.4	29.8	301	74.9	92.9	303	16.85	8.66	338	70.0	121	315
Filter #4:	Turbidity in Beaker Tank (NTU)	8.77	2.99	299	18.44	94.8	313	9.79	108	320	86.7	141	286
	Filter Rate (vol/time)	1.06	1.05	1.12	1.03	1.01	N/A	1.00	1.01	1.01	1.01	1.07	1.17
	Filtrate Tubidity (UTU)	2.74	4.00	10.6	8.72	27.9	10.91	3.53	40.4	22.0	56.8	33.50	48.4
Sand	Filtrate Tubidity (UTN)	7,45	12.72	128	14.97	42.9	78.6	6.61	68.7	142	67.7	83.7	246
10-20	Filtrate Tubidity (UTU)	34.6	42.2	215	61.9	105	205	21.5	106	307	94.0	215	289
Filter #3:	Turbidity in Beaker Tank (NTU)	4.82	5.05	197	10.8	55.0	210	6.63	88.3	296	98.0	114	241
	Filter Rate (vol/time)	1.06	1.05	1.19	1.02	1.56	1.32	1.00	1.06	N/A	0.59	1.03	1.05
	Filtrate Tubidity (UTU)	3.17	3.03	11.8	16.45	23.6	8.24	5.28	33.2	30.5	37.8	36.1	28.3
Sand	Filtrate Tubidity (UTN)	7.26	10.97	136	10.98	24.5	133	8.68	82.5	139	71.9	162	189
: 4-10 S	Filtrate Tubidity (UTN)	34.3	23.5	255	17.96	66.3	275	24.1	98.6	188	76.5	169	189
Filter #2	Turbidity in Beaker Tank (NTU)	5.09	3.19	262	8.38	25.8	225	8.58	84.8	256	85.6	126	199
	Filter Rate (vol/time)	1.05	1.04	1.42	1.06	1.04	1.32	1.02	1.03	1.17	1.02	1.03	1.40
	Filtrate Tubidity (UTU)	10.39	6.56	19.04	10.43	30.0	17.14	3.88	48.9	23.8	53.0	4.4	58.4
0	Filtrate Tubidity (UTN)	14.15	7.12	206	19.02	74.1	206	12.86	117	195	104	141	198
#1: #4 GAC	Filtrate Tubidity (UTN)	34.7	261	136	69.7	104	193	19.09	134	398	142	211	500
Filter #1	Turbidity in Beaker Tank (NTU)	16.00	7.47	305	13.74	103	372	7.00	166	350	143	188	422
	Settle	20 min	٤	9	3 F	9	00	20 ri	e e	00	20 min	9	٤
	xiM wol2	20 n min	20 min	01	n min	n min	n no	20 n min	n min	n no	n min	n min	2
	Coagulant Rapid Mix	r li min	ا ا ا ا	다 를 기	r n min	ا ا m 1	ار سin	ر الر الا	l li	m L min	4 E	± E	M 1
	Final Dose of	@ 10 mL /L /2 L	3 @ 10 mL 1/L /2 L	@ 10 mL /L /2 L	@ 5 mL /L /2 L	@ 5 mL /L /2 L	@ 5 mL /L /2 L	@ 2.5 mL /L /2 L	@ 2.5	2.5	@ /L NONE	@ /L NONE	@ /L NONE
	Coagulant & Strength	FeCI3 20mL	FeCI 20m	FeCI3 20mL	FeCI3 @ 20mL/L	FeCI3 @ 20mL/L	FeCI3 20mL	FeCI3 20mL	FeCI3 20mL	FeCI3 20mL	FeCI3 @ 20mL/L	FeCI3 20mL	FeCI3 @ 20mL/L
ļ	Sediment Mixing	.o 2 I min	.0 m n	0. I min 2	min	.o 2 I min	2 min	.o 2 I min	min	o 2 I min	.o 2 I min	0. T min	2 m 2
	Sediment	Mankato Topsoil	Mankato Topsoil	Mankato Topsoil	Mankato Topsoil	Mankato Topsoil	Mankato Topsoil	Mankato Topsoil	Mankato Topsoil	Mankato Topsoil	Mankato Topsoil	Mankato Topsoil	Mankato Topsoil
	Sediment Dose	50 g /2L	50 g /2L	50 g /2L	50 g /2L	50 g /2L	50 g /2L	50 g /2L	50 g /2L	50 g /2L	50 g /2L	50 g /2L	50 g /2L
	Вun	-	2	т	4	2	9	7	ø	6	10	11	12

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Garnet	Filter Rate (vol/time)	1.07	1.10	1.16	0.59	1.01	2.39	1.00	1.02	1.06	0.59	1.05	1.04
+ 54	Average Filtrate Tubidity (UTU)	10.4	11.7	166.1	35.4	54.3	139.7	10.4	75.7	166.3	64.3	138.7	196.0
Filter #6: #4 GAC -	Turbidity in Beaker Tank (NTU)	98'9	8.44	268	27.0	87.9	237	7.56	124	319	95.8	204	332
Sand	Filter Rate (vol/time)	1.04	1.10	1.29	1.02	1.05	1.06	1.04	1.05	1.04	1.03	1.00	2.11
10-20	Average Filtrate Tubidity (UTN)	11.7	30.7	189.4	25.3	58.5	361.5	16.3	112.3	263.7	103.2	99.7	222.5
Filter #5: #4 GAC +	Turbidity in Beaker Tank NTU)	7.90	5.75	350	9.77	83.3	341	8.03		312	187	159	349
	Filter Rate (vol/time)	1.06	1.06	1.14	1.01	1.06	N A	1.02	1.03	1.15	0.59	1.16	0.58
54 Garnet	Average Filtrate Tubidity (UTU)	10.7	11.3	200.4	31.5	62.9	308.0	9.2	9.69	219.7	52.6	87.7	200.7
Filter #4:	Turbidity in Beaker Tank NTU)	8.77	2.99	299	18.44	94.8	313	9.79	108	320	86.7	141	286
Sand	Filter Rate (vol/time)	1.06	1.05	1.12	1.03	1.01	۷ ۷	1.00	1.01	1.01	1.01	1.07	1.17
10-20	Average Filtrate Tubidity (UTN)	13.5	16.1	137.7	24.6	53.9	138.8	9.7	65.1	301.5	64.2	110.0	177.0
Filter #3:	Turbidity in Beaker Tank NTU)	4.82	5.05	197	10.8	55.0	210	6.63	88.3	296	98.0	114	241
Sand	Filter Rate (vol/time)	1.06	1.05	1.19	1.02	1.56	1.32	1.00	1.06	N/A	0.59	1.03	1.05
4-10	Average Filtrate Tubidity (UTN)	13.5	9.2	172.8	9.1	31.0	167.1	11.2	61.5	148.4	54.4	98.7	129.8
Filter #2:	Turbidity in Beaker Tank NTU)	5.09	3.19	262	8.38	25.8	225	8.58		256	85.6	126	199
	Filter Rate (vol/time)	1.05	1.04	1.42	1.06	1.04	1.32	1.02	1.03	1.17	1.02	1.03	1.40
#4 GAC	Average Filtrate Tubidity (UTN)	19.7	91.6	120.3	33.1	69.4	138.7	11.9		205.6	7.66	132.1	155.1
Filter #1:	Turbidity in Beaker Tank NTU)	16.00	7.47	305	13.74	103	372	7.00		350	143	188	422
	Settle	20 min	0 0	no	20 min	0 L	01	20 min	0 0	no	20 min	0 0	no
	xiM wol2	20 min	20 min	0U	20 min	20 min	00	20 min		no	20 min	20 min	OL
	Kapid Mix	1 min	п п	1 min	1 min	nin	nin	n H		1 min	1 min	n m	min 1
	Final Dose of Coagulant	10 mL /2 L	10 mL /2 L	10 mL /2 L	5 mL /2 L	5 mL /2 L	5 mL /2 L	2.5 mL /2 L	2.5 mL /2 L	2.5 mL /2 L		NONE	NONE
	Coagulant & Strength	FeCI3 @ 20mL/L	FeCl3 @ 10 mL 20mL/L /2 L	FeCI3 @ 20mL/L	FeCI3 @ 20mL/L	FeCI3 @ 20mL/L	FeCI3 @ 20mL/L	FeCI3 @ 2	FeCI3 @ 2.5 mL 20mL/L /2 L	FeCI3 @ : 20mL/L	FeCI3 @ 20mL/L	FeCI3 @ 20mL/L	FeCI3 @ 20mL/L
	Sediment Mixing	2 I	2 min	2 F	2 I	2 nim	2 min	2 I	2 nim	2 F	2 min	2 min	2 I
	JnəmibəS	ato oil	Mankato Topsoil	Mankato Topsoil	Mankato Topsoil	Mankato Topsoil	Mankato Topsoil			Mankato Topsoil		Mankato Topsoil	Mankato Topsoil
	Sed Inemibe2	50 g I	50 g I	50 g //2L		50 g   1				50 g //2L		50 g I	50 g   1
	uny	1	2	3	4	ι	9	7		6	10	11	12

Analyst: Ryan Johnson Date: May 15, 2013

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			Average Tu	Average Turbidity of Filtrate (NTUs)	te (NTUs)	Reduction I	Reduction Due to Flocculant Addition	nt Addition
Filter	Filtration Technique	Turbidity of Blank (NTUS)	1,25 ml /l	2.5 ml /l	7 m 7	1,25 ml /l	2.5 ml /l	5 ml /l
1	conv	99.7	11.9	33.1	19.7	88.1%	66.8%	80.2%
1	direct	132.1	100	69.4	91.6	24.3%	47.5%	30.7%
Н	inline	155.1	205.6	138.7	120.3	-32.6%	10.6%	22.4%
7	conv	54.4	11.2	9.1	13.5	79.4%	83.3%	75.2%
7	direct	98.7	61.5	31	9.5	37.7%	%9'89	90.7%
7	inline	129.8	148.4	167.1	172.8	-14.3%	-28.7%	-33.1%
m	CONV	64.2	9.7	24.6	13.5	84.9%	61.7%	79.0%
n	direct	110	65.1	53.9	16.1	40.8%	51.0%	85.4%
m	inline	177	301.5	138.8	137.7	-70.3%	21.6%	22.2%
4	CONV	52.6	9.5	31.5	10.7	82.5%	40.1%	79.7%
4	direct	87.7	9.69	62.9	11.3	20.6%	28.3%	87.1%
4	inline	200.7	219.7	308	200.4	-9.5%	-53.5%	0.1%
Ŋ	CONV	103.2	16.3	25.3	11.7	84.2%	75.5%	88.7%
2	direct	99.7	112.3	58.5	30.7	-12.6%	41.3%	69.2%
2	inline	222.5	263.7	361.5	189.4	-18.5%	-62.5%	14.9%
9	CONV	64.3	10.4	35.4	10.4	83.8%	44.9%	83.8%
9	direct	138.7	75.7	54.3	11.7	45.4%	%6.09	91.6%
9	inline	196	166.3	139.7	166.1	15.2%	28.7%	15.3%

	Filter Rate (vol/time)	1.07	1.10	1.16	0.59	1.01	2.39	1.00	1.02	1.06	0.59	1.05	1.04
	Paper Filter Filtrate Tubidity (UTN)	4.70	3.62	9.55	18.13	32.7	11.11	3.92	37.6	22.6	38.3	41.3	46.2
	Woven Filter Filtrate Tubidity (UTU)	5.64	6.16	155	26.3	78.9	126	6.71	94.4	112	72.7	124	202
	Nonwoven Filter Filtrate Tubidity (UTN)	23.7	25.7	229	78.3	73.8	181	22.5	102	179	96.1	211	254
Beaker 6	Turbidity in Beaker Tank (NTU)	6.36	8.44	268	27.0	87.9	237	7.56	124	319	95.8	204	332
	Filter Rate (vol/time)	1.04	1.10	1.29	1.02	1.05	1.06	1.04	1.05	1.04	1.03	1.00	2.11
	Paper Filter Filtrate Tubidity (UTN)	2.93	4.40	13.76	9.91	32.5	12.98	4.05	8.8	20.1	74.1	39.40	0.42
	Woven Filter Filtrate Tubidity (UTN)	5.42	19.94	141	25.0	75.6	149	10.72	157	208	110	108	276
	Nonwoven Filter Filtrate Tubidity (NTU)	26.0	85.4	217	65.2	91.0	182	39.7	162	478	122	139	318
Beaker 5	Turbidity in Beaker Tank (NTU)	7.90	5.75	350	9.77	83.3	341	8.03	174	312	187	159	349
	Filter Rate (vol/time)	1.06	1.06	1.14	1.01	1.06	N/A	1.02	1.03	1.15	0.59	1.16	0.58
	Paper Filter Filtrate Tubidity (UTU)	3.27	3.17	18.15	13.57		15.82	3.95	47.1	16.55	36.5	45.6	50.3
	Woven Filter Filtrate Tubidity (UTN)	5.39	5.13	159	19.3	79.6	135	10.12	89.3	318	67.1	104	256
	Nonwoven Filter Filtrate Tubidity (NTU)	22.4	29.8	301	74.9	92.9	303	16.85	8.66	338	70.0	121	315
Beaker 4	Turbidity in Beaker Tank (NTU)	8.77	2.99	299	18.44	94.8	313	9.79	108	320	86.7	141	286
	Filter Rate (vol/time)	1.06	1.05	1.12	1.03	1.01	N/A	1.00	1.01	1.01	1.01	1.07	1.17
	Paper Filter Filtrate Tubidity (UTU)	2.74	4.00	10.6	8.72	27.9	10.91	3.53	40.4	22.0	56.8	33.50	48.4
	Woven Filter Filtrate Tubidity (UTU)	7.45	12.72	128	14.97	42.9	78.6	6.61	68.7	142	67.7	83.7	246
_	Nonwoven Filter Filtrate Tubidity (NTU)	34.6	42.2	215	61.9	105	205	21.5	106	307	94.0	215	280
Beaker 3	Turbidity in Beaker Tank (NTU)	4.82	5.05	197	10.8	55.0	210	6.63	88.3	296	98.0	114	241
	Filter Rate (vol/time)	1.06	1.05	1.19	1.02	1.56	1.32	1.00	1.06	N/A	0.59	1.03	1.05
	Paper Filter Filtrate Tubidity (UTU)	3.17	3.03	11.8	16.45	23.6	8.24	5.28	33.2	30.5	37.8	36.1	28.3
	Woven Filter Filtrate Tubidity (NTU)	7.26	10.97	136	10.98	24.5	133	89'8	82.5	139	71.9	162	189
2	Nonwoven Filter Filtrate Tubidity (NTU)	34.3	235	255	17.96	66.3	275	24.1	98.6	188	76.5	169	189
Beaker	Turbidity in Beaker Tank (NTU)	5.09	3.19	262	8.38	25.8	225	8.58	84.8	256	85.6	126	199
	Filter Rate (vol/time)	1.05	1.04	1.42	1.06	_	1.32	1.02	1.03	1.17	1.02	1.03	1.40
	Paper Filter Filtrate Tubidity (UTU)	10.39	6.56	19.04	10.43	30.0	17.14	3.88	48.9	23.8	53.0	44.4	58.4
	Woven Filter Filtrate Tubidity (NTU)	14.15	7.12	206	19.02	74	206	12.86	117	195	104	141	198
1	Nonwoven Filter Filtrate Tubidity (UTU)	34.7	261	136	69.7	104	193	19.09	134	398	142	211	209
Beaker	Turbidity in Beaker Tank (NTU)	16.00	7.47	305	13.74		372	7.00	166	350	143	188	422
	əjijəS	20 20 min min	20 min no	0 10	20 20 min min		0 10	20 20 min min	20 min no		20 20 min min	0.5	
	XiM biqsA XiM wol2		_	1 min no			1 min no	1 2 min m			_	2 E	1 in o
	Final Dose of Coagulant Manid Mix	10 mL 1 /2 L mir	10 mL /2 L min	10 mL 1	5 mL 1/2 L mir	5 mL 1 /2 L min	5 mL 1/2 L mir	_	i @ 2.5 mL 1 -/L /2 L min		NONE min	NONE min	NONE mi
	Coagulant & Strength	FeCI3 @ 10 20mL/L /	FeCI3 @ 10 20mL/L /	FeCI3 @ 10 20mL/L /	FeCI3 @ 5 20mL/L /	FeCI3 @ 5 20mL/L /	FeCI3 @ 5 20mL/L	FeCI3 @ 2. 20mL/L /	FeCI3 @ 2. 20mL/L	FeCI3 @ 2.	FeCI3 @ 20mL/L N		FeCI3 @ 20mL/L N
	Sediment Mixing	2 min	2 min	2 min	2 min	2 min	2 min	2 min	2 nin	2 min	2 min	2 nin	2 E
	Sediment	Mankato Topsoil	Mankato Topsoil	Mankato Topsoil	Mankato Topsoil	Mankato Topsoil	Mankato Topsoil	Mankato Topsoil	Mankato Topsoil	Mankato Topsoil	Mankato Topsoil	Mankato Topsoil	Mankato Topsoil
	Sediment Dose	50 g N	50 g N	50 g N	50 g h		50 g N	50 g N	50 g N		50 g N	_	50 g //2L
	Run		2	6	4	'n	9	7	œ	6	10	=======================================	12

Analyst: FA HM Date: May 28, 2013

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Run	Sediment Dose	Sediment	Sediment Mixing	Coagulant & Strength	Final Dose of Coagulant	Rapid Mix	Slow Mix	Settle	Average Turbidity in Beaker Tank (NTU)	Average Nonwoven Filter Filtrate Tubidity (NTU)	Average Woven Filter Filtrate Tubidity (NTU)	Average Paper Filter Filtrate Tubidity (NTU)	Average Nonwoven Filter Rate (vol/time)
1	50 g /2L	Mankato Topsoil	2 min	FeCl3 @ 20mL/L	10 mL /2 L	1 min	20 min	20 min	8.2	29.3	7.6	4.5	1.1
2	50 g /2L	Mankato Topsoil	2 min	FeCl3 @ 20mL/L	10 mL /2 L	1 min	20 min	no	5.5	113.2	10.3	4.1	1.1
3	50 g /2L	Mankato Topsoil	2 min	FeCl3 @ 20mL/L	10 mL /2 L	1 min	no	no	280.2	225.5	154.2	13.8	1.2
4	50 g /2L	Mankato Topsoil	2 min	FeCl3 @ 20mL/L	5 mL /2 L	1 min	20 min	20 min	14.7	61.3	19.3	12.9	1.0
5	50 g /2L	Mankato Topsoil	2 min	FeCl3 @ 20mL/L	5 mL /2 L	1 min	20 min	no	75.0	88.8	62.6	31.5	1.1
6	50 g /2L	Mankato Topsoil	2 min	FeCl3 @ 20mL/L	5 mL /2 L	1 min	no	no	283.0	223.2	137.9	12.7	#VALUE!
7	50 g /2L	Mankato Topsoil	2 min	FeCl3 @ 20mL/L	2.5 mL /2 L	1 min	20 min	20 min	7.9	24.0	9.3	4.1	1.0
8	50 g /2L	Mankato Topsoil	2 min	FeCl3 @ 20mL/L	2.5 mL /2 L	1 min	20 min	no	124.2	117.1	101.5	45.3	1.0
9	50 g /2L	Mankato Topsoil	2 min	FeCl3 @ 20mL/L	2.5 mL /2 L	1 min	no	no	308.8	314.7	185.7	22.6	#VALUE!
10	50 g /2L	Mankato Topsoil	2 min	FeCl3 @ 20mL/L	NONE	1 min	20 min	20 min	116.0	100.1	82.2	49.4	0.8
11	50 g /2L	Mankato Topsoil	2 min	FeCI3 @ 20mL/L	NONE	1 min	20 min	no	155.3	177.7	120.5	40.1	1.1
12	50 g /2L	Mankato Topsoil	2 min	FeCl3 @ 20mL/L	NONE	1 min	no	no	304.8	260.8	227.8	47.6	1.2

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Flocculant Filtration Analysis

Analyst: FA HM Date: May 28, 2013

Nonwoven Membrane Filtration

	Turbidity of Blank as Filtered	Average Tu	Average Turbidity of Filtrate (NTUs)	ite (NTUs)	Reduction D	Reduction Due to Flocculant Addition	nt Addition
	(NTUS)	1.25 mL/L	2.5 mL/L	5 mL/L	1.25 mL/L	2.5 mL/L	2 mL/L
Conv	100	24.0	61.3	29.3	%1.9/	38.7%	70.7%
Direct	178	117.1	88.8	113.2	34.1%	20.0%	36.3%
Indirect	261	314.7	223.2	225.5	-20.6%	14.4%	13.5%

# Woven Membrane Filtration

	Turbidity of Blank as Filtered	Average Tu	Average Turbidity of Filtrate (NTUs)	ite (NTUs)	Reduction D	Reduction Due to Flocculant Addition	nt Addition
	(NTUS)	1.25 mL/L	2.5 mL/L	5 mL/L	1.25 mL/L	2.5 mL/L	5 mL/L
Conv	82	9.3	19.3	7.6	%2'88	%9'92	%8'06
Direct	121	101.5	62.6	10.3	15.8%	48.0%	91.4%
Indirect	228	185.7	137.9	154.2	18.5%	39.4%	32.3%

## Paper Membrane Filtration

	Turbidity of Blank as Filtered	Average Tu	Average Turbidity of Filtrate (NTUs)	te (NTUs)	Reduction [	Reduction Due to Flocculant Addition	ıt Addition
	(NTUS)	1.25 mL/L	2.5 mL/L	5 mL/L	1.25 mL/L	2.5 mL/L	5 mL/L
Conv	49	4.1	12.9	4.5	%2'16	74.0%	%8'06
Direct	Direct 40	45.3	31.5	4.1	-13.1%	21.4%	89.7%
Indirect	48	22.6	12.7	13.8	52.5%	73.3%	71.0%